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Lorenz Ratios of Technically Important Metals and Alloys

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Lorenz Ratios of Technically Important Metals and Alloys

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Lorenz Ratios of Technically Important
Metals and Alloys

J. G. Hust and L. L. Sparks

A comprehensive review and compilation of the world literature on Lorenz ratio of technically important metals and alloys is presented. Lorenz ratio, electrical resistivity, thermal conductivity, and characterization data are compiled in tabular form and the Lorenz ratio data are presented in graphical form as well. Data are included here only if the research reported both thermal conductivity and electrical resistivity of the specimens. No attempt has been made to smooth data or present recommended values.

Key Words: Alloys; compilation; cryogenic; electrical resistivity; Lorenz Ratio; metals; thermal conductivity.

1. Introduction

The development of new materials and a renewed interest in existing materials by modern technological industry is creating a demand for thermal and electrical property data. These data are needed for the selection of suitable construction materials and the prediction of operating characteristics of low temperature systems. Apparatus are available at this laboratory as well as others for the accurate measurement of thermal conductivity, electrical resistivity, and thermopower of metals and alloys over the range from 4 to 300 K. However, the measurement of thermal conductivity is slow and expensive; one cannot hope to fill the immediate demand for data by measurement alone. Furthermore, metals and alloys display a wide range in their conductivity values at a given temperature and a single material may exhibit values varying over several orders of magnitude as a function of temperature. One cannot look to theory for these values, since, presently, theoretical predictions represent qualitative guidelines only. However,

a strong relationship between thermal conductivity and electrical resistivity of metals was observed many years ago and the nature of this correlation has been studied ever since. This relationship, called the Wiedemann-Franz-Lorenz law, makes it possible to predict approximate values of thermal conductivity from less expensive electrical resistivity measurements. For this reason we feel a compilation of Lorenz ratios of technically important metals is overdue. The predictive value of these data is explained below.

2. Theory

In order to better understand the methods of predicting thermal conductivity from the Lorenz ratio, it will be helpful to have an understanding of the general thermal and electrical properties of metals and alloys. The following sections on thermal conductivity and electrical resistivity are included as a review of the fundamental physical behavior.

2.1 Thermal Conductivity

The thermal conductivity, λ , of a metal or alloy usually is considered to be the sum of two components- the electronic, λ_e , and lattice, λ_g :

$$\lambda = \lambda_e + \lambda_g. \quad (1)$$

There are other mechanisms of heat transport; however, they generally are not applicable here. The lattice term designates the energy carried by the lattice vibrations, called phonons. The subscript g comes from the German word for lattice, Gitter.

In "electrically pure" materials the lattice term is small (usually less than 5 and almost never more than 20 percent of the total) compared to the electron term. Electrical purity is characterized by the residual

resistivity ratio, RRR, between 273 K and 4 K ($\text{RRR} = \rho_{273}/\rho_{4 \text{ K}}$). The electrical purity of a material is not specified uniquely by the chemical purity of that material; it also depends on the distribution and location of the impurity atoms and on the physical imperfections in the solid. For example, whereas the chemical purity may remain essentially unchanged in annealing, the electrical purity of a material can, in some instances, be changed by an order of magnitude or more. As will be described later, such a change also affects the thermal conductivity.

In alloys, the electronic conductivity is so small, especially at lower temperatures, that the lattice conductivity no longer is negligible. As a matter of fact, λ_g is often much larger than λ_e . The total conductivity of alloys is less than that of pure metals, and some alloys have conductivities approaching those of thermal insulators. The general temperature dependence and relative magnitude of the conductivities of several materials are illustrated in figure 1. More details are given by Powell [1].¹

2.2 Electrical Resistivity

For commercially pure metals and some alloys, the electrical resistivity is described adequately by Matthiessen's rule. This rule, eq(2), states that the electrical resistivity of a metal is the sum of two parts: the intrinsic resistivity, ρ_i , which is characterized by electrons interacting with phonons only, and the residual resistivity, ρ_o , which is characterized by electrons interacting with the chemical and physical imperfections of the metal.

$$\rho(T) = \rho_o + \rho_i(T) \quad (2)$$

The residual resistivity is assumed to be temperature independent while

¹ Numbers in brackets refer to list of text references at the end of this paper.

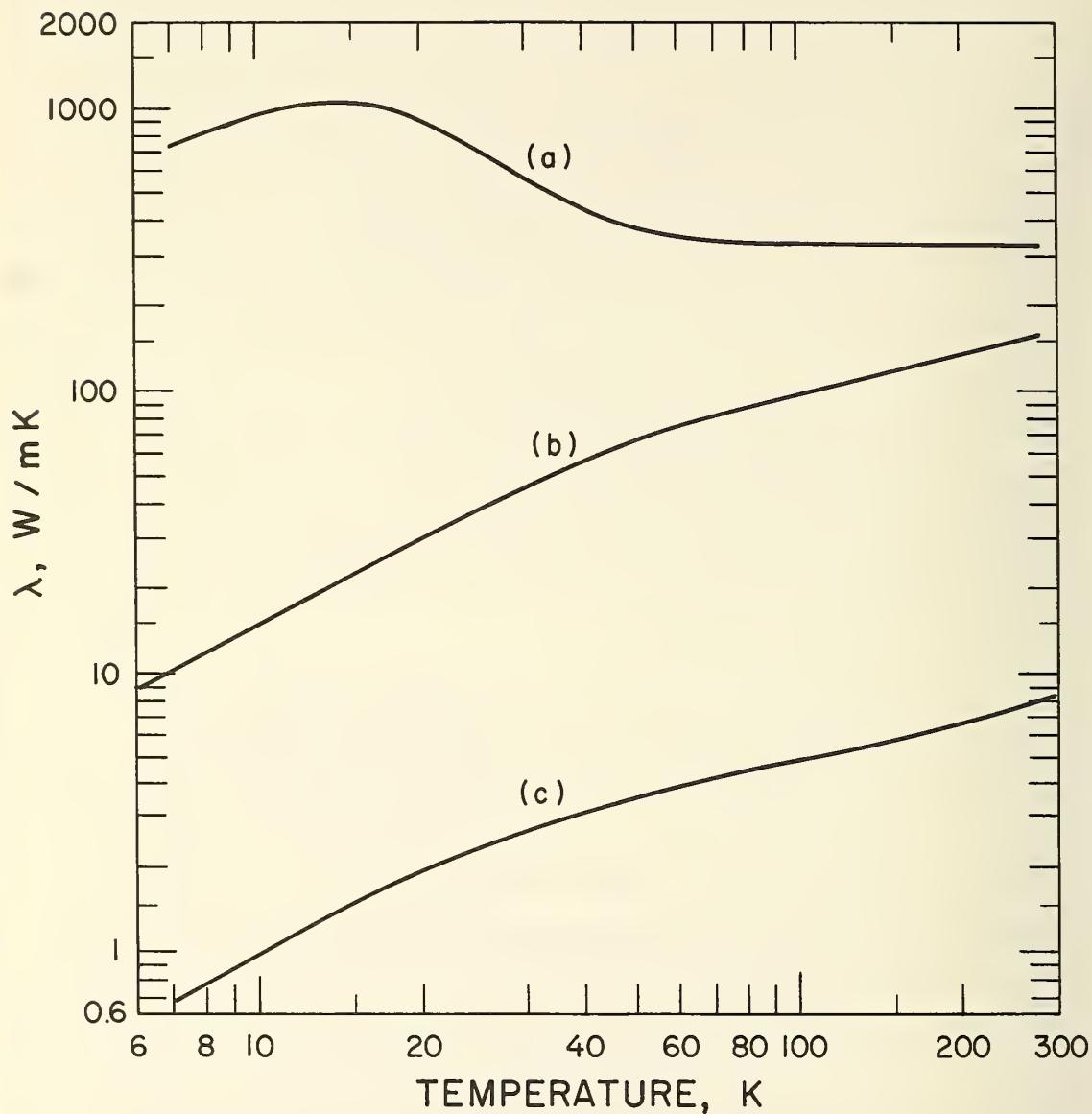


Figure 1. Thermal conductivity of metals and alloys. (a) pure metal (gold), (b) high conductivity alloy (aluminum-7039), and (c) low conductivity alloy (titanium-5A 1/2-2.5 Sn).

the intrinsic resistivity increases rapidly with temperature. The specific temperature dependence varies strongly but systematically both with temperature and material. Nevertheless, ρ_i is not dependent significantly upon small composition changes for a given metal or alloy. Thus, if one knows $\rho_i(T)$ for a given metal, $\rho(T)$ for different compositions can be obtained from eq(2) after measuring only ρ_o . The value of ρ_o is obtained by measuring ρ at low temperatures where ρ_i is negligible (liquid helium temperature is adequate). A few characteristic electrical resistivity curves of pure metals and alloys are given in figure 2.

2.3 Lorenz Ratio

In 1853 Wiedemann and Franz formulated an empirical law relating the thermal and electrical conductivities of metals, namely, that the ratio of the electrical and thermal conductivities (WF ratio) at a given temperature is the same for all metals. In 1872 Lorenz discovered that the WF ratio is proportional to temperature. The result was the Wiedemann-Franz-Lorenz (WFL) law:

$$\frac{\lambda}{\sigma} = \lambda \rho = LT, \quad (3)$$

where σ = electrical conductivity, L = Lorenz number, and T = absolute temperature.

Drude gave a theoretical derivation of the WFL law for the electronic component of thermal conductivity in 1900 and obtained a value of $2.228 \times 10^{-8} \text{ V}^2/\text{K}^2$ for L . Sommerfeld calculated the first order approximation of L from the more recent free electron theory of metals. His value, $2.443 \times 10^{-8} \text{ V}^2/\text{K}^2$, is commonly designated L_o . Electron Lorenz numbers, $\rho \lambda_e / T$, other than the Sommerfeld value, will be designated L_e to distinguish them from total Lorenz numbers, $L = \rho \lambda / T$.

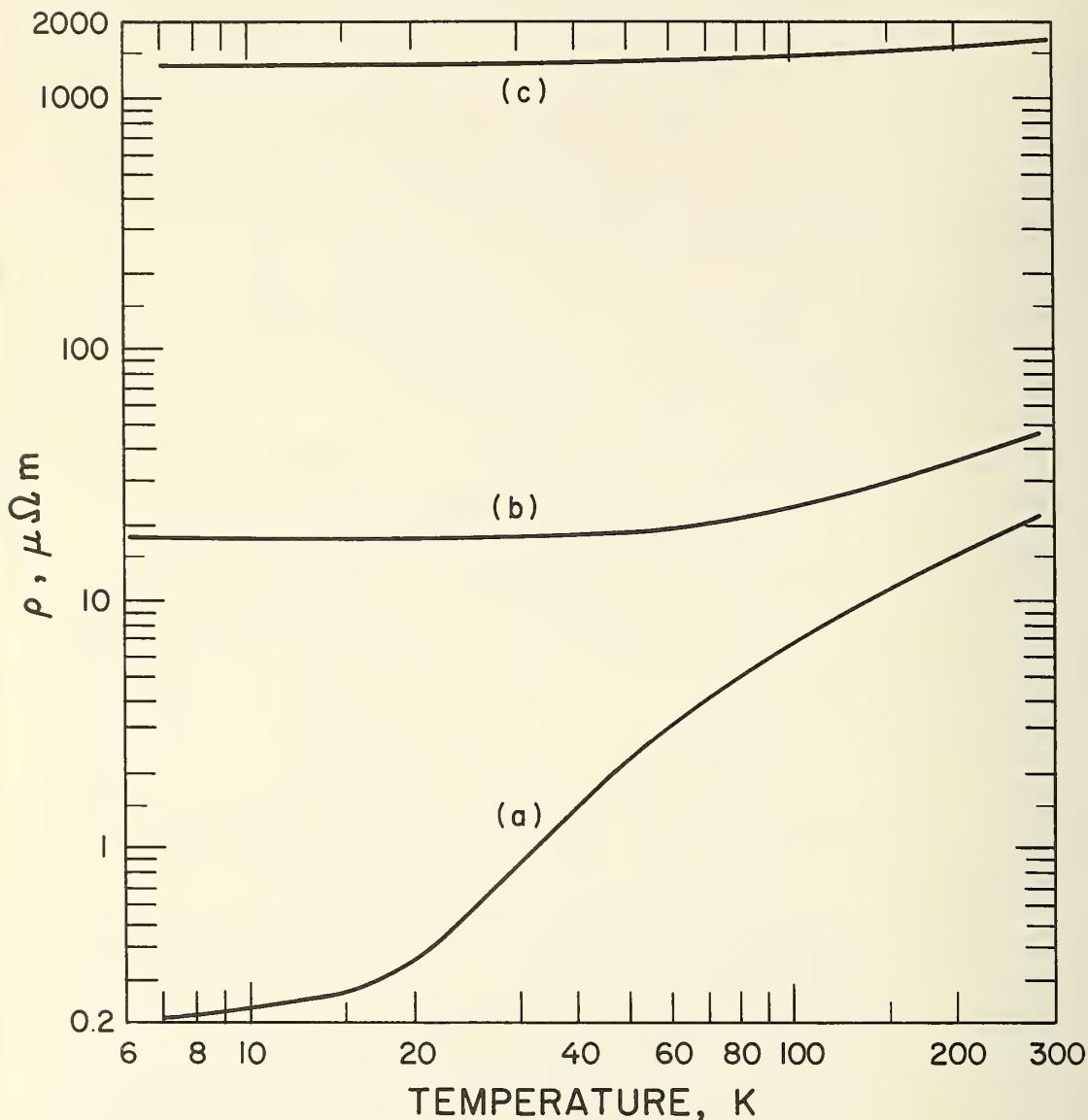


Figure 2. Electrical resistivity of metals and alloys. (a) pure metal (gold), (b) high conductivity alloy (aluminum-7039), and (c) low conductivity alloy (titanium-5A & 2.5 Sn).

For many pure metals the experimentally determined values of L fall between 2.2 and $3.0 \times 10^{-8} V^2/K^2$ at room temperature and only slightly higher (2 to 3 percent) at $100^\circ C$. At very low temperatures the experimental values of L are also near the Sommerfeld value. At intermediate temperatures, values of L below L_o are observed. The magnitude of the decrease of L below L_o increases with increasing purity. The values of L and L_e are nearly identical for pure metals since the lattice conductivity is small.

The decrease of L_e from L_o in pure metals is caused primarily by the fact that the relaxation times for thermal and electrical conduction processes are different at intermediate temperatures. The electrical conductivity increases with decreasing temperature faster than the thermal conductivity. As the residual resistivity decreases, L_e approaches the value for a defect free crystal, thus, resulting in a curve with a lower minimum. The electronic Lorenz ratios of several metals as well as a defect free crystal are shown in figure 3.

For alloys the measured values of Lorenz ratio are always higher than their pure metal counterparts, in some instances considerably higher. However, a similarity still exists in that the L vs T curve tends toward L_o at both extremes of temperature. At intermediate temperatures values of L vary from slightly below L_o for dilute alloys to more than ten times larger than L_o for some highly alloyed structural metals. A few typical curves for alloys are shown in figures 4 and 5. The large values of Lorenz ratio observed for these alloys is caused by the presence of a significant lattice conductivity as mentioned previously. Often, knowledge of the magnitude of the lattice conductivity is so limited that it is ignored in the calculation of L_e for alloys as it is for pure metals. This is incorrect for alloys, of course, but until λ_g is known more accurately, our knowledge of L_e will also be severely limited.

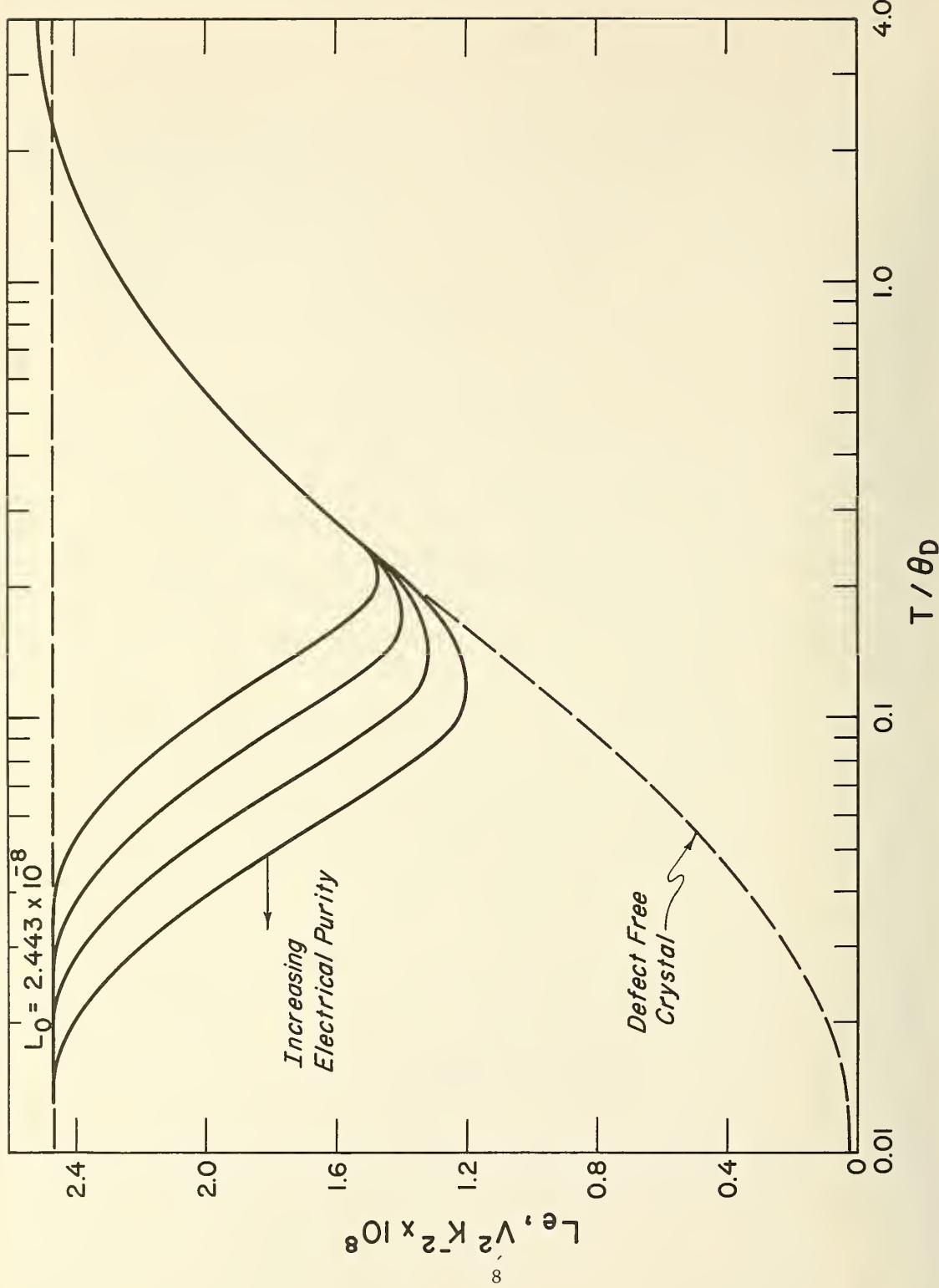


Figure 3. Electronic Lorenz ratio for pure metals and defect free metals.

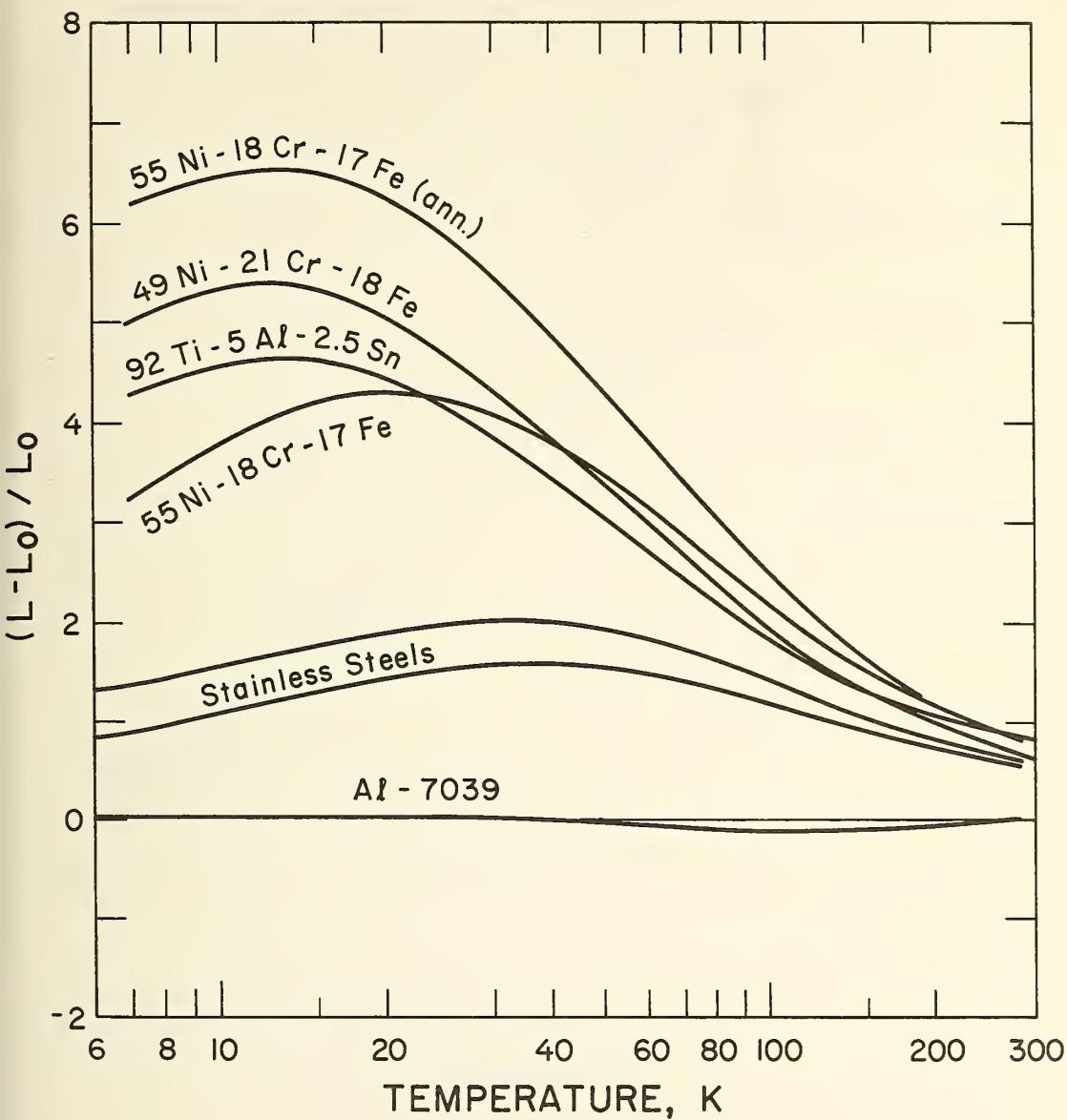


Figure 4. Values of $(L - L_0)/L_0$ for several classes of commonly used materials.

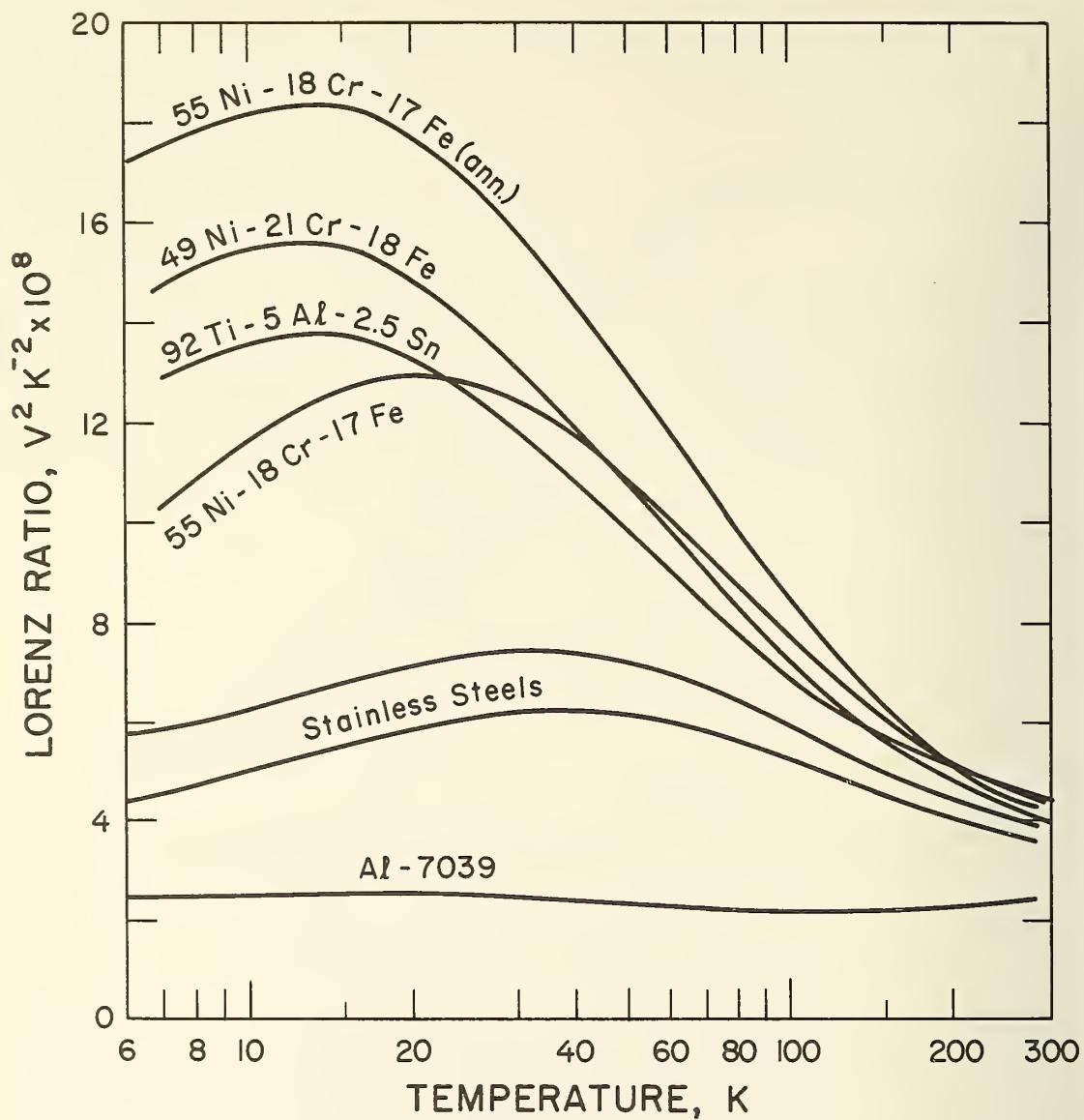


Figure 5. Values of the total Lorenz number for several classes of commonly used materials.

White [2] recently has compiled values of L_e as a function of reduced temperature, T/θ_D , for several metals (θ_D is the Debye temperature). Values of θ_D for selected metals are shown in table 1 [3]. White's L_e data show a variation of only about ± 2 percent from the mean values to temperatures as low as $0.2\theta_D$. White calculated λ_g from eq (4) below, whose form was proposed by Leibfried and Schlömann [4] among others:

$$\lambda_g = 180 \alpha A \theta_D^3 / \gamma^3 T \text{ W m}^{-1} \text{ K}^{-1}. \quad (4)$$

In eq (4) α = lattice spacing, A = atomic weight, θ_D = Debye temperature, and γ = the Grüneisen parameter. For monovalent metals, values of λ_g calculated from eq (4) are only about 2 percent of the total experimentally determined conductivity. For the transition elements iron, platinum, and tungsten, λ_g values are about 20 percent of the total.

3. Data

Childs, et al., [5] of this laboratory have recently made an extensive compilation of thermal conductivities of materials at low temperatures. From their work, publications which contained both electrical and thermal conductivity data for the same materials were identified. These references were searched for data useful to calculate Lorenz ratios and for citations to additional measurements. This search resulted in the reference sources listed in Appendix III. The electrical resistivity and thermal conductivity data from these reference sources were extracted and converted to a common set of units (SI). Lorenz ratios were calculated and are plotted in the figures of Appendix I. The tabular values of electrical resistivity, thermal conductivity, and Lorenz ratios as reported in the literature are given in the right hand columns of Appendix II. The computed values of Lorenz ratio in SI units are given in the left hand columns.

TABLE 1. Debye temperatures of selected common metals

<u>Substance</u>	Debye temperature, θ_D (K)
Aluminum	426
Beryllium	1160
Chromium	610
Copper	344
Gold	165
Iron	464
Lead	96
Magnesium	406
Manganese	476
Molybdenum	440
Nickel	440
Platinum	240
Silver	225
Tantalum	250
Tin	195
Titanium	420
Tungsten	405
Zinc	300

In many instances, data had to be read from small graphs. The error introduced by reading these graphs was sometimes above 10%. This is unfortunate especially in those instances where the uncertainty of the original experimental data was smaller than 10%. In some instances where data are closely spaced, curves were drawn to represent these data. No attempt was made to smooth the data or to present best values. Considerably more work, both experimental and theoretical, must be done before smoothing can be done with confidence.

It is difficult to assess the uncertainty of thermal conductivity data. Probably the best one should expect is about 1% and the worst (except for blunders) is about 50%. Many sources list uncertainties between 5 and 10%. The Lorenz ratios will have comparable uncertainties since the uncertainty in electrical resistivity is generally much smaller. Childs, et al., [5] made estimates of thermal conductivity uncertainty for each of the references. These estimates are not repeated here.

4. Predictive Procedures

In order to use the Lorenz ratio data presented here in predicting thermal conductivities of new materials one must associate this material with one whose $L(T)$ is already known and measure the electrical resistivity of the new material. Then one may compute

$$\lambda = \frac{LT}{\rho}.$$

It is stressed that obtaining ρ is significantly easier than measuring λ . The association of one material with another may be done primarily from a knowledge of its composition but also from other characterization parameters such as electrical resistivity, hardness, density, or crystalline structure. The thermal and mechanical histories of the specimen are important as well.

In the event that no similar material has been measured, generally less accurate techniques of prediction may be considered. For purposes of prediction, divide materials into the following classes:

- (1) pure metals,
- (2) high conductivity alloys,
- (3) low conductivity alloys,

and consider temperatures in three ranges:

- (a) $T > \theta_D$,
- (b) $T < \theta_D$,
- (c) $T \ll \theta_D$.

We will consider pure metals first. At temperatures above θ_D or much less than θ_D (say $0.01 \theta_D$), one can make reasonable estimates of thermal conductivity of pure metals by calculating λ_e as $L_o T/\rho$. The lattice conductivity, λ_g , is computed from eq(4) and the total conductivity λ , from eq(1). The contribution of the lattice conductivity for pure metals is usually negligible. The electrical resistivity, $\rho(T)$, can often be obtained directly from the literature [6, 7] or from a measurement of ρ_o and values of $\rho_i(T)$, also found in the literature. At temperatures from about $0.2 \theta_D$ to θ_D , L and L_e are temperature dependent but not appreciably dependent on purity. Thus, one can use the L_e curve obtained by White [2], shown in figure 3, to calculate λ_e from $L_e T/\rho$. Between $0.01 \theta_D$ and $0.2 \theta_D$, the electronic and total Lorenz numbers are dependent on purity and temperature. No correlation between L and purity has been mathematically formulated, although it is clear that one exists. It is noted that at temperatures above about $0.2 \theta_D$, the thermal conductivity of pure metals is not strongly dependent upon the chemical or physical condition of the metal. Thus, if accurate conductivity values have been obtained for a pure metal, these values above $0.2 \theta_D$ are applicable to other specimens to within about 10%.

Next we will consider high conductivity alloys. High conductivity alloys are generally dilute alloys but may also be the result of alloying similar elements. Ordered alloys also have high conductivities. The values of L for high conductivity alloys are generally near L_0 at all temperatures. Thus, a first approximation of thermal conductivity can be obtained from $L_0 T/\rho$.

Last we will consider the low conductivity alloys. Generally, these are highly alloyed structural alloys. Prediction of the thermal conductivity is most difficult at temperatures $T \leq \theta_D$ in the low conductivity alloys. Here the lattice conductivity is comparable to and sometimes much larger than the electron conductivity. If a comparable alloy has not been measured, it is probably impossible to predict a reasonable accurate value of λ for this case. We presently are engaged in a program to improve the predictive capabilities for such alloys. It appears reasonably obvious that to do this one must be able to calculate the electron component from the electrical resistivity and the lattice component by some other means. The lattice component probably will be obtained from a modified form of eq(4), but additional research will be required before a more accurate equation is found and its limitations are understood. At temperatures above θ_D , one can obtain a first approximation of thermal conductivity, usually on the low side, from $L_0 T/\rho$. The above methods of prediction are summarized as follows:

	Pure Metals	High Conductivity Alloys	Low Conductivity Alloys
$T \ll \theta_D$	$\lambda \approx L_o T / \rho$	$\lambda \approx L_o T / \rho$	
$T < \theta_D$	$\lambda \approx L_e T / \rho$ for $0.2 \theta_D < T < \theta_D$ L_e from White [2] ----- Predictions difficult for $0.01 \theta_D < T < 0.2 \theta_D$	$\lambda \approx L_o T / \rho$	No Valid Predicting Method Known
$T > \theta_D$	$\lambda \approx L_o T / \rho$	$\lambda \approx L_o T / \rho$	$\lambda \approx L_o T / \rho_o$

5. Further Work

It is seen from this compilation that the Lorenz ratios of pure metals are near the Sommerfeld value, $L_o = 2.443 \times 10^{-8} V^{-2}/K^2$, at low temperatures (residual region), fall to lower values at intermediate cryogenic temperatures, and increase again to near L_o at high temperatures. At intermediate temperatures a dependence on impurity concentration is observed. For alloys one obtains the expected increase in Lorenz ratio at intermediate temperatures instead of a dip as for the pure metals. This increase in L is evidence of a relatively large lattice conductivity. Somewhat challenging from a predictive standpoint is the large spread in Lorenz ratios for the given classes of materials at each temperature. This large spread will result in large uncertainties in predicted thermal conductivities based on these curves. Further subdivision of materials may be necessary to achieve a sufficiently small spread to be useful.

The objective of this Lorenz ratio compilation is to separate materials into classes based on similar Lorenz ratio curves. From

these curves and a knowledge of the electrical resistivity of a new material from that class, it is often possible to approximate the thermal conductivity of the new material. This compilation has revealed some other interesting possibilities for prediction of thermal conductivities. For example, in several instances data exist in the literature which will allow computation of the specific thermal resistivity resulting from controlled addition of impurity atoms in a given host material. It has been noted that some impurities are much more effective scatterers than others. From a table of specific thermal resistivities, similar to that compiled by Blatt [8] for electrical resistivity, one may be able to accurately predict the electronic thermal conductivity of metals and alloys. Of the huge number of host-impurity element combinations which exist, only a few have been measured. However, there is a good possibility that one can construct such a table from a knowledge of $L(T)$ for pure metals and Blatt's specific electrical resistivities. The measurements which have been compiled here and by Childs, et al., [5] can be used to check the accuracy of such calculations. The available data will also be useful in examining lattice conductivity.

6. Acknowledgments

The authors express appreciation to the many people who contributed to make this work possible. Most of the program was supported by NASA (SNSO-C) Contract R-45. Steve Schmidt and Mark Prouhet assisted in data extracted from the many sources. Gregg Childs and Bob Powell were very helpful in providing a bibliography, as yet unpublished, from their thermal conductivity compilation as well as opening their reference files for our use.

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APPENDIX I - FIGURES
AND
DATA REVIEW TABLES

The use in this paper of trade names is essential to a proper understanding of the work presented. Their use in no way implies any approval, endorsement, or recommendation by NBS.

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Beryllium

CURVE	INVESTIGATOR(S) [YEAR]	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
HWP	Hast, Weitzel, and Powell (1971)	Be (98.7), Be O (1.18), Al (0.04), Ni (0.014), Mn (0.009)	Neutron irradiated and room temperature annealed, RRR = 3.83, Rockwell hardness = C-12, Grain size = 0.03 mm, axis of specimen is perpendicular to pressing axis, uncertainty = 2.5%	
L	Lewis (1929)		Annealed at 700 °C, RRR = 4	
PHC (1)	Powell, Harten, and Gibson (1960)	Be (99.5) Commercially Pure	RRR = 6.01, Uncertainty = 6%, axis of specimen parallel to pressing axis	
PHC (2)	Powell, Harten, and Gibson (1960)	Los Alamos Scientific Laboratory Be (98.7), Be O (1.2), Al (0.056), Si (0.035), Ni (0.015), Mn (0.010)	RRR = 4.28, Same specimen used by Hast, Weitzel, and Powell (1971)	
WW (1)	White and Woods (1955)	Los Alamos Scientific Laboratory Be (98.7), Be O (1.18), Al (0.04), Ni (0.014), Mn (0.009)	RRR = 4, Data taken from small graph	
WW (2)	White and Woods (1955)	Beryllium Brush Company Be, Mg (0.2)	RRR = 4, Data taken from small graph	
		A.D. Mackay (Spectrographic) Be, Mg (0.1), trace of iron		

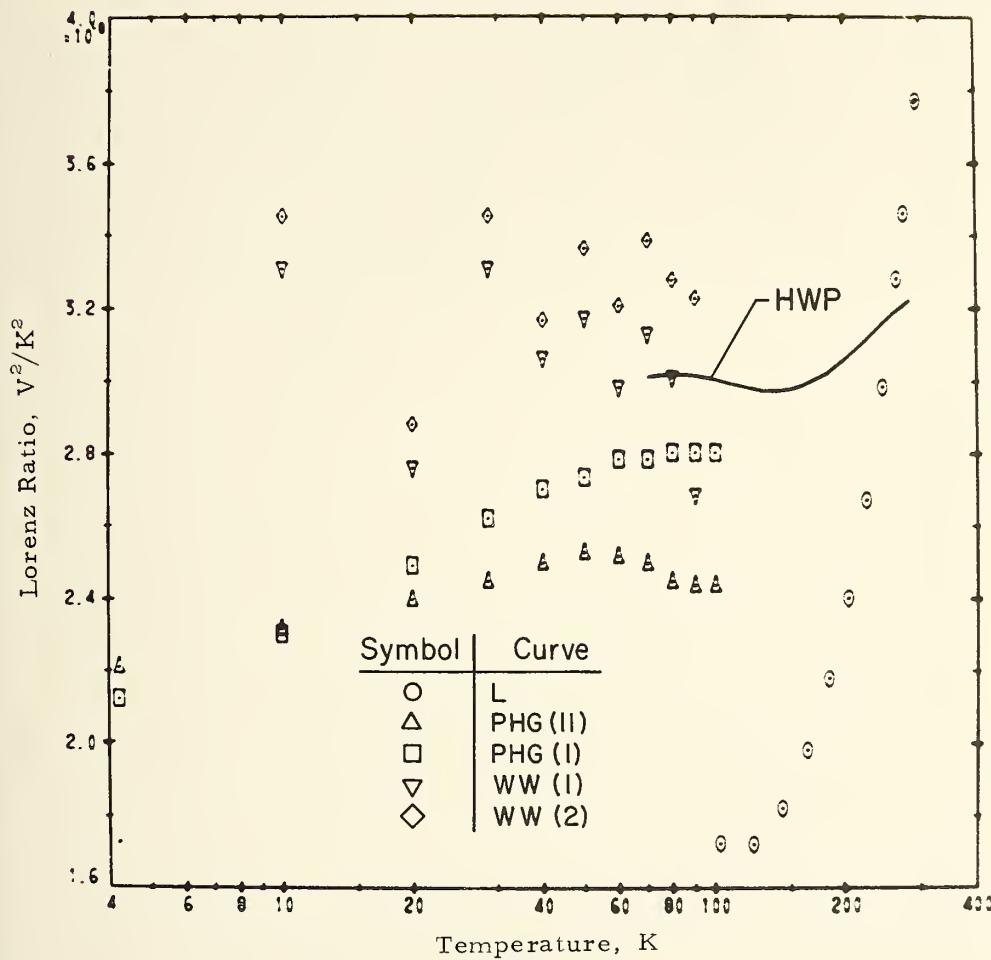


Figure 1. Lorenz ratio of beryllium

Magnesium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
	Ahrens, Spohn, and Weber (1955) Staehler (1929); Same data to Maierchen (1931)	Mg, Mn (0.04) Mg, Si (0.7) Mg, Mn (0.8) Mg, Mn (0.5)	$L = 2.64 \times 10^{-8} v^2 / K^2$ from 1.4 to 5 K $L = 1.12 \times 10^{-8} v^2 / K^2$ at 87 K $L = 2.88 \times 10^{-8} v^2 / K^2$ at 273 K $L = 2.18 \times 10^{-8} v^2 / K^2$ at 87 K $L = 3.24 \times 10^{-8} v^2 / K^2$ at 273 K $L = 1.64 \times 10^{-8} v^2 / K^2$ at 87 K $L = 3.08 \times 10^{-8} v^2 / K^2$ at 273 K

Aluminum

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		SOURCE	COMPOSITION	
ANS	Aamundsen, Nytre, and Salter (1972)	Pure single crystal of Aluminum, Zone refined		RER = 210, uncertainty = 2%
ANS	Aamundsen, Nytre, and Salter (1972)	Pure single crystal of Aluminum, Zone refined		RER = 2250, uncertainty = 2%
ANS	Aamundsen, Nytre, and Salter (1972)	Pure single crystal of Aluminum, Zone refined		RER = 3490, uncertainty = 2%
ANS	Anilnews, Webster, and Spoerl (1951)	Aluminum Company of America		RER = 840, uncertainty = 2%
		Al (99.99%), Mg (0.001), Si (0.001), Fe (0.0004), Na (0.0004)		$L = 2.21 \times 10^{-3} \frac{V^2}{K^2}$ at 1.2 K
		single crystal		$L = 1.55 \times 10^{-3} \frac{V^2}{K^2}$ at 1.5 K
ANS	Anilnews, Webster, and Spoerl (1951)	Aluminum Company of America		RER = 673, uncertainty = 2%
		Al (99.99%), Mg (0.001), Si (0.001), Fe (0.0004), Cu (0.0004), Na (0.0004)		
		single crystal		
ANS	Andrews, Webster, and Spoerl (1951)	Johnson and Wattley		RER = 176, uncertainty = 2%
		Al (99.999%), Mg (0.002), Si (<0.001), Fe (<0.0005), Cu (<0.0005), Na (Faint trace)		
		polycrystal		
FNW	Fenton, Rogers, and Roode (1951)	Consolidated Mining and Smelting Company		Act. steamed, 350°C in air for 10 minutes
		Al (99.9999%)		RER = 2770, uncertainty = 1.5%
		polycrystal		Data read from small graph
FNW	Fenton, Rogers, and Roode (1951)	Consolidated Mining and Smelting Company		Act. cleaned, 350°C in air for 10 minutes
		Al (99.9999%)		RER = 1.75, uncertainty = 1.5%
		polycrystal		Data read from small graph
FNW	Fenton, Rogers, and Roode (1951)	Recrystallized Aluminum (specimen 3)		$L = 1.7 \times 10^{-3} \frac{V^2}{K^2}$ at 21.2 K
		Al (99.9999%)		$L = 1.5 \times 10^{-3} \frac{V^2}{K^2}$ at 33.2 K
FNW	Grunefeld and Goens (1927)	Aluminum, 2.5 hours at 300°C in vacuum, Specimen 2		$L = 1.77 \times 10^{-3} \frac{V^2}{K^2}$ at 21.2 K
				$L = 1.27 \times 10^{-3} \frac{V^2}{K^2}$ at 33.2 K
				$L = 2.18 \times 10^{-3} \frac{V^2}{K^2}$ at 21.2 K
				$L = 1.47 \times 10^{-3} \frac{V^2}{K^2}$ at 33.2 K
				$L = 2.20 \times 10^{-3} \frac{V^2}{K^2}$ at 21.2 K
				$L = 1.55 \times 10^{-3} \frac{V^2}{K^2}$ at 33.2 K

Aluminum (Cont.)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS		
L	Gruneisen and Goens (1927)	Aluminum, specimen 21	$L = 2.20 \times 10^{-5} V^2/K^2$ at 21.2 K $L = 1.66 \times 10^{-5} V^2/K^2$ at 83.2 K		
	Lees (1908)	Johnson and Matthiessen Company Al (99)	Density $\approx 2.70 \text{ g/cm}^3$ at 20°C		
PoB	Moore, McElroy, and Barroni (1966)	Reynolds Aluminum Company Al (99.999)	$R_{RR} = 520$, uncertainty $\pm 2\%$		
PRR (sc)	Powell, Hall, and Roder (1960)	Johnson - Matthiessen Company Al (99.99%) sample crystal	2 hours at 400°C in vacuum, (001) 6° from axis of rod, $R_{RR} = 100$		
PRR (1100-O)	Powell, Hall, and Roder (1960)	Aluminum Company of America Al (commercial purity 1100-O); Si (0.13); Cu, Ga, Fe, Mg, V (0.1); Cr (0.02); Pb, Mn, Sn, Ti (0.01); Ca, Zr (0.001)	1 hour at 350°C in vacuum, RRR ≈ 12 , Hardness (DPH) ≈ 22 , grain size $\approx 0.024 \pm 0.008 \text{ mm}$ (long.), grain size $\approx 0.012 \text{ mm}$ (trans.)		
PTW	Powell, Tyse, and Woodward (1965)	Al (99.993)			
	Staebler (1929)	Aluminum	$L = 2.075 \times 10^{-8} V^2/K^2$ at 39 K $L = 2.233 \times 10^{-8} V^2/K^2$ at 273 K		
	Wilkes and Powell (1968)	Advanced Research Materials Al (99.9999), Cu (0.5 ppm), Si (0.5 ppm), Mg (0.1 ppm)	Data received Density $\approx 2.698 \text{ g/cm}^3$ at 23°C Data read from a small graph		

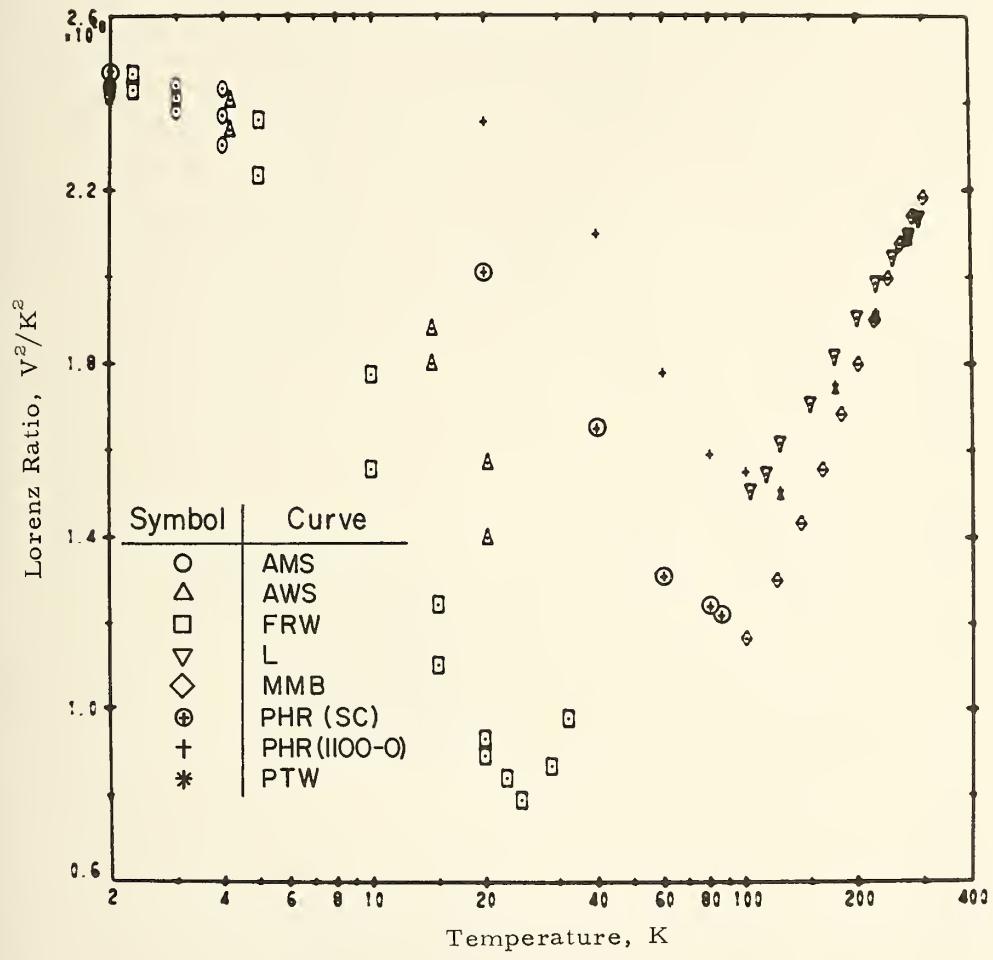


Figure 2. Lorenz ratio of aluminum

Aluminum Alloys

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
RS (796)	Hust and Sparks (1971c)	Al 2024 Cu (4.1), Mg (1.4), Mn (0.5), Fe (0.2), Si (0.1), Zn (0.1), Al (bal.)	RHR = 2.6, Density = 2.768 g/cm ³ , grain size = 0.025 mm Rockwell hardness = 833	
HS (Ann.)	Hust and Sparks (1970c)	Al 2024 Cu (4.1), Mg (1.4), Mn (0.5), Fe (0.2), Si (0.1), Zn (0.1), Al (bal.)	full anneal condition, in vacuum, 1 hour at 427°C, 6 hours at 232°C, grain size = 0.035 mm, Rockwell hardness = 836, Density = 2.768 g/cm ³	
HWP	Hust, Weitzel, and Powell (1971)	Al 7039 Al (93), Zn(3.5), Mg(2.55), Mn (0.23), Cr(0.20); Fe, Cu, Si, Ti, Be (<0.1)	RHR = 2.6, Rockwell hardness = 875, grain size = 0.005 × 0.05 mm uncertainty = 2%	
HWP	Hust and Powell (1963)			
HWP	Hust, Powell, and Weitzel (1969)			
PWR (5053)	Powell, Hall, and Roder (1960)	Al 6063 - T5 Al (69.5), Mg (0.65), Si (0.38); Ga, Fe, Mn (0.1); Cr, Cu, Ti, V, Zn (0.01); Ca, Pb (0.001)	RHR = 10, DP hardness = 30, grain size = 0.05 mm	
PWR (5154)	Powell, Hall, and Roder (1960)	Al 5154 - O Al (90), Mg (3.32), Cr (0.21); Cu, Fe, Mn, Si (0.1); Ti, V, Zn, Ar (0.01); Ga, Pb (0.001)	RHR = 2.5, DP Hardness = 35, grain size = 0.044	
PWR (5086)	Powell, Hall, and Roder (1960)	Al 5086 - F Al (91), Mg (4.10), Mn (0.51), Fe (0.28); Cr, Si, Zn (0.1); Cu (0.07), Ti (0.02)	RHR = 1.3, DP Hardness = 22, grain size = 0.07 × 0.02 mm	
PWR (5052)	Powell, Hall, and Roder (1960)	Al 5052 - O Al (97), Mg (2.46), Cr (0.22); Cu, Ga, Fe, Mn, Si, Zn (0.1); Ti, V (0.01); Zr (0.001)	RHR = 2.5, DP Hardness = 28, grain size = 0.03 mm	
PWR (2024)	Powell, Hall, and Roder (1960)	Al 2024 - T4 Al, Cu (4.53), Mg (1.70); Ga, Fe, Mn, Si, V, Zn(0.1); Cr(0.05); Sn, Ti(0.01); Cs, Ag, Zr(0.001)	RHR = 1.9, DP Hardness = 66, grain size = 0.06 mm	

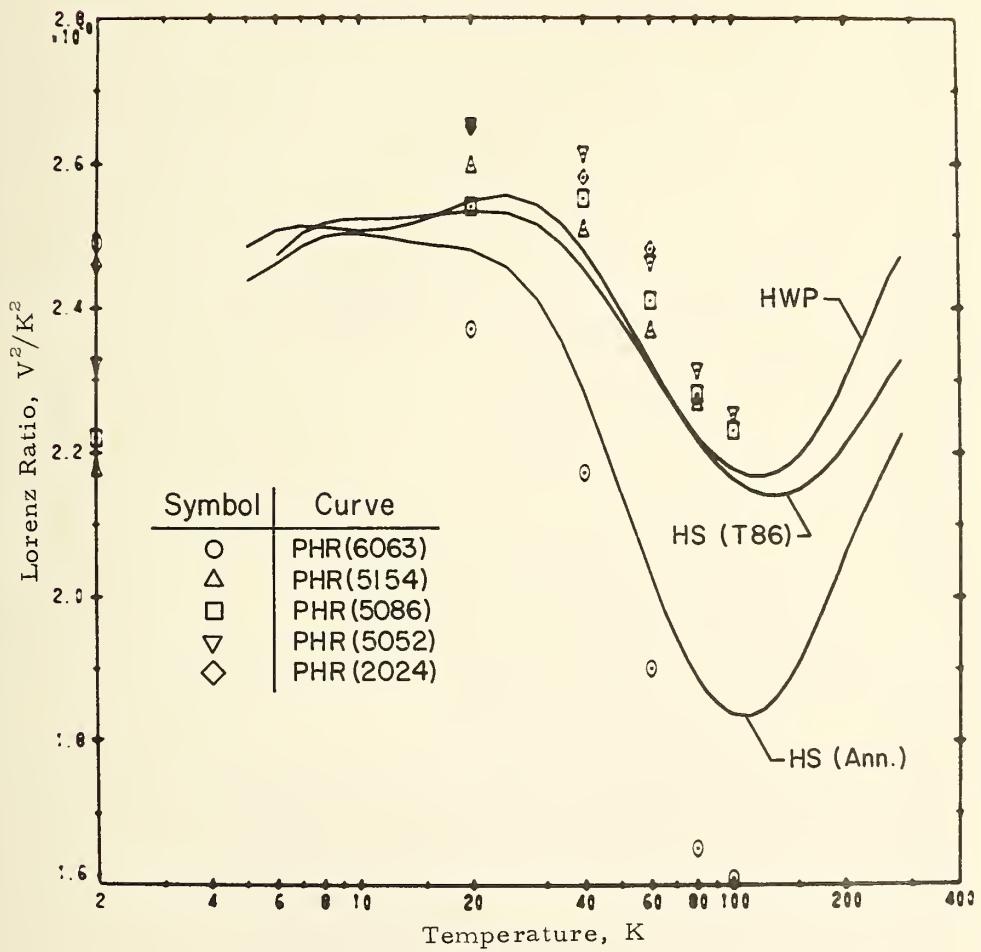


Figure 3. Lorenz ratio of aluminum alloys

Lead and Tin

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
			DENSITY	REFERENCE
L (Pb)	Lees (1905)	Pb-cadmalite Pure Pb	Density = 11.29 g/cm ³ at 25°C	
L (Sn)	Lees (1905)	Kahlbaum Pure Sn	Density = 7.28 g/cm ³ at 21 °C	

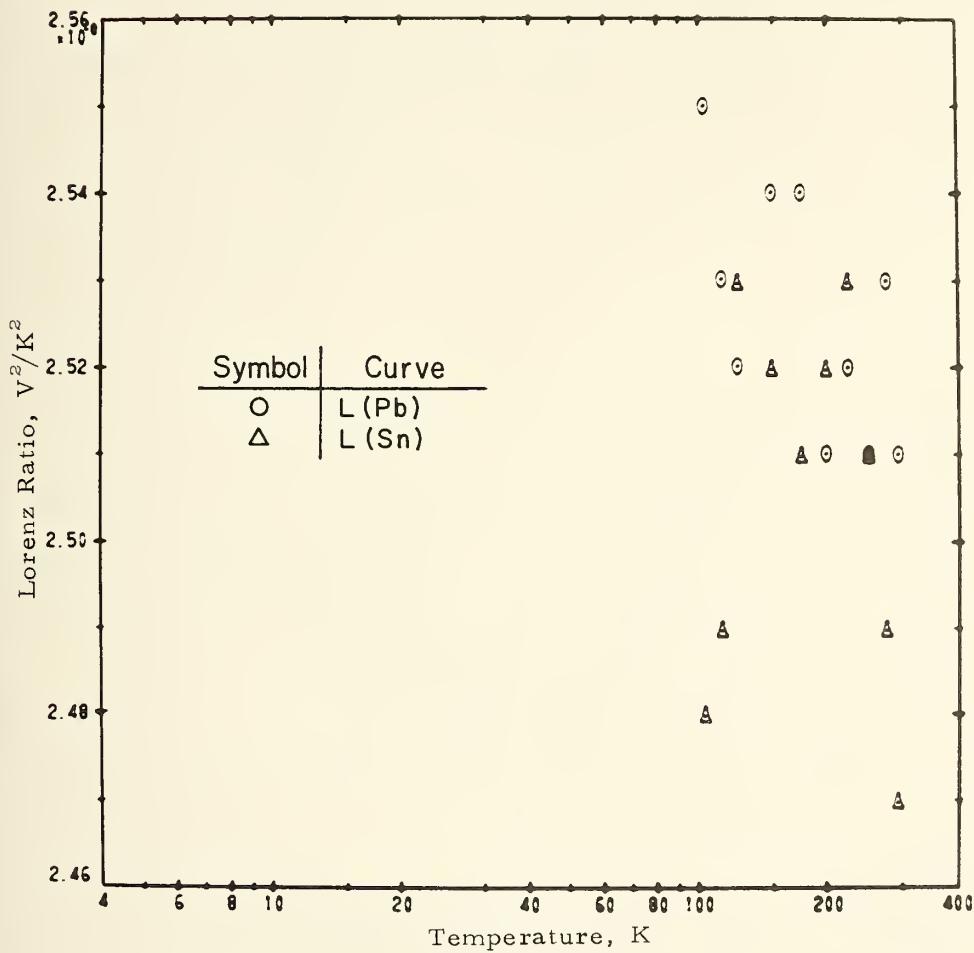


Figure 4. Lorenz ratio of lead and tin

Gold

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
			L	
	Gruneisen and Goens (1927)	Au, single crystal, very pure, beaten, V. Mg/tun	$L = 1.05 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 1.05 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
	Gruneisen and Goens (1927)	Au, single crystal, very pure, unheated, Dieselic	$L = 1.16 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 1.99 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
	Gruneisen and Goens (1927)	Au, single crystal, Dieselic, 5.5 hours at 320°C	$L = 1.08 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 1.95 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
	Gruneisen and Goens (1927)	Au, technically pure, untempered, Silberscheidanstalt	$L = 1.05 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 2.09 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
	Gruneisen and Goens (1927)	Au, technically pure, Dieselic, 3 hours at 390°C	$L = 1.65 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 2.05 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
	Gruneisen and Goens (1927)	Au, untempered, very unpure, Dieselic	$L = 2.6 \times 10^{-5} \frac{V^2}{K^2}$ at 21.2 K $L = 3.0 \times 10^{-5} \frac{V^2}{K^2}$ at 33.2 K	
HS (1)	Hast and Sparks (1971)	Ag/Al, Au (99.99%)	Density = 25.20 g/cm³, DP Hardness = 50, Grain size = 0.001 mm, RRR = 93	
HS (2)	Hast and Sparks (1971)	Same specimen after 7% reduction in area	Density and DP Hardness are the same, grain size = 0.009 mm, RRR = 57	
HS (3)	Hast and Sparks (1971)	Same specimen annealed for 2 hours at 400°C in a vacuum	Density = 19.28 g/cm³, DP Hardness = 29, grain size = 0.013 mm, RRR = 209	
	Ramdhalik (1931)	Hercus	Uncertainty = 5%	
		Au (99.99)	$L = 2.39 \times 10^{-5} \frac{V^2}{K^2}$ at 273 K	

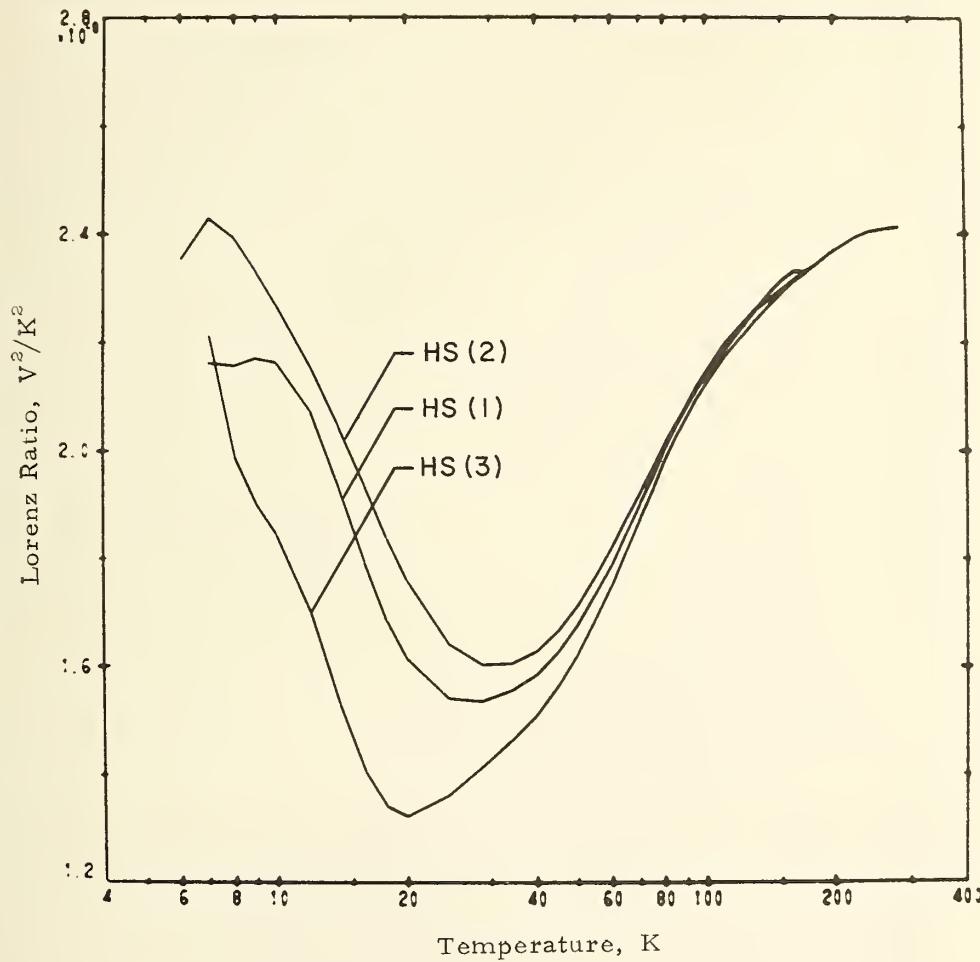


Figure 5. Lorenz ratio of gold

Gold Cobalt Alloy

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		SG	Co	
FEC	Pouelli, Bunch, and Gibson (1950)	SG: gold Cobalt Corporation Hard drawn Au-2.1% Co thermocouple wire		

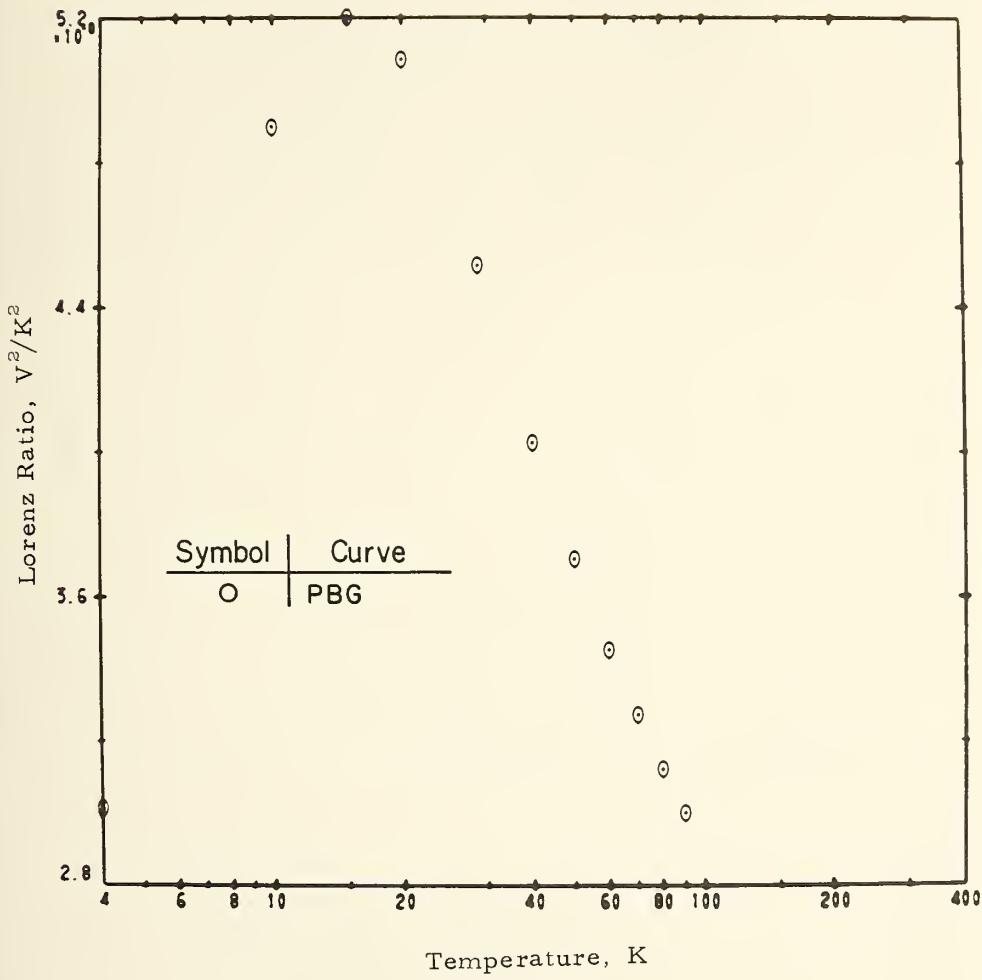


Figure 6. Lorenz ratio of gold-cobalt alloy

Silver

CURVE	INVESTIGATOR(S) [YEAR]	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION · REMARKS
		SOURCE	COMPOSITION (%)	
FRW (1)	Fenton, Rogers, and Woods (1963)	Engelhard		3 hours at 650°C in vacuum, RRR = 1.890
FRW (2)	Fenton, Rogers, and Woods (1963)	Pure Ag (specimen 1) Engelhard		3 hours at 650°C in vacuum, RRR = 2.200
K (1)	Kamaluk (1933)	Pure Ag (specimen 2) Hilger, Ltd.		Drawn
K (2)	Kamaluk (1933)	Ag		Specimen with 500°C anneal
L	Kamaluk (1933)	Ag (Commercially Pure, electrolytic)		$\lambda = 2.32 \times 10^{-8} \text{ } \mu^2/\text{K}^2$ at 273 K
	Kamaluk (1933)	Ag (Spectrographic purity - contaminated)		$\lambda = 2.41 \times 10^{-8} \text{ } \mu^2/\text{K}^2$ at 273 K
M	Lees (1963)	Johnson - Matthey Company		Density = 10.47 at 21°C
N	Malm and Woods (1966)	Ag (99.9%)		4 hours at 750°C in vacuum, RRR = 1.030
N	Malm and Woods (1966)	Consolidated Mining and Smelting Company of Canada Ag (99.999%)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	Johnson - Matthey Company Ag (99.999%)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	MG (99.999%)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	Ag + Mg (0.005)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	Ag + Mg (0.067)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	Ag + Mg (0.11)		4 hours at 750°C in vacuum
N	Malm and Woods (1966)	Ag + Mg (0.31)		4 hours at 750°C in vacuum

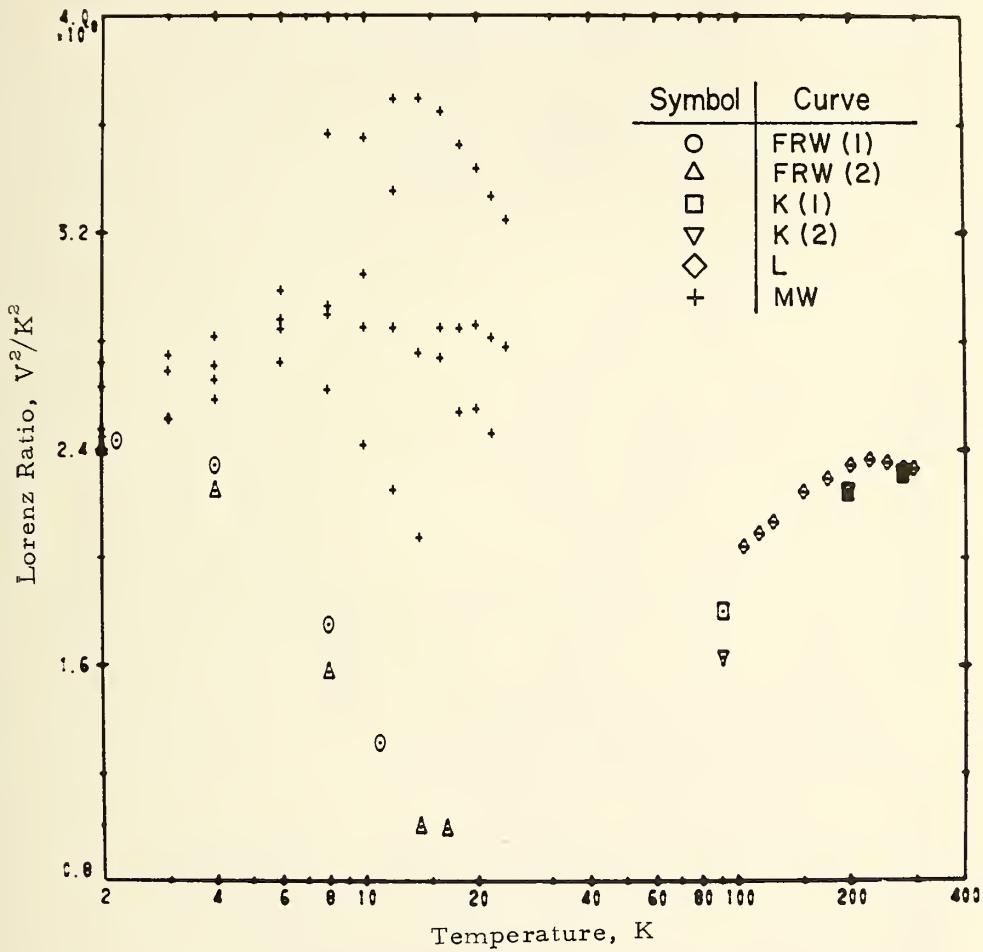


Figure 7. Lorenz ratio of silver

Copper

CURVE	INVESTIGATOR(S)	(YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
			Source	Composition	
	Allen and Mendozza (1947)		Johnson - Mettler Company Cu (99.99); As, Mn, Pb (0.003)		$L = 3.43 \times 10^{-3} \frac{V^2}{K^2}$ from 2 to 4 K
	Aoyama (1946)		Electrolytic Cu		$L = 1.649 \times 10^{-3} \frac{V^2}{K^2}$ at 87.15 K
	Gruwezen and Goense (1927)		Cu		An extensive series of coppers in various states of anneal and chemical imperfection was measured at 21.2 K and 83.2 K. Values of L varied from 0.77×10^{-3} to $2.03 \times 10^{-3} \frac{V^2}{K^2}$ at 21.2 K and from 1.56×10^{-3} to $1.62 \times 10^{-3} \frac{V^2}{K^2}$ at 83.2 K
K	Kemp, Klemens, and Tainsh (1959b)		Cu (99.55), As (0.35), P (0.05) Specimen 0		Prolonged anneal at 450°C, RRR = 1.6
	Kemp, Klemens, and Tainsh (1959b)		Cu (99.55), As (0.35), P (0.05) Specimen 1		$L = 3.23 \times 10^{-3} \frac{V^2}{K^2}$ at 4.2 K $L = 3.16 \times 10^{-3} \frac{V^2}{K^2}$ at 90 K
	Kemp, Klemens, and Tainsh (1959b)		Cu (99.55), As (0.35), P (0.05) Specimen 4		same as specimen 0 but deformed, RRR = 1.6
L	Lees (1903)		Soft drawn high conductivity Cu		$L = 2.32 \times 10^{-3} \frac{V^2}{K^2}$ at 4.2 K $L = 2.97 \times 10^{-3} \frac{V^2}{K^2}$ at 90 K
	Moore, McElroy, and Graves (1967)		National Research Council, Ottawa		same as 1 but reannealed at 450°C, RRR = 1.6
	Powell, Roiter, and Rall (1959)		Cu (99.99)		$L = 2.64 \times 10^{-3} \frac{V^2}{K^2}$ at 4.2 K $L = 3.20 \times 10^{-3} \frac{V^2}{K^2}$ at 90 K
PRH (CD)	Powell, Roiter, and Rall (1959)		Central Research Laboratory of American Smelting and Refining Company Cu (99.99)		Density = 8.84 g/cm^3 at 23 °C
PRH (AM)	Powell, Roiter, and Rall (1959)		Central Research Laboratory of American Smelting and Refining Company Cu (99.99)		Cold drawn for 25% reduction in area - no post anneal RRR = 100, uncertainty = 2%
WT	White, and Tainsh (1960)		American Smelting and Refining Company Cu (99.999), Fe (1 ppm), Sb (1 ppm), Se (1 ppm), Te (2 ppm), As (2 ppm)		Annealed for 2 hours at 400°C in vacuum, RRR = 1350, uncertainty = 2% Several hours at 550°C RRR = 2000

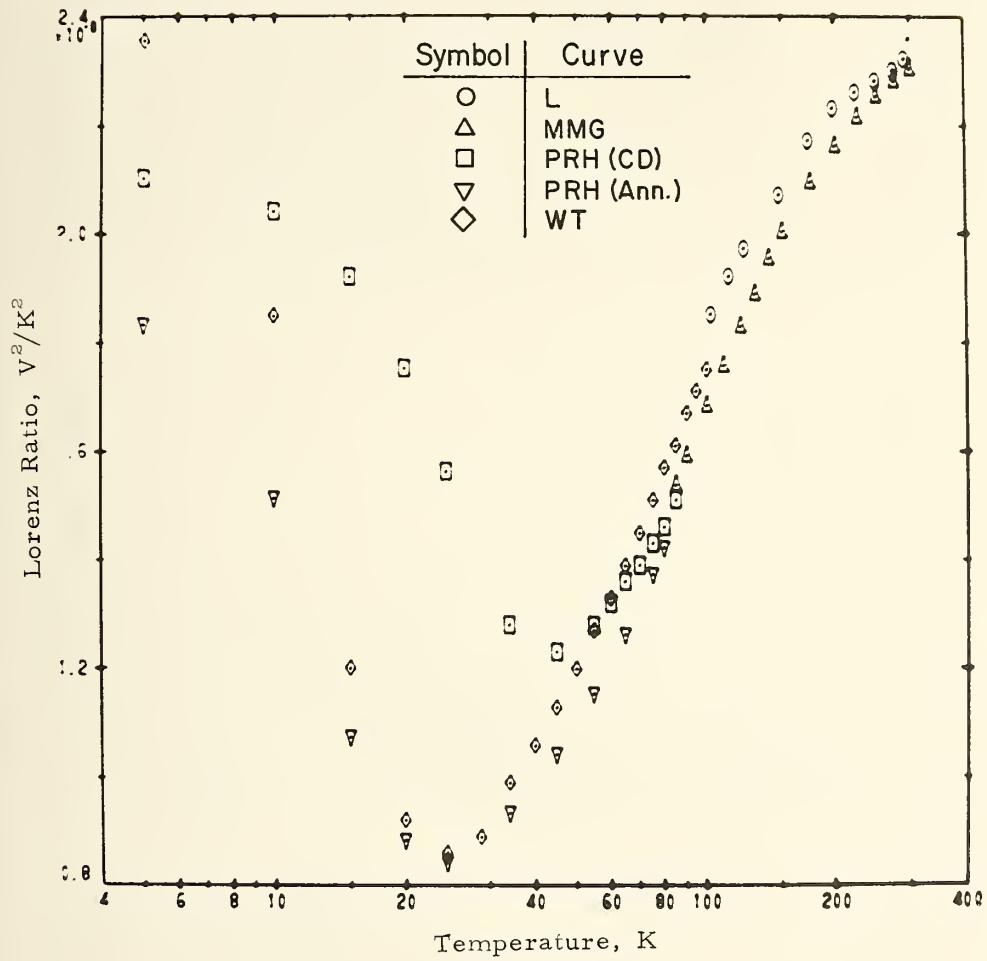


Figure 8. Lorenz ratio of copper

Copper Alloys (German Silver & Brass)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
AM	Allen and Mendoza (1947)	German Silver		
B (G-Ar.)	Berman (1951)	Cu(5.9), Zn(42.1), Ni(9.8), Fe(2.0), Fe(0.15), Ni(0.05)		Grain size \approx 0.02 mm
B (con)	Berman (1951)	German Silver		
KS (SE)	Karweil and Schaefer (1939)	Cu(47), Zn(41), Pb(2), Ni(9)		
KS (NS)	Karweil and Schaefer (1939)	Constantan		
L (G-kt)	Lees (1908)	Cu(60), Ni(15), Zn (22)		
L (Pd)	Lees (1908)	Platinoid (Similar to German Silver)		
	Kemp, Klemens, and Tahmeh (1957)	Cu (62), Ni (15), Zn (22)	4 hours at 890°C	
	Kemp, Klemens, and Tahmeh (1957)	Cu (95), Zn (5)	4 hours at 890°C	
	Kemp, Klemens, and Tahmeh (1957)	Cu (90), Zn (10)	4 hours at 890°C	
	Kemp, Klemens, and Tahmeh (1959)	Alpha brass, deformed	4 hours at 890°C	No anneal, RRR = 1.58
		Cu (53), Zn (32)	4 hours at 890°C	$L = 2.65 \times 10^{-8} V^2/K^2$ at 5 K $L = 2.69 \times 10^{-8} V^2/K^2$ at 90 K
				$L = 2.59 \times 10^{-8} V^2/K^2$ at 5 K $L = 2.41 \times 10^{-5} V^2/K^2$ at 90 K
				$L = 2.68 \times 10^{-8} V^2/K^2$ at 5 K $L = 2.69 \times 10^{-8} V^2/K^2$ at 90 K
				$L = 2.72 \times 10^{-8} V^2/K^2$ at 5 K $L = 2.80 \times 10^{-8} V^2/K^2$ at 90 K
				$L = 2.65 \times 10^{-8} V^2/K^2$ at 4.2 K $L = 2.61 \times 10^{-8} V^2/K^2$ at 90 K

Copper Alloys (Cont.)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS		
			RER	L	
	Kemp, Klemens, and Tainsh (1959)	Alpha brass, deformed Cu(68), Zn(32), 250°C	RER = 1.64 $L = 3.00 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 4.2 K $L = 2.74 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 90 K		
	Kemp, Klemens, and Tainsh (1959)	Alpha brass, deformed Cu (68), Zn (32), 290°C	RER = 1.69 $L = 2.79 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 4.2 K $L = 2.67 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 90 K		
	Kemp, Klemens, and Tainsh (1959)	Alpha brass, deformed Cu (68), Zn (32), 400°C	RER = 1.72 $L = 3.49 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 4.2 K $L = 2.78 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 90 K		
L	Lees (1908)	Brass Cu (70), Zn (30)	Density = 8.44 at 22°C		

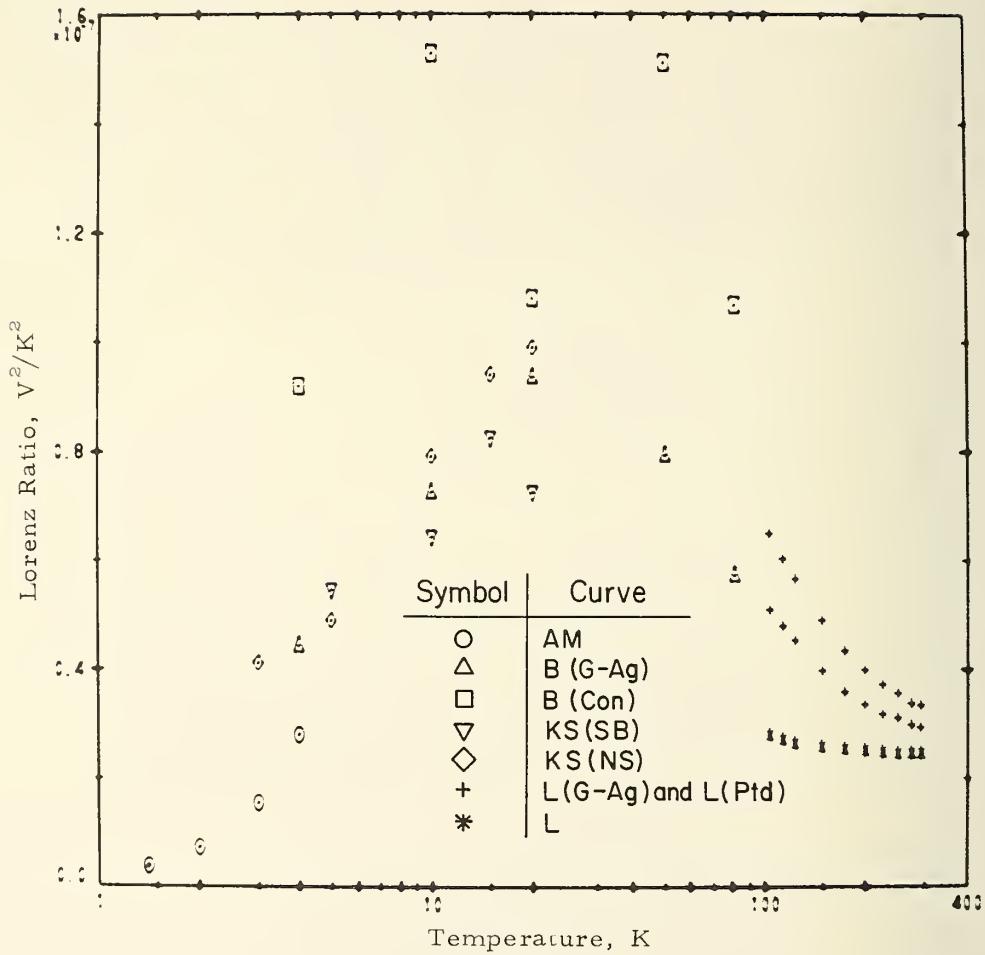


Figure 9. Lorenz ratio of copper alloys
(German silver and brass)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
CG (ZnI)	Goens and Gruneisen (1932)	Zn (pure)	Perpendicular to C axis	
CG (ZnII)	Goens and Gruneisen (1932)	Zn (pure)	Parallel to C axis	
L (Zn)	Lees (1998)	Zn (pure, redistilled)	Density = 7.10 g/cm ³ at 21°C $L = 2.51 \times 10^{-3} \cdot T^2/K^2$ at 83.2 K	
	Staebler (1929)	Zn	$L = 2.60 \times 10^{-3} \cdot T^2/K^2$ at 273 K	
CG (CaII)	Goens and Gruneisen (1932)	ca (pure)	Parallel to C axis	
CG (CaI)	Goens and Gruneisen (1932)	ca (pure)	Perpendicular to C axis	
	Lees (1998)	Ca (pure, redistilled)	Density = 8.56 g/cm ³ at 21°C	

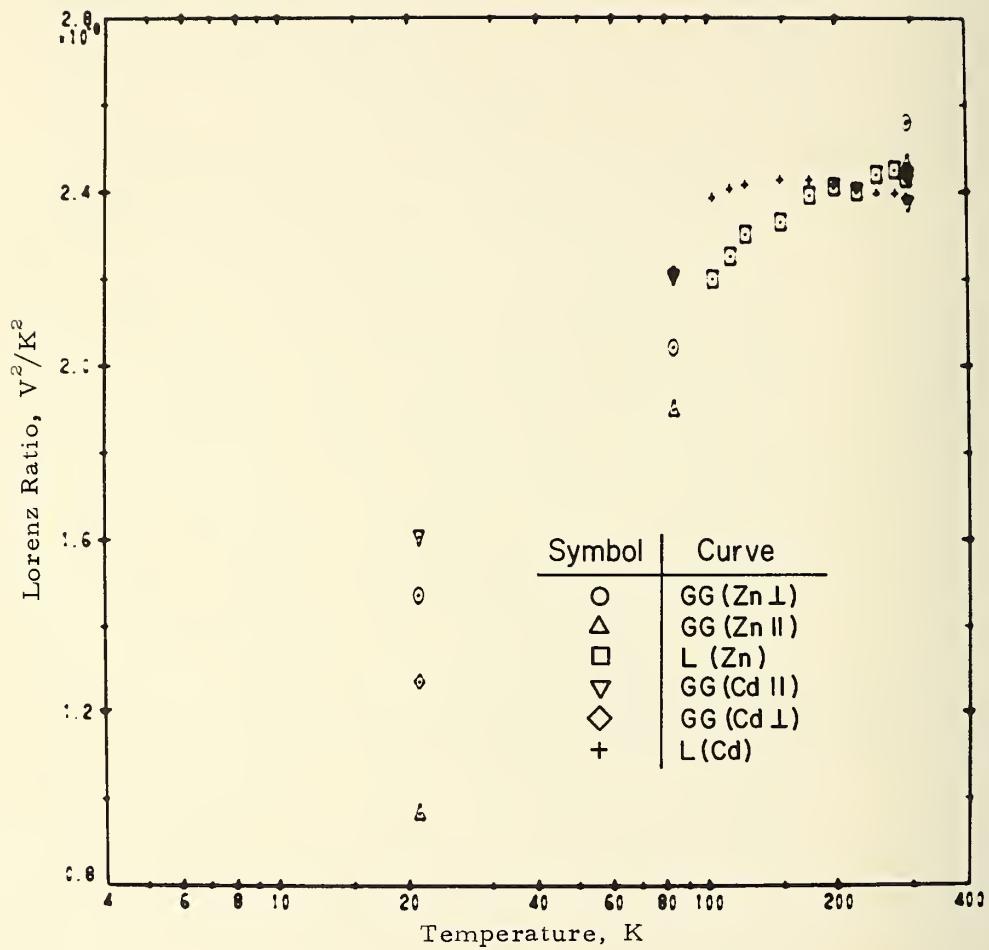


Figure 10. Lorenz ratio of zinc and cadmium

Scandium and Yttrium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
AC	Aliyev and Volkenstein (1965)	Sc (99.9)	RRR = 9.5 $L = 2.96 \times 10^{-8} V^2/K^2$ at 4.2 K	
	Aliyev and Volkenstein (1965)	Y (99.9)	RRR = 7.3 $L = 3.00 \times 10^{-8} V^2/K^2$ at 4.2 K	
	Arnold and Colvin (1954)	Sc	RRR = 4 Uncertainty = 5% $L = 2.70 \times 10^{-8} V^2/K^2$ at 291 K	
	Powell and Jolliffe	Johnson - Matthey Company, Ltd. Sc (high purity)	Parallel to C axis RRR = 14	
	Tanarit, Chuprikov, and Shalyt (1959)	Y (99.7), Cu (0.002), Pb (0.001), O ₂ (0.15), Ga (0.01); Dy, H ₂ , T _D , He (< 0.001); N ₂ (0.12)		
	Tanarit, Chuprikov, and Shalyt (1959)	Y(99.7), Cu(0.002), Pb(0.001), O ₂ (0.15), Ga(0.01), Ga(0.001); N ₂ (0.12)	Perpendicular to C axis RRR = 9	

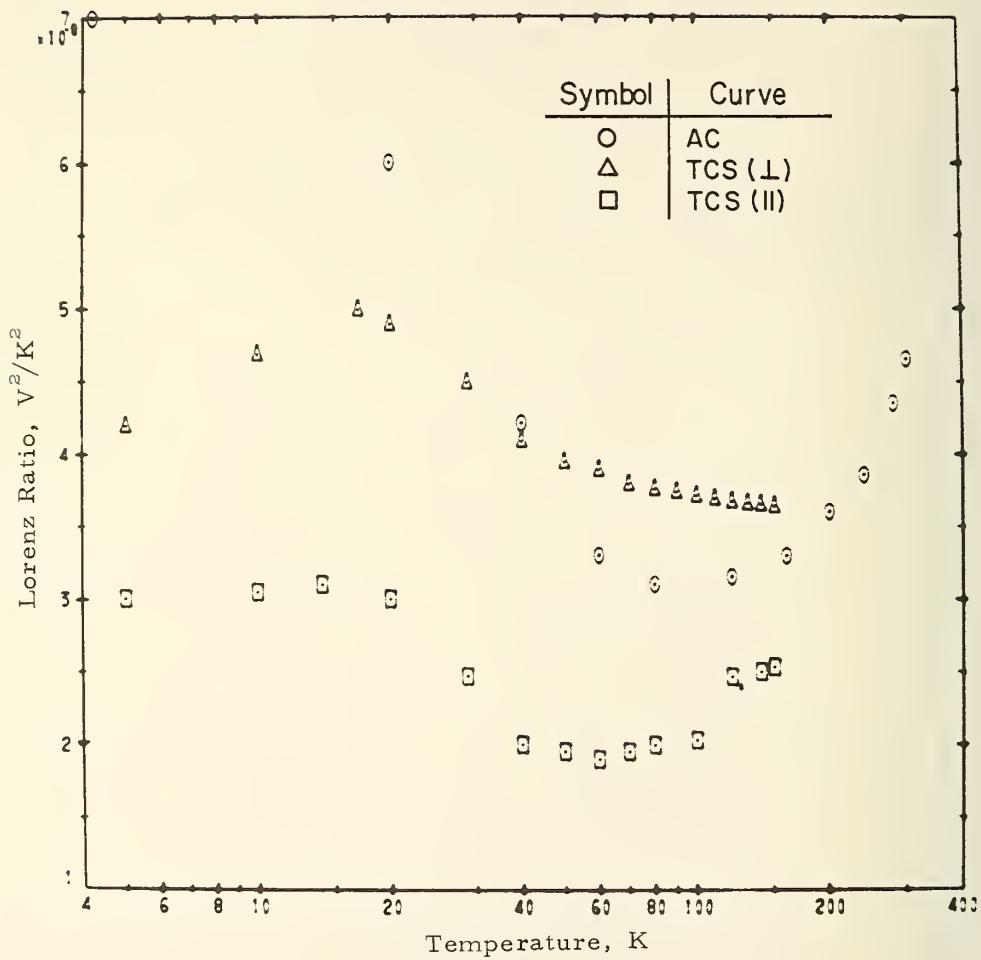


Figure 11. Lorenz ratio of scandium and yttrium

Titanium, Hafnium, and Zirconium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		SOURCE	COMPOSITION (%)	
KGW (Ti)	Kemp, Clements, and White (1956)	Johnson - Matthey Company Ti (98)		5 hours at 950°C RRR = 2.4
KGW (Zr)	Kemp, Clements, and White (1956)	Johnson - Matthey Company Zr (99.99)		5 hours at 950°C RRR = 2.9
Powell and Dye (1951)		Ti {very high purity}		MP Hardness ≈ 60
WD (HP)	White and Woods (1957a)	Foote Mineral Company Hf (99.3), Zr (0.7)		$L = 3.21 \times 10^{-8} \frac{V^2}{h^2}$ at 193 K RRR = 8

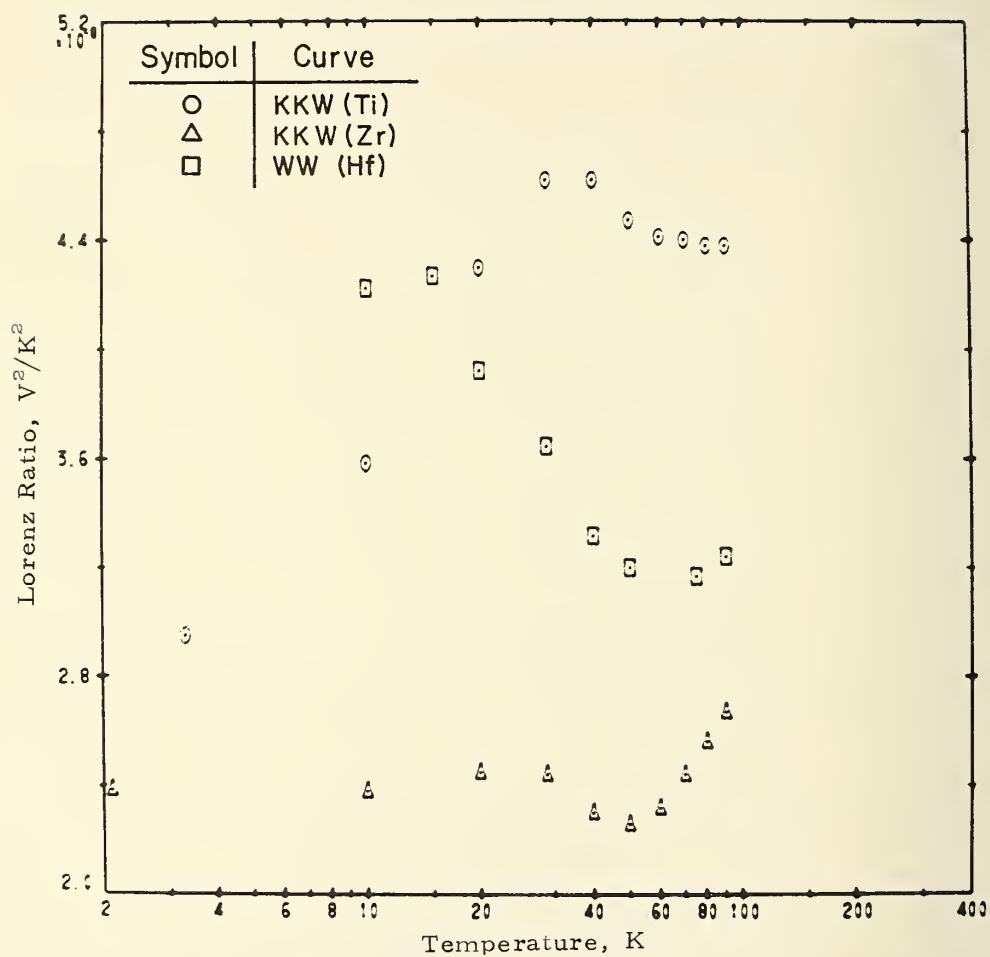


Figure 12. Lorenz ratio of titanium, hafnium, and zirconium

Titanium Alloys

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
			Rockwell hardness = C32, grain size = 0.015 mm, annealed	
HWP	Rust, Weitzel, and Powell (1967) and (1971) Rust, Powell, and Weitzel (1969) Rust and Powell (1968)	Ti - Al 0.4Ti 2.5Sn Ti (9.5), Al (5.5), Sn(2.5), Fe (0.2), Cr, N, H, C		

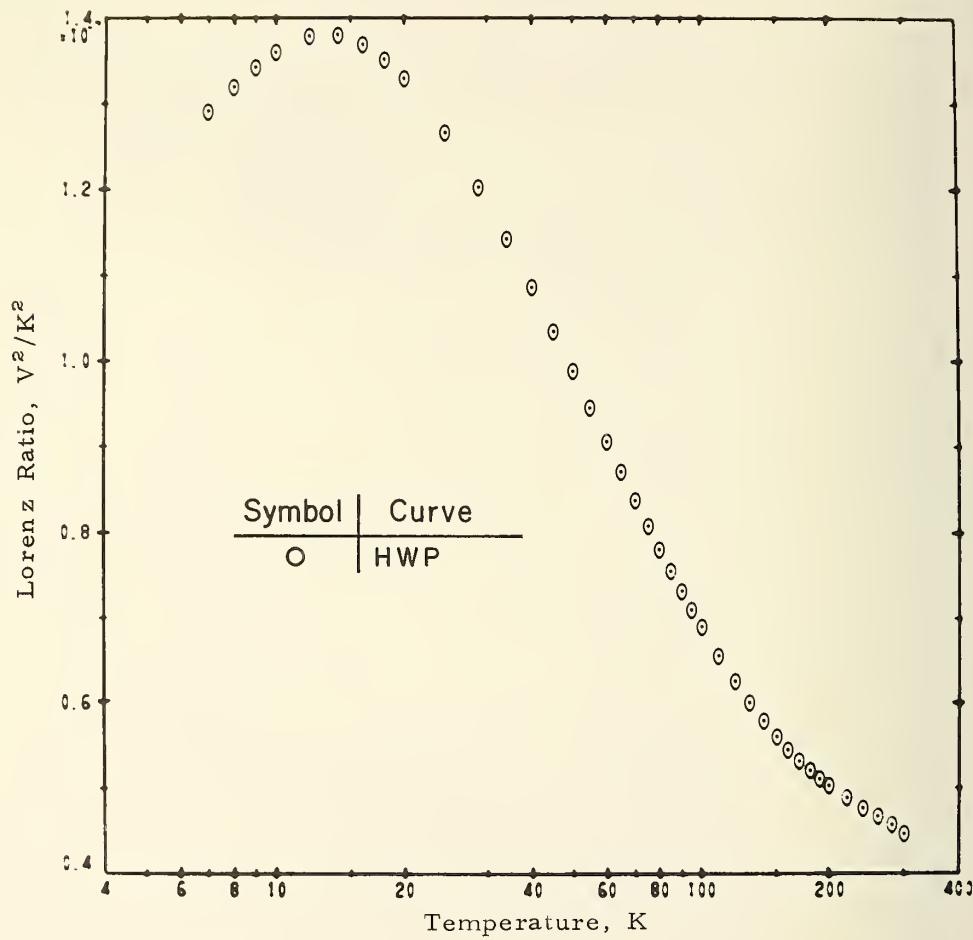


Figure 13. Lorenz ratio of titanium alloys

Tungsten

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		C	H	
B	Baeklund (1967)	Johnson - Matthey Company, Ltd. W (Spectroscopically standardized)		RRR = 3.9 uncertainty = 1.5%
DD	Defasis and Delobbe (1938)	W		Single crystal (111) direction RRR = 100
	Grundsen and Goens (1927)	W (very pure)		Single crystal
	Gruneisen and Geens (1927)	W (Impure)		L = 0.95 × 10 ⁻⁸ V ² /K ² at 21.2 K L = 1.9 × 10 ⁻⁸ V ² /K ² at 83.2 K
	Kammlukk (1931)	General Electric		L = 2.26 × 10 ⁻⁸ V ² /K ² at 21.2 K L = 2.25 × 10 ⁻⁸ V ² /K ² at 83.2 K
	Kammlukk (1931)	W (Impure)		230°C anneal L = 3.58 × 10 ⁻⁸ V ² /K ² at 273 K
	Kammlukk (1931)	General Electric		1300°C anneal L = 3.44 × 10 ⁻⁸ V ² /K ² at 273 K
K	Kammlukk (1933)	W (99.97); Co, Cr, In, O ₂ , Pt, Sn (0.001); Si, Te, V (0.01); Se (trace)		Single crystal (100) direction
K	Kammlukk (1933)	W (99.97); Co, Cr, In, O ₂ , Pt, Sn (0.001); Si, Te, V (0.01); Se (trace)		Single crystal (111) direction
MB	Moore, McElroy, and Barrisoni (1965)	ORNL Metals and Ceramics Division		RRR = 400, Density = 19.29 g/cm ³ uncertainty = 2%
MB	Moore, McElroy, and Barrisoni (1965)	W (High purity)		RRR = 31.4, Density = 19.08
WB	White and Woods (1927)	Johnson - Matthey, Ltd.		RRR = 181, annealed at 1350°C
WB	White and Woods (1927)	W (99.99), Mo (0.01)		(111) direction, RRR = 9400
WB	Wagner, Garland, and Boers (1971)	W (Zone Refined)		(110) direction, RRR = 30,000
WB	Wagner, Garland, and Boers (1971)	W (Zone Refined)		(110) direction, RRR = 133,000
WB	Wagner, Garland, and Boers (1971)	W (Zone Refined)		(110) direction, RRR = 59,000
WB	Wagner, Garland, and Boers (1971)	W (Zone Refined)		(110) direction, RRR = 75,000
WB	Wagner, Garland, and Boers (1971)	W (Zone Refined)		(110) direction, RRR = 95,000

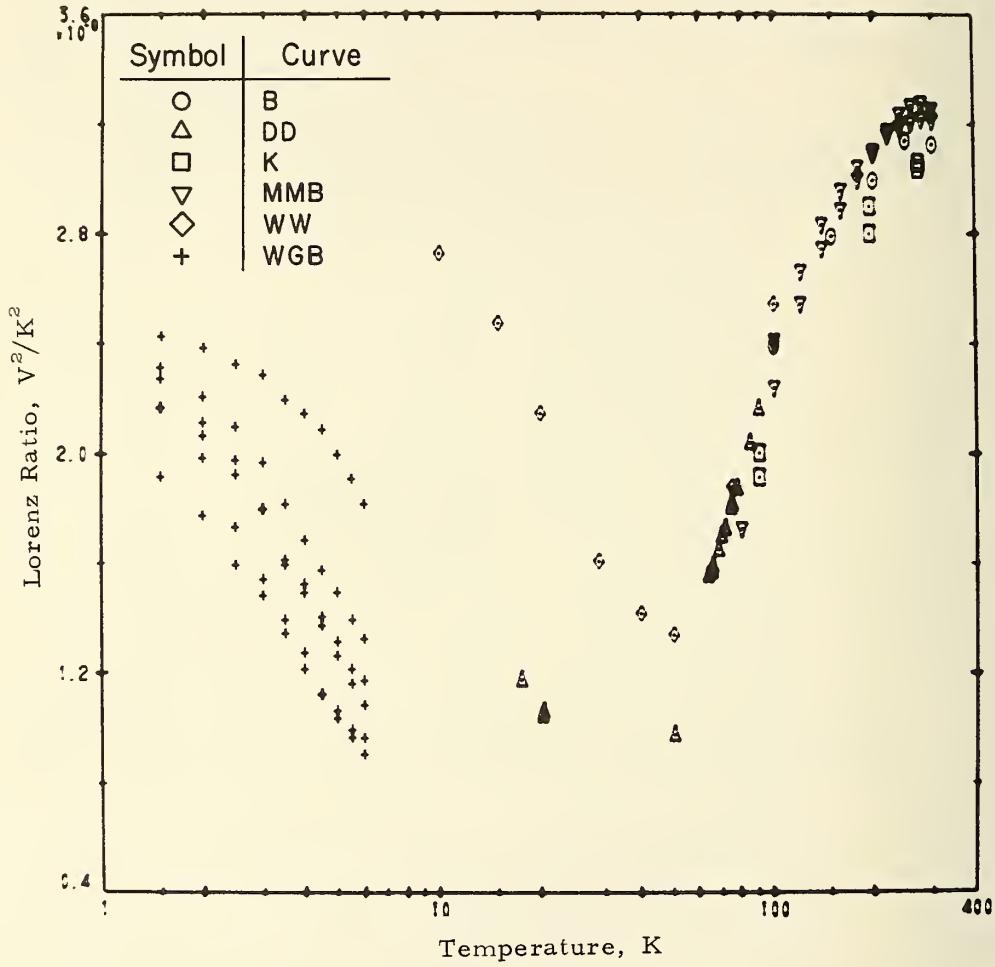


Figure 14. Lorenz ratio of tungsten

Molybdenum

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
B	Bachlund (1967)	Johnson - Matthey, Ltd. Mo (spectroscopically standardized)	$R_{HR} = 3.3$, uncertainty = 1.5%
K (1)	Kammlerik (1933)	Mo (99.8); Al, Ge, W, V, Ti, Sn (0.01); Bi, Ca (0.05); C (trace); Co, Cu, Pt, Rh (0.001)	
K (2)	Kammlerik (1933)	Mo (99.8); Al, Ge, W, V, Ti, Sn (0.01); Bi, Ca (0.05); C (trace); Co, Cu, Pt, Rh (0.001)	Annealed at 220°C $L = 2.6 \times 10^{-8} V^2/K^2$ at 273 K
	Kammlerik (1933)	N.V. Philips (Holland) Mo (very pure)	Annealed at 90°C $L = 2.67 \times 10^{-8} V^2/K^2$ at 273 K
	Kammlerik (1933)	N.V. Philips (Holland) Mo (very pure)	Annealed at 220°C $L = 2.67 \times 10^{-8} V^2/K^2$ at 273 K
	Kammlerik (1933)	General Electric Mo (less pure)	

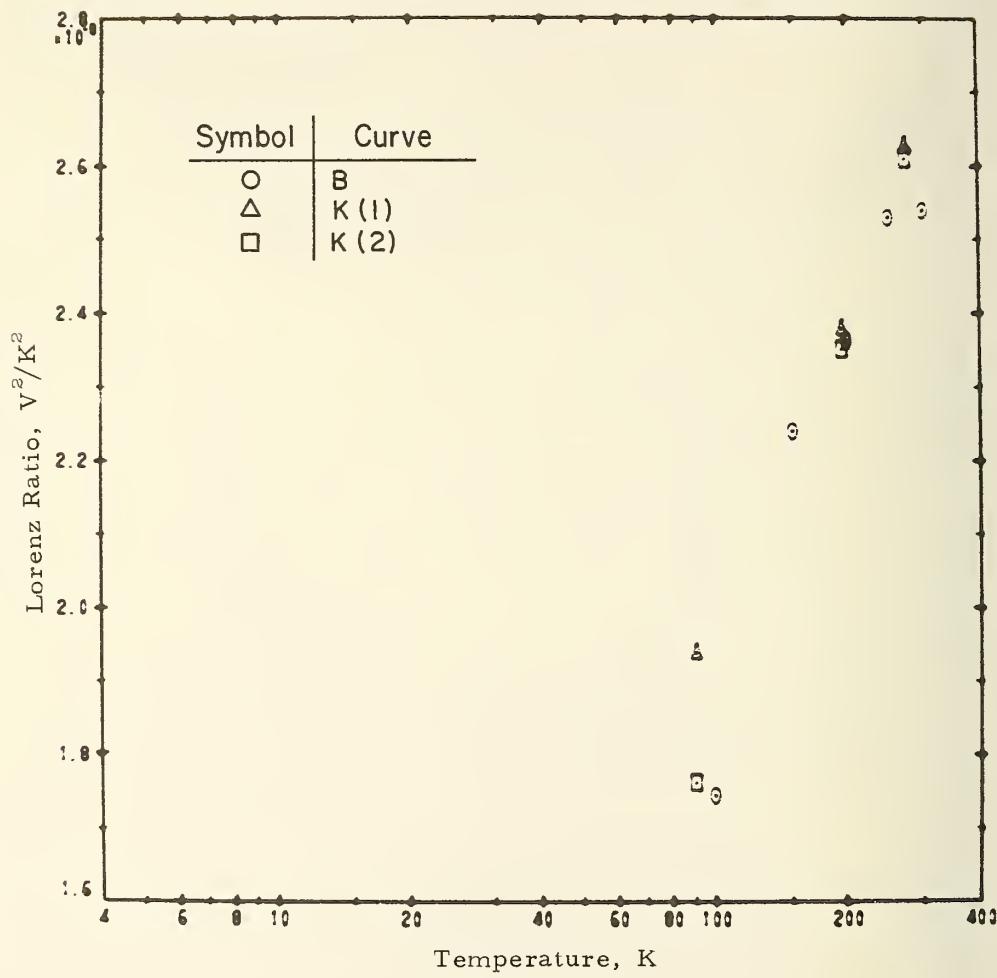


Figure 15. Lorenz ratio of molybdenum

Chromium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
G	Goff (1970)	Cr (99.92), Mn (0.004), Fe (0.005), Ni (0.002), Cu (0.003), Si, P, Ni	Arc Cast, 24 hours at 900°C RRR = 72
G	Goff (1970)	Cr (99.92), Mn (0.004), Fe (0.005), Ni (0.002), Cu (0.003), Si, P, Ni	twice annealed RRR = 88
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	Cold worked RRR = 47
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	4 hours at 1050°C RRR = 67
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	Partially recrystallized RRR = 97
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	4 hours at 1050°C RRR = 134
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	Fully recrystallized RRR = 220
HCKTW	Harper, Kemp, Klemens, Taitish, and White (1957)	Cr (99.998)	RRR = 70, uncertainty = 2%
HMM (53)	Moore, Williams, and McElroy (1963)	Dr. Goff's Cr specimen	
HMM (57)	Moore, Williams, and McElroy (1968)	Cr (99.92), Ni (0.004), Fe (0.005), Ni (0.002), Cu (0.003), Si, P, Ni	Extruded, RRR = 280, Density = 7.19 g/cm ³ , DP Hardness = 121,
HMM (57)	Moore, Williams, and McElroy (1968)	Ni (0.1ppm), V (0.3ppm), Ca (0.1ppm), Cu (0.1ppm), Fe (1.0ppm), Mg (0.3ppm), Si (1.0ppm), O (5ppm), H (0.2ppm)	grain size = 0.053 mm, uncertainty = 2%
HMM (57)	Moore, Williams, and McElroy (1969)	Cr (99.92), Ag (0.1ppm), Ca (0.1ppm), Cu (0.1ppm), Fe (1.0ppm), Mg (0.3ppm), Mn (0.1ppm), V (0.3ppm), Si (1.0ppm), O (5ppm), H (0.2ppm)	grain size = 0.38 mm, uncertainty = 2%

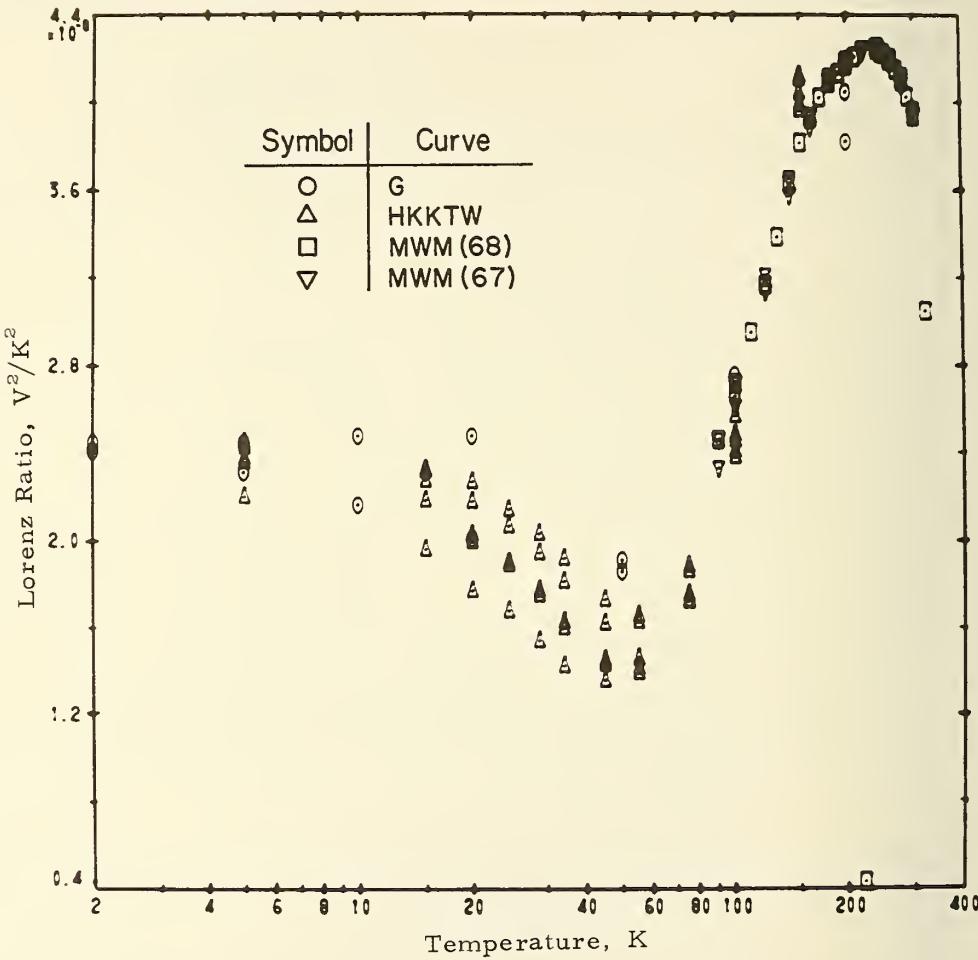


Figure 16. Lorenz ratio of chromium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		Mn	Re	
R	Reidemann (1935)	$\beta = 1\mu$		
W ₁ (Mn)	White and Woods (1957e)	A.D. MacKay Inc.		RRR = 8.33; 10 hours at 600°C Data read from small graph
W ₂ (Re 1)	White and Woods (1957e)	α - Mn (99.9), Mg (10 pp)		RRR = 20, unannealed, Density = 21.3 g/cm ³
W ₃ (Re 2)	White and Woods (1957e)	A.D. MacKay Inc.		annealed at 700°C RRR = 36
W ₄ (Re 3)	White and Woods (1957e)	Re (99.5)		Zone melted RRR = 1357

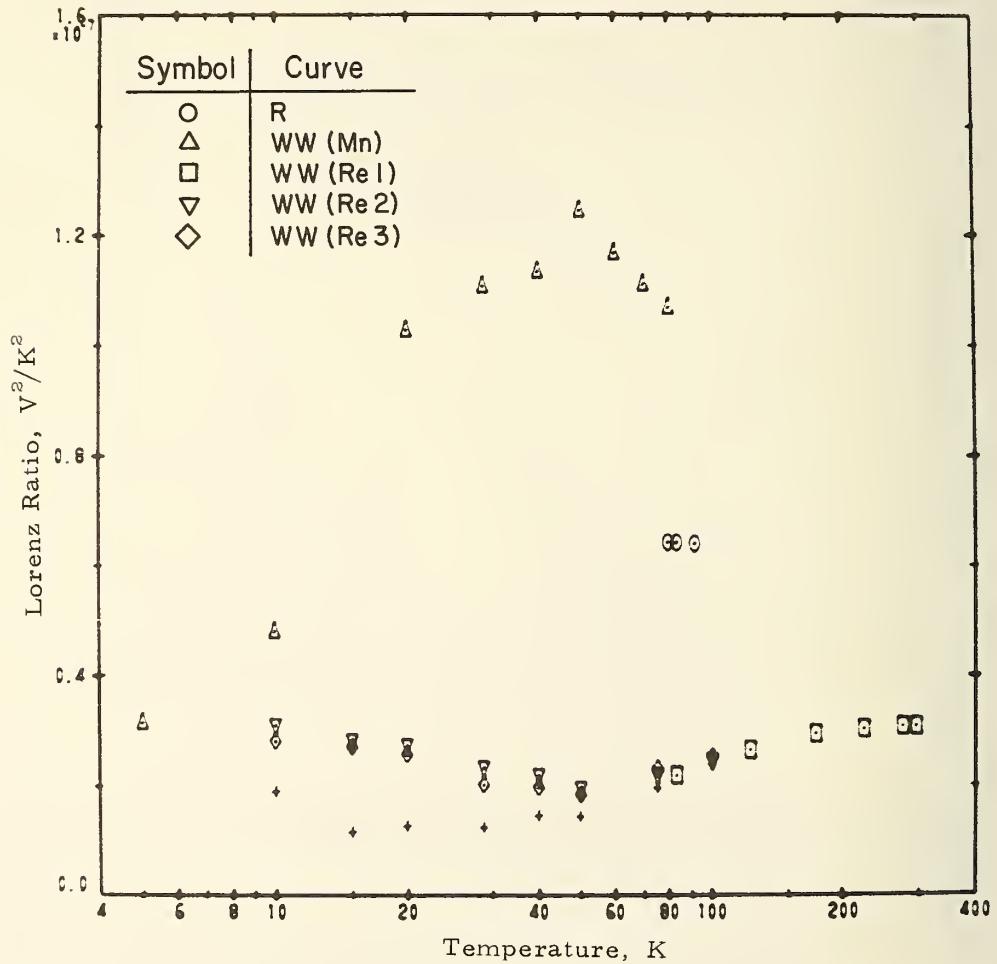


Figure 17. Lorenz ratio of manganese and rhenium

Iron

		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS		
CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		
B	Bucklund (1961)	Heraeus A.G. Fe, Mn (0.26)		10 hours at 500°C RRR = 4.6 Data read from a small graph
B	Bucklund (1961)	Fe, Ni (0.9)		RRR = 4.8
B	Bucklund (1961)	Fe, Mn (0.53)		RRR = 3.5
B	Bucklund (1961)	Fe, Ni (1.31)		RRR = 3.6
B	Bucklund (1961)	Fe, Si (1.17)		RRR = 2.4
D	Bucklund (1961)	Fe, Si (2.89)		RRR = 1.5
B	Bucklund (1961)	Phillips Research Laboratory Fe (Pure)		20 hour at 500°C RRR = 82
Eucken and Dittrich (1927)		Fe (Electrolytic)		$L = 1.99 \times 10^{-2} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 80 K $L = 3.32 \times 10^{-3} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 273 K
Eucken and Dittrich (1927)		Specimen 1		$L = 1.89 \times 10^{-2} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 80 K $L = 3.39 \times 10^{-3} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 273 K
Eucken and Dittrich (1927)		Specimen 2		$L = 2.39 \times 10^{-2} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 80 K $L = 3.20 \times 10^{-3} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 273 K
Eucken and Dittrich (1927)		Specimen 3		RRR = 700 to 2000
Beitshman, Trissel, and Coleman (1970)		Fe, Single crystal		$L = 1.80 \times 10^{-5} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 21.2 K $L = 1.41 \times 10^{-6} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 83.2 K
Grundisen and Goens (1927)		Fe (tempered)		$L = 2.04 \times 10^{-5} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 21.2 K $L = 1.52 \times 10^{-6} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 83.2 K
Grundisen and Goens (1927)		Specimen 1		$L = 2.47 \times 10^{-5} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 21.2 K $L = 2.09 \times 10^{-6} \frac{V^2}{A^2} \frac{K^2}{m^2}$ at 83.2 K
Grundisen and Goens (1927)		Fe (technically pure)		As received, machined
Hust, Powell, and Weitzer (1970)		Fe (Armco)		RRR = 13.8, Rockwell hardness = Brinell, grain size = 0.053 mm, uncertainty = 2.5%
HR (2)	also	Fe (99.9), Cr(0.015), Mn(0.028), P(0.005), Si(0.025), Si(0.003), Cu(0.04)		
	Hust (1969) and (1969a)			

Iron (Cont.)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		SOURCE	COMPOSITION	
HFA (2a)	Hust, Powell, and Weitzel (1970) also	Battelle Memorial Institute Fe (Armco)	Si(0.028), Mn(0.028), Ni(0.007), S(0.025), Si(0.003), Cu(0.04)	1½ hours at 875°C, RRR = 12.57, Rockwell hardness = 331, grain size = 0.65mm., uncertainty = 2.5%
HFA (4)	Hust, Powell, and Weitzel (1970) also	Battelle Memorial Institute Fe (Armco)	Fe(99.9), Cr(0.015), Mn(0.028), Ni(0.007), S(0.025), Si(0.003), Cu(0.04)	As received, machined, RRR = 13.4, Rockwell hardness = 310, grain size = 0.05 mm., uncertainty = 2.5%
HS	Hust, Powell, and Weitzel (1970a) (1971?) Hust, and Sparks (1970a), (1971?)	NBS - CRM Fe (Electrolyte)	Fe(99.9), Cr(0.015), Mn(0.028), Ni(0.007), S(0.025), Si(0.003), Cu(0.04)	2 hours at 1000°C, RRR = 21.1, Density = 7.87 g/cm³, Rockwell hardness = 324, grain size = 0.05 mm., uncertainty = 2.5%
K	Kammlauf (1933)	Fe(Armco)	Fe(99.9), Cr(0.015), Mn(0.005), Ni(0.005), S(0.003), Si(0.025), Al(0.002)	RRR = 29
KG	Karweil and Schaefer (1939)	Fe	Fe(99.9), Cr(0.011), Ni(0.017), P(0.006), Si(0.026), Cu(0.059), Si(0.002)	
KS	Karweil and Schaefer (1939)	Fe, Si(0.35), S(0.03), Cr(0.1), Ni(0.5)		
KW	Kemp, Klemens, and White (1956)	Johnson - Matthey Company	Fe(99.99), Ni(0.005), Cu(0.0002), Ag(0.0001)	4 hours at 750°C, RRR = 40 Data read from small graph
KCT	Kemp, Klemens and Tannah (1959)	Fe (Doubly refined electrolytic)	Fe(99.99), Ni(0.005), Cu(0.0002), Ag(0.0001)	950°C anneal, RRR = 100 Data read from small graph
KK	Kolbhas and Kiersepe (1965)	Fe (Pure)	Fe(99.9%), Cr(0.1), Ni(0.15), Si(0.15)	Density = 7.71 g/cm³ at 21°C
L	Lees (1968)	Materials Research Corporation	Moore, McElroy, and Baroni (1966)	RRR = 201, Density = 7.32 g/cm³, uncertainty = 2%
LD			Fe (electron beam zone refined)	

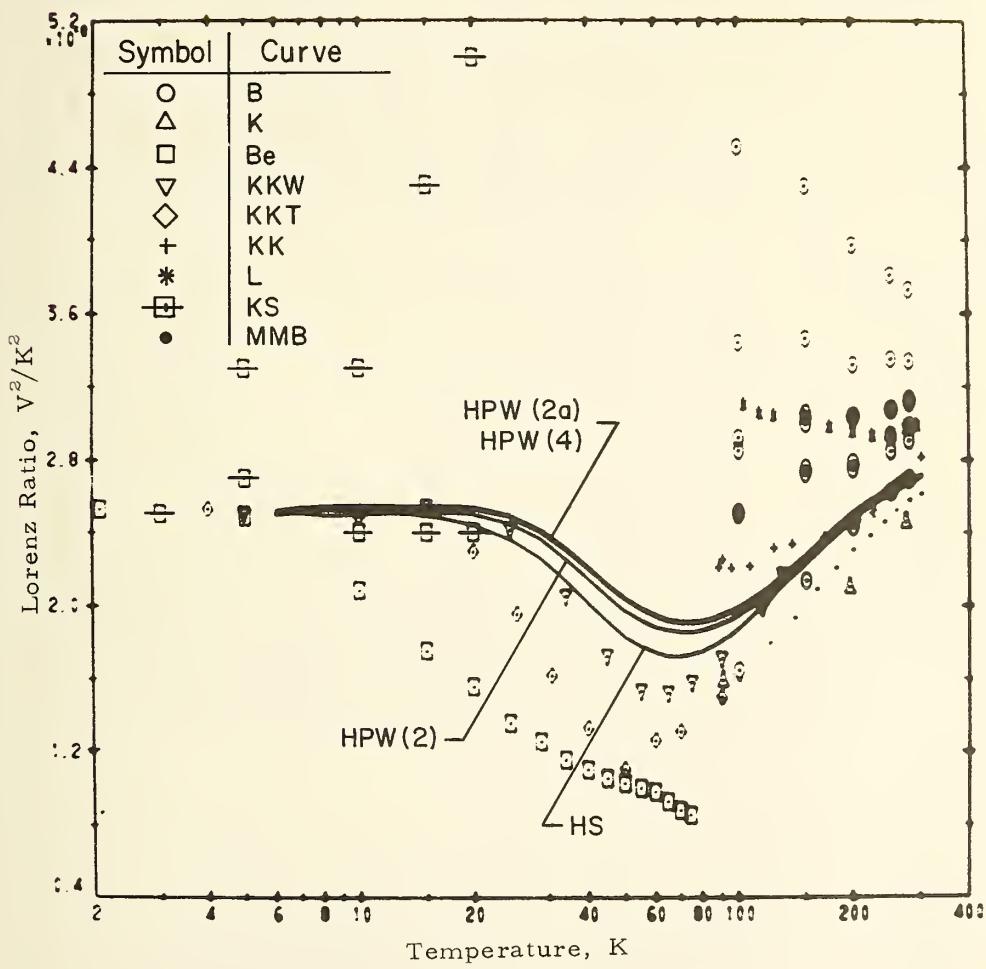


Figure 18. Lorenz ratio of iron

Stainless and Alloy Steels

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		SOURCE	COMPOSITION (%)	
B	Berwin (1951)	Austenitic Stainless Steel.	Fe(71.4), Ni(7.2), Cr(12.2), Si(0.7), Ti(1), C(0.1)	$\sigma_{ymin} = 1400$ MPa
EZ (303)	Esterman and Zimmerman (1952)	AISI 303 Stainless Steel		
EZ (347)	Esterman and Zimmerman (1952)	AISI 347 Stainless Steel		
F (1015)	Flynn (1971)	AISI 1015 steel	DP Hardness = E97	
F (M4)	Flynn (1971)	Fe (97.34), Ni (0.99)	DP Hardness = E72	
F (2315)	Flynn (1971)	AISI 2315 Steel	DP Hardness = E94	
F (1310)	Flynn (1971)	AISI 1310 (37) Steel	DP Hardness = C41	
F (2515)	Flynn (1971)	Fe(94.22), Ni(1.75)	DP Hardness = E99	
F (M4)	Flynn (1971)	AISI 2515 Steel	DP Hardness = E99	
F (NP 49)	Flynn (1971)	Fe (93.48), Ni (4.64)	DP Hardness = E100	
F (NB90)	Flynn (1971)	Fe (90.68), Ni (8.13)	DP Hardness = E95	
F (1322)	Flynn (1971)	High Permeability 49	DP Hardness = E99	
F (INVAR)	Flynn (1971)	Fe (50.4), Ni (47.6)	DP Hardness = E99	
F (rc1)	Flynn (1971)	High Mn 80	DP Hardness = E79	
F (NS2)	Flynn (1971)	Fe (16.12), Ni (79.55)	DP Hardness = E79	
		Low expansion 12	DP Hardness = E92	
		Fe (57.2), Ni(40.7)	DP Hardness = E92	
		INVAR	DP Hardness = E92	
		Fe (54.84), Ni (34.15)	DP Hardness = E92	
		Free Cut INVAR	DP Hardness = E92	
		Fe (62.96), Ni (34.5)	DP Hardness = E92	
		NI Span C	DP Hardness = C24	
		Fe (49.59), Ni (40.21)	DP Hardness = C24	

Stainless and Alloy Steels (Cont.)

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)		MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
		REMARKS		
HS (1)	Rust and Sparks (1971a) and (1972)	Austenitic stainless steel specimen 1 Fe (62.1), Mn (19.9), Cr(16.41), Mn(1.20), Si(0.27), Ni(0.10), Mo(0.01), C(0.009), N (0.009), P(0.005), S(0.006)	RR = 1.33, Density = 8.000 g/cm ³ , Rockwell hardness = BHN, grain hardness = 0.08 mm, uncertainty = 2.5%	
HS (4R)	Rust and Sparks (1971a) test (1972)	Austenitic stainless steel specimen WH Fe(62.1), Mn(19.9), Cr(16.41), Mn(1.20), Si(0.27), Ni(0.10), Mo(0.01), C(0.009), N(0.009), P(0.005), S(0.006)	RR = 1.33, Density = 8.000 g/cm ³ , Rockwell hardness = BHN, grain hardness = 0.08 mm, uncertainty = 2.5%	
HS (2)	Rust and Sparks (1971a) and (1972)	Austenitic stainless steel specimen Lot 2 Fe(62.1), Mn(19.9), Cr(16.41), Mn(1.20), Si(0.27), Ni(0.10), Mo(0.01), C(0.009), N(0.009), P(0.005), S(0.006)	RR = 1.33, Density = 8.000 g/cm ³ , Rockwell hardness = BHN, grain hardness = 0.08 mm, uncertainty = 2.5%	
RS (266)	Rust and Sparks (1971b)	AISI A286 Steel Fe(64.0), Mn(25.4), Cr(14.8), Mn(1.4), Mo(1.2), Si(0.5), V(0.3), Al(0.2), C(0.01), P(0.01), S(0.01)	As received, RRR = 1.202, Density = 7.917 g/cm ³ , Rockwell hardness = 039, grain size = 0.036 mm, uncertainty = 2.5%	
HS (266A)	Rust and Sparks (1971b)	AISI A286 Steel Fe(64.0), Mn(25.4), Cr(14.8), Mn(1.4), Mo(1.2), Si(0.5), V(0.3), Al(0.2), C(0.01), P(0.01), S(0.01)	1 hour at 932°C, RRR = 1.209, Density = 7.925, Rockwell hardness = C20, grain size = 0.015 mm, uncertainty = 2.5%	
RS (22)	Rust and Sparks (1971a)	Stainless Steel Fe(64.0), Cr(21.48), Mn(12.36), Mn(5.44), Mo(2.12), Si(0.42), Ni(0.27), V(0.20), Ni(0.19), C(0.05), Si(0.01)	grain size = 0.08 mm, Rockwell hardness = B52, uncertainty = 2.5%	
H (347)	Hunt (1970a)	AISI 347 steel Fe(64.0), Cr(17.16), Mn(11.52), Mn(1.34), Mn + Ta(1.11), Si(0.59), C(0.027), P(0.014), S(0.007)	Triple Furnace treated RRR = 1.18, Rockwell Hardness = B77, grain size = 0.0359 mm, uncertainty = 2.5%	
KK (1)	Kohlheas and Kierspe (1965)	Fe, Mn(10.85), Cr(16.83), Mo(19.4), Si(0.37), Mn		
KK (1)	Kohlheas and Kierspe (1965)	Fe, Mn(11.5), Cr(18.55), Mn(1.3), Si(0.7)		
KK (1)	Kohlheas and Kierspe (1965)	Fe, Mn(12.76), Cr(17.16), Mn(1.37), Si(0.5)		

Stainless and Alloy Steels (Cont.)

CURVE	INVESTIGATOR(S) [YEAR]	SPECIMEN SOURCE AND COMPOSITION [%]	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS			
KK (1)	Kohlhaas and Kierspe (1965)	Fe, Ni(15.8), Cr(15.6), Mn(19.3), Ni(1.3), Si				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.13), Si(0.24), Mn(0.16), P(0.021), S(0.027), N(0.005), Al(0.036), Cr(0.09)				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.13), Si(0.27), Mn(0.53), P(0.05), S(0.027), N(0.005), Al(0.027), Cr(0.09)				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.10), Si(0.32), Mn(0.83), P(0.016), S(0.031), N(0.009), Al(0.22), Cr(0.076)				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.09), Si(0.32), Mn(0.15)				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.085), Si(0.35), Mn(0.40)				
KK (2)	Kohlhaas and Kierspe (1965)	Fe, C(0.051), Si(0.28), Mn(0.74)				
KK (m)	Kohlhaas and Kierspe (1965)	Fe, Cr(3.55), Mn(21.75), C(0.324)				
TW	Tyler and Wilson (1952)	Allegheny Lallemand Corporation AISI 316 steel	25% cold reduction, Brinell hardness = 255, grain size = ASTM # 6			
	Tyler and Wilson (1952)	AISI 304 steel	grain size = ASTM #5			
	Tyler, Neabolt, and Wilson (1953)	Allegheny Lallemand Corporation AISI 316 steel	$I = 5 \times 10^{-8} V^2/K$ at 23.17 K $I = 5.4 \times 10^{-8} V^2/K^2$ at 76.77 K			
	Tyler, Neabolt, and Wilson (1953)	AISI 304 steel	grain size = ASTM #5			
	Tyler, Neabolt, and Wilson (1953)		$I = 5.97 \times 10^{-8} V^2/K^2$ at 30 K $I = 6.57 \times 10^{-8} V^2/K^2$ at 70 K			

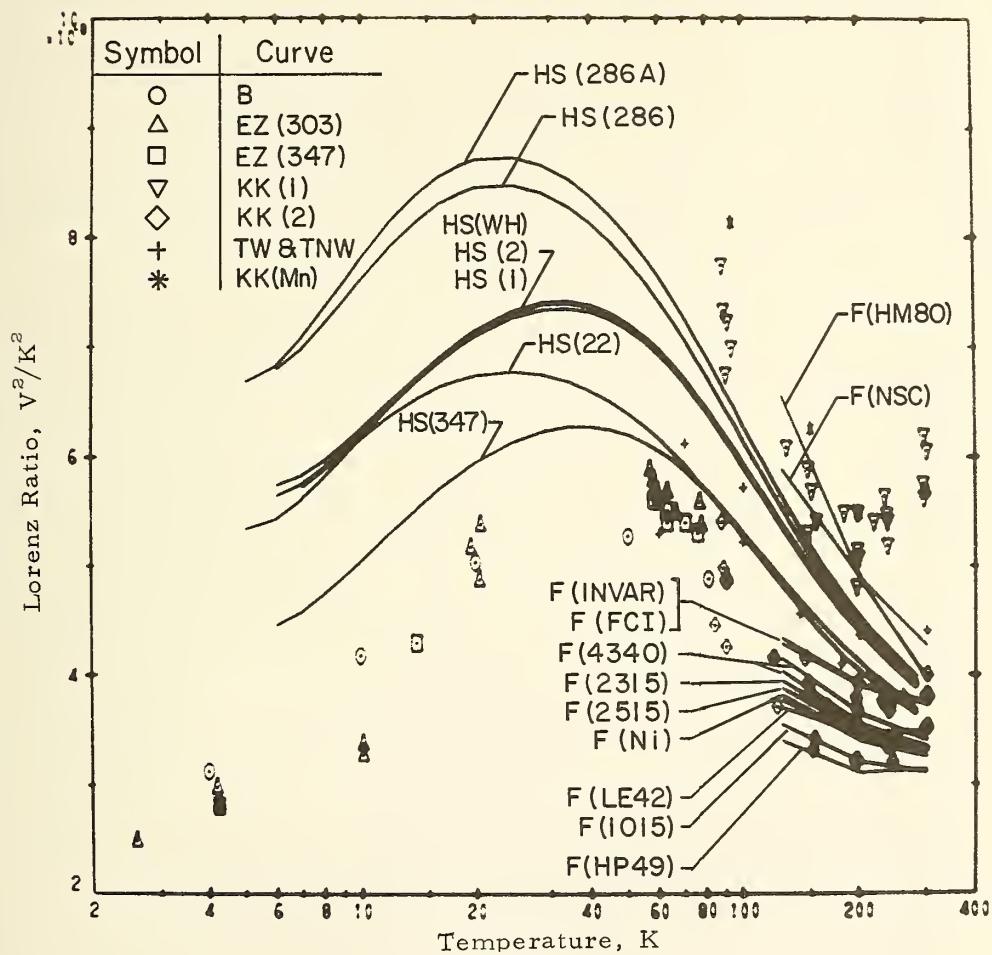


Figure 19. Lorenz ratio of stainless and alloy steels

Carbon Steels

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
			Density = 7.84 g/cm ³ at 24 °C
L	Lees (1908)	Fe (99), C (1)	

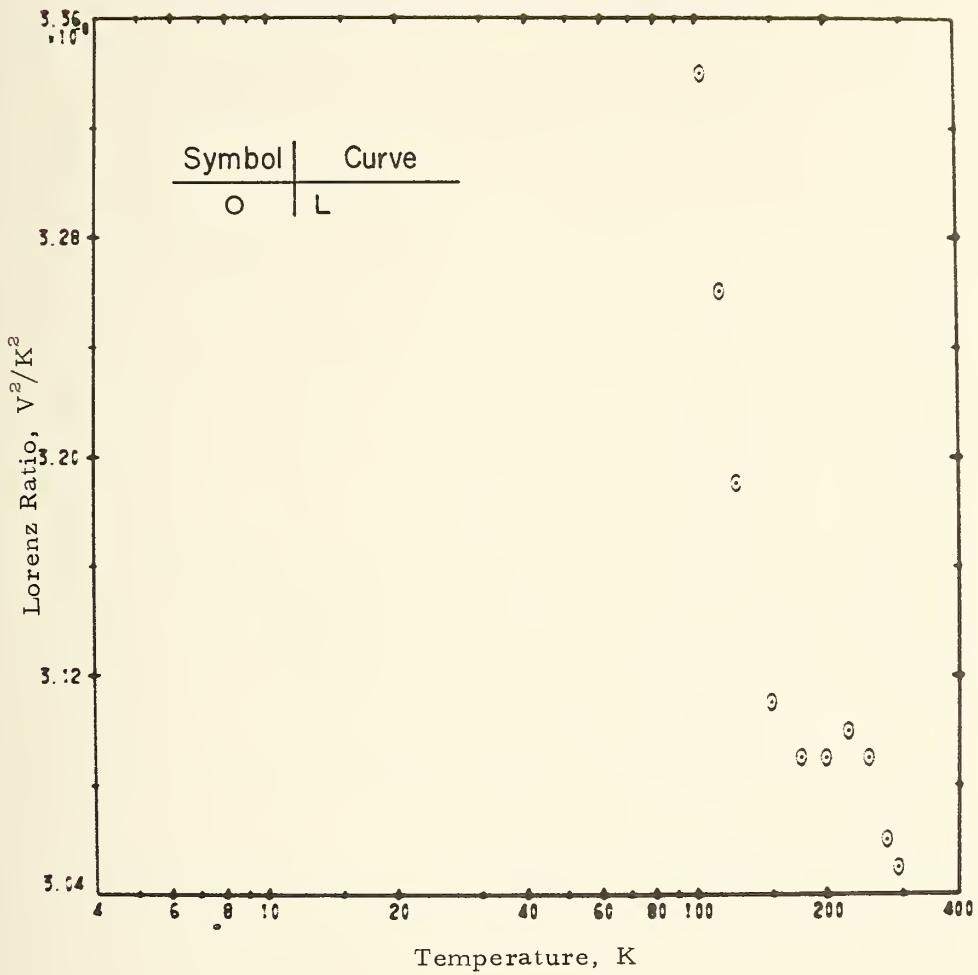


Figure 20. Lorenz ratio of carbon steels

Cobalt

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS
RI	Buddekrüger and Nielsen (1965)	Johnson - Matthey Company Co (pure)	3 hours at 100°C in vacuum RRR = 60
WW	White and Woods (1957b)	Johnson - Matthey Company Co (99.999), Si(0.0002), Fe(0.0005), Al(0.0001)	RRR = G ₁ , 2 hours at 700°C
WPD	Wilkes, Powell, and Dedit (1969)	Co(99.99%), Ni(0.01%), Cu(0.001), Pb(0.0003), Mn(0.0001), Pb(0.0001), Cu(0.0001), Si(0.0003), Al(0.0002), Mg(0.0002), Si(0.0008), C(0.004)	Density = 8.8 g/cm ³ at 20°C

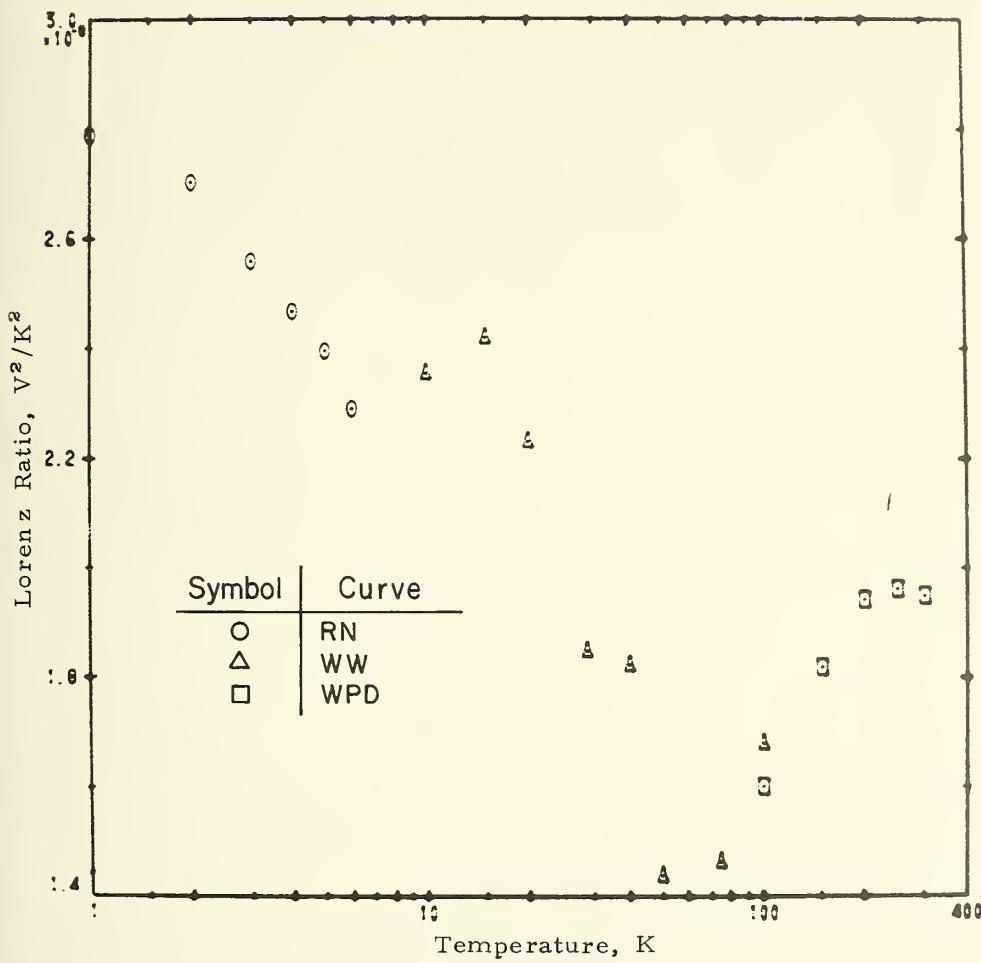


Figure 21. Lorenz ratio of cobalt

Nickel

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
			-B	-B
F	Acyorn (1940)	Ni (electrolytic)	$L = 1.41 \times 10^{-8} \text{ V/K}^2$ at 79.15 K	
G	Bucken and Dittrich (1927)	Ni (electrolytic)	$L = 1.54 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 80 K	
FG	Farrell and Credé (1959)	Ni	$L = 2.35 \times 10^{-8} \text{ V}^2/\text{K}^2$ at 273 K	
GH	Grieg and Harrison (1965)	Johnson - Matthey Company Ni, Fe + Si (16 ppm)	RHR = 650, uncertainty = 5%	
KGM	Kemp, Clemens, and White (1956)	Johnson - Matthey Company Ni (99.99)	12 hours at 950°C grain size = 2.1 nm Data read from small graph	
L	Lees (1968)	Johnson - Matthey Company Ni (99)	4 hours at 750°C RHR = 208	
WT	White and Polanski (1967)	Ni (high purity)	Density = 6.90 g/cm ³ at 21°C RHR = 2500 annealed at 500°C in vacuum	

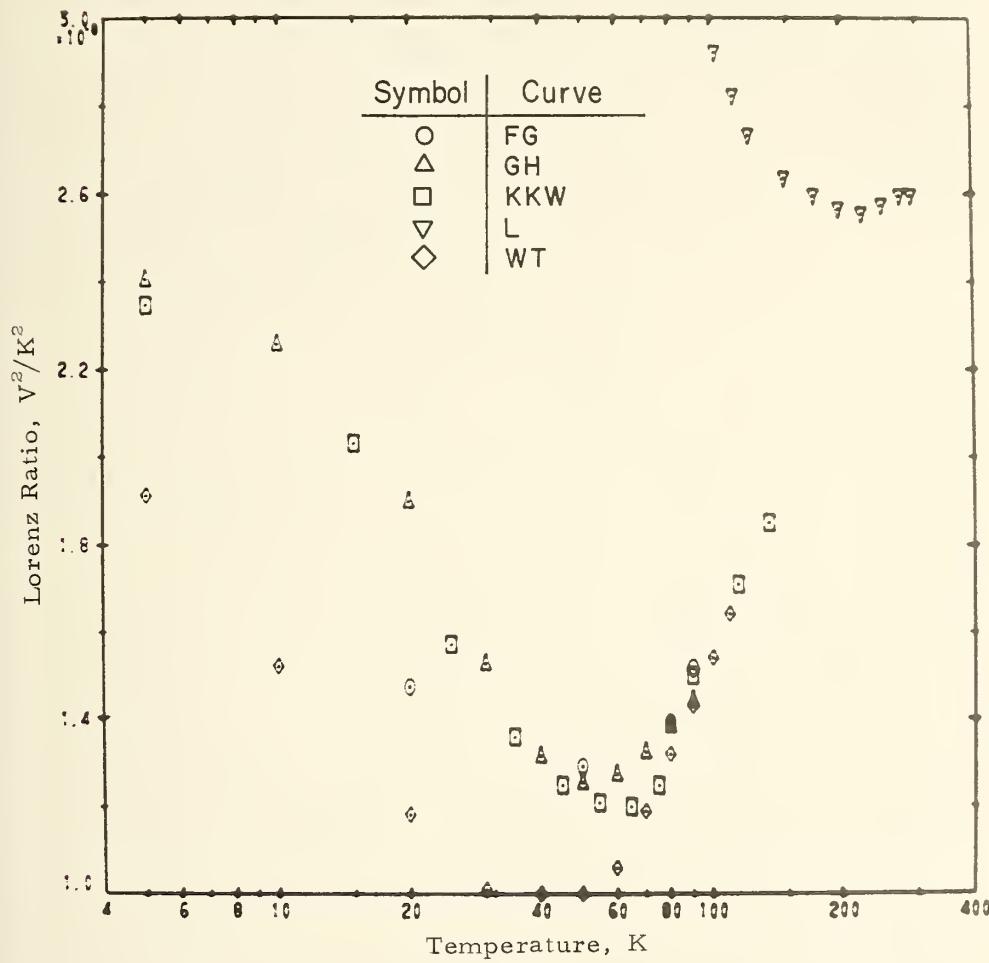


Figure 22. Lorenz ratio of nickel

Nickel Alloys

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS		
A	Aoyama (1940)	Moneal			
EZ (NPR)	Estermann and Zimmerman (1922)	Moneal (hard drawn tubing)			
EZ (NPR)	Estermann and Zimmerman (1922)	Moneal (annealed tubing)			
EZ (NPR)	Estermann and Zimmerman (1922)	Moneal (hard drawn rod)			
EZ (IDF)	Estermann and Zimmerman (1922)	Inconel (hard drawn tubing)			
EZ (JARV)	Estermann and Zimmerman (1922)	Inconel (annealed tubing)			
HPW (X)	Rust, Powell, and Weitzel (1969) and Rust, Weitzel, and Powell (1971)	Hastelloy X Ni(49), Cr(21), Fe(17.53), Mo(9.15), Co(1.49), W(0.65), Mn(0.53), Si(0.13), Cr(0.12), P, S	Rockwell hardness = 33B grain size = 0.08 mm		
HPW (7B)	Rust, Powell, and Weitzel (1969) and Rust, Weitzel, and Powell (1971)	Inconel 718 (age hardened) Ni(54.57), Cr(18.06), Fe(17.08), Mo(5.12), Mo(3.16), Ti(0.85), Al(0.44), Mn(0.29), Si(0.24), Cu, C, S	Rockwell hardness = C39, Density = 8.261 g/cm ³ at 20°C, RRR = 1.06, grain size = 0.06 uncertainty = ± 5%		
HPW (7B)	Rust, and Sparks (1970)	Inconel 718 full anneal condition Ni(54.57), Cr(18.06), Fe(17.08), Mo(5.12), Mo(3.16), Ti(0.85), Al(0.44), Mn(0.29), Si(0.24), Cu, C, S	anneal at 1035°C for 1 hour RRR = 1.32, Hardness = 895, Density = 8.261 g/cm ³ at 20°C, Grain size = 0.01 mm		

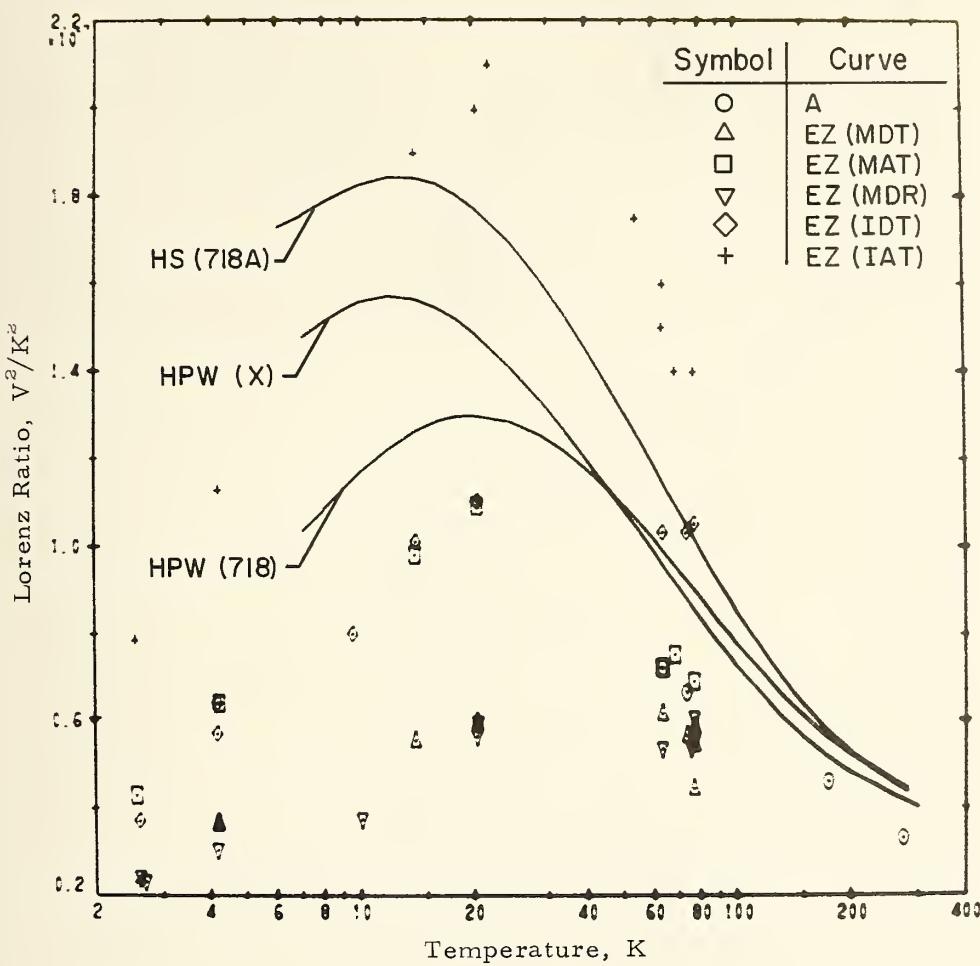


Figure 23. Lorenz ratio of nickel alloys

Platinum

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS	
PR4	Powell, Tyre and Goodman (1957)	Johnson - Matthey Company Pt., (Ag, Fe, Ni, Pt, and Cu each less than 1 ppm)	Annealed at 1273 K, RRR = 740, Density = 21.5 g/cm ³ , uncertainty = 7%	
PR4	Powell, Tyre and Goodman (1957)	Johnson - Matthey Company Pt., (Ag, Fe, Ni, Pd, and Cu each less than 1 ppm)	Annealed at 1250 K, density = 21.5 g/cm ³ $L = 2.66 \times 10^{-3} \frac{\Omega^2}{K^2}$ at 300 K	
WW	White and Woods (1957)	Baker Platinum Company Pt (99.99)	Annealed $L = 1.13 \times 10^{-3} \frac{\Omega^2}{K^2}$ at 21.1 K $L = 83.2 \times 10^{-3} \frac{\Omega^2}{K^2}$ at 83.2 K Data read from small graph	
WW	Gruneisen and Goens (1927)	Herenus Pt (very pure)	Annealed $L = 1.25 \times 10^{-3} \frac{\Omega^2}{K^2}$ at 21.2 K	
WWB (1)	Moore, McElroy, and Berisoli (1966)	Pt (99.999) Powell, Tyre and Goodman's (1957) specimen	RRR = 700, Density = 21.5 g/cm ³ , uncertainty = 2%	
WWB (2)	Moore, McElroy, and Berisoli (1966)	Pt (99.98)	RRR = 425	
			Specimen supplied by D.R. Flynn of NBS, Washington	

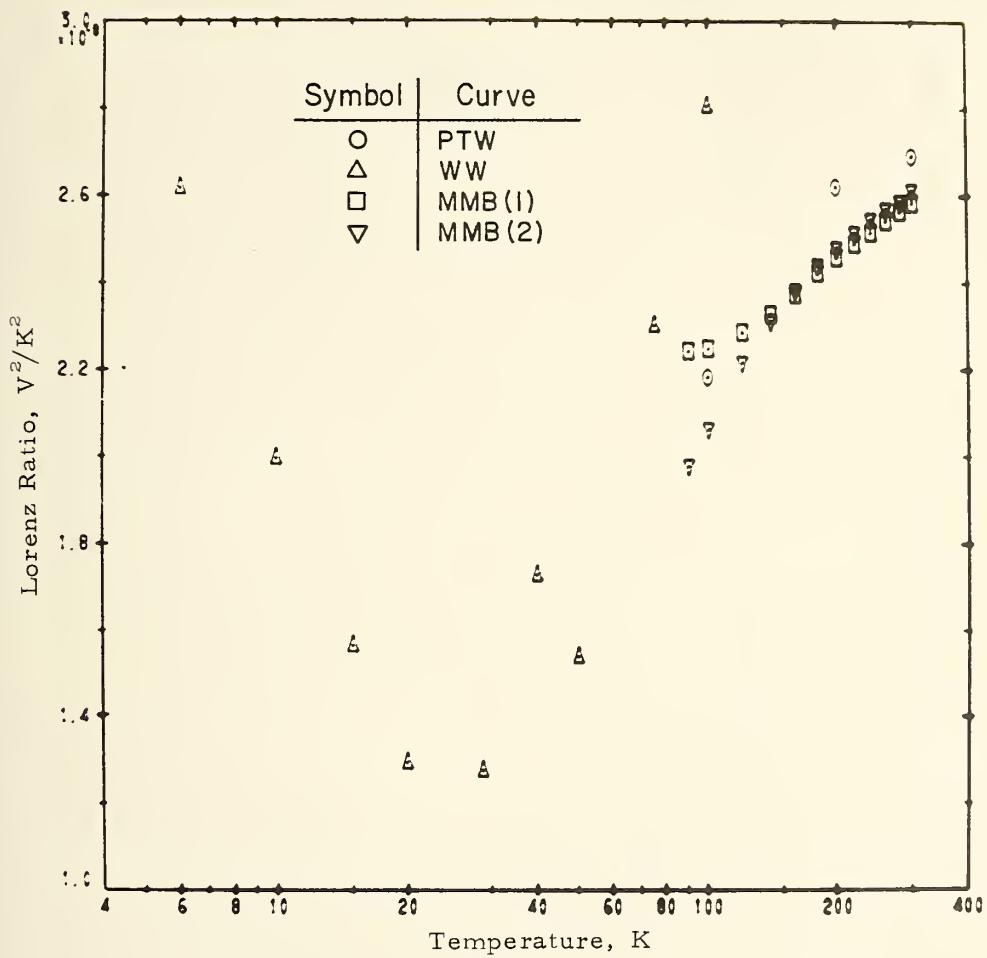


Figure 24. Lorenz ratio of platinum

Palladium

CURVE	INVESTIGATOR(S) (YEAR)	SPECIMEN SOURCE AND COMPOSITION (%)	MISCELLANEOUS SPECIMEN CHARACTERIZATION - REMARKS		
			RBB	RRR	Other
	Powell, Tye, and Woodman (1957)	Johnson - Matthey Company Pd, Fe (5 ppm), Au (5 ppm), Ag (<1 ppm), Cu (1 ppm), Pt (50 ppm), Pt (2 ppm)	RBB = 69, As received Density = 12.02 g/cm ³ , uncertainty = 7% $L = 2.73 \times 10^{-5} \varphi^2/k^2$ at 300 K		
KCSM	Lemp, Clemens, Sreechar, and White (1955)	Johnson - Matthey Company Pd (99.99), Cu, Sn, Mg, Ag	annealed at 250°C to 1000°C for four hours. Specimen was measured in six different physical conditions	500	

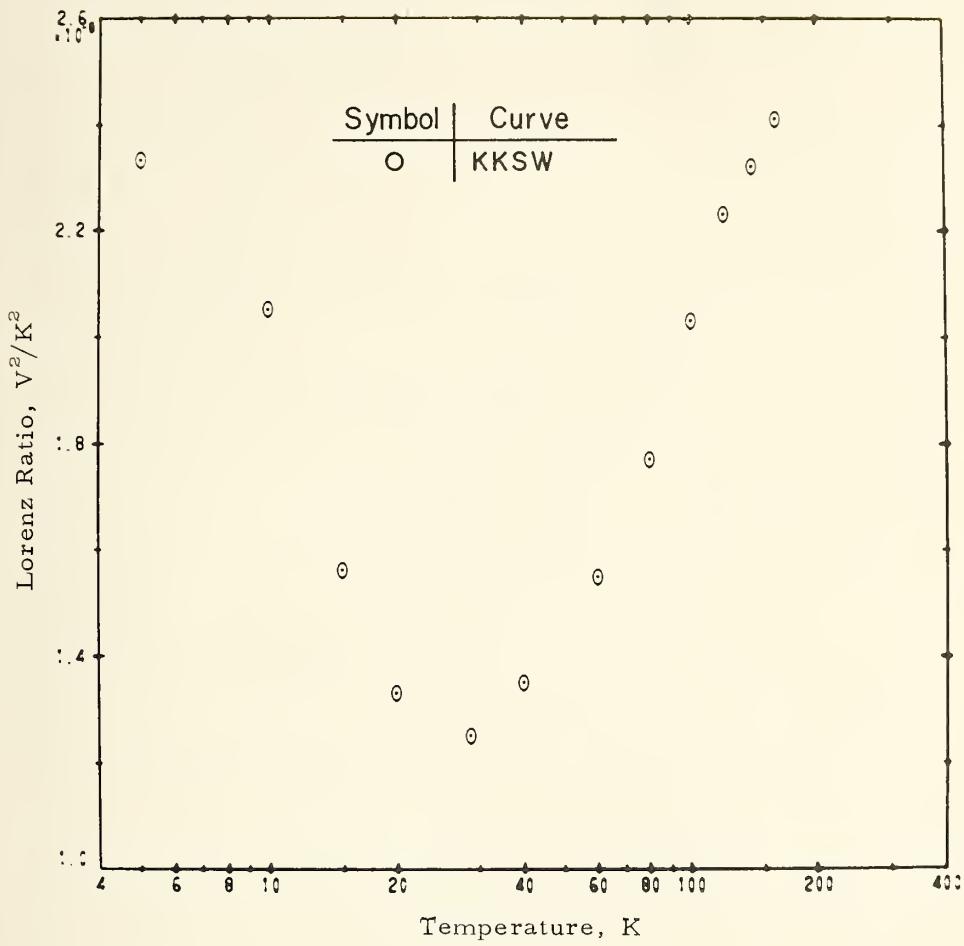


Figure 25. Lorenz ratio of palladium



	Page
Beryllium	II- 2
Magnesium	II- 3
Aluminum	II- 4
Aluminum Alloys	II- 6
Lead and Tin	II- 8
Gold	II- 9
Gold Cobalt Alloy	II-11
Silver	II-12
Copper	II-14
Copper Alloys (German Silver and Brass)	II-15
Zinc and Cadmium	II-16
Scandium and Yttrium	II-17
Titanium, Hafnium, and Zirconium	II-18
Titanium Alloys	II-19
Tungsten	II-20
Molybdenum	II-22
Chromium	II-23
Manganese and Rhenium	II-25
Iron	II-26
Stainless and Alloy Steels	II-30
Carbon Steels	II-37
Cobalt	II-37
Nickel	II-38
Nickel Alloys	II-39
Platinum	II-42
Palladium	II-42

Beryllium

HUST, WEITZEL, AND POWELL (1971), BE-HWP

LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
70.0	3.00-008	70.0	3.00-008	4.2	2.12-008	4.2	2.12-008
75.0	3.01-008	75.0	3.01-008	10.0	2.30-008	10.0	2.30-008
80.0	3.01-008	80.0	3.01-008	20.0	2.49-008	20.0	2.49-008
85.0	3.01-008	85.0	3.01-008	30.0	2.62-008	30.0	2.62-008
90.0	3.01-008	90.0	3.01-008	40.0	2.70-008	40.0	2.70-008
95.0	3.01-008	95.0	3.01-008	50.0	2.73-008	50.0	2.73-008
100.0	3.00-008	100.0	3.00-008	60.0	2.78-008	60.0	2.78-008
110.0	2.99-008	110.0	2.99-008	70.0	2.78-008	70.0	2.78-008
120.0	2.97-008	120.0	2.97-008	80.0	2.80-008	80.0	2.80-008
130.0	2.97-008	130.0	2.97-008	90.0	2.80-008	90.0	2.80-008
140.0	2.97-008	140.0	2.97-008	100.0	2.80-008	100.0	2.80-008
150.0	2.97-008	150.0	2.97-008				
160.0	2.98-008	160.0	2.98-008				
170.0	3.00-008	170.0	3.00-008				
180.0	3.02-008	180.0	3.02-008				
190.0	3.04-008	190.0	3.04-008				
200.0	3.06-008	200.0	3.06-008				
220.0	3.10-008	220.0	3.10-008				
240.0	3.15-008	240.0	3.15-008				
260.0	3.19-008	260.0	3.19-008				
280.0	3.22-008	280.0	3.22-008				

LEWIS(1929), BE-L

LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
103.0	1.72-008	103.0	1.72-008	123.0	1.72-008	123.0	1.72-008
123.0	1.72-008	143.0	1.82-008	143.0	1.82-008	163.0	1.98-008
143.0	1.82-008	183.0	2.18-008	183.0	2.18-008	203.0	2.40-008
183.0	2.18-008	203.0	2.40-008	203.0	2.40-008	223.0	2.67-008
203.0	2.40-008	223.0	2.67-008	223.0	2.67-008	243.0	2.98-008
223.0	2.67-008	243.0	2.98-008	243.0	2.98-008	263.0	3.28-008
243.0	2.98-008	263.0	3.28-008	263.0	3.28-008	273.0	3.46-008
263.0	3.28-008	273.0	3.46-008	273.0	3.46-008	293.0	3.77-008

POWELL, HARDEN, AND GIBSON(1960), BE-PHG(PARALLEL)

LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
4.02	2.01-008	4.2	2.01-008	4.2	2.01-008	4.2	2.01-008
10.0	2.32-008	10.0	2.32-008	10.0	2.32-008	10.0	2.32-008
20.0	2.45-008	20.0	2.45-008	20.0	2.45-008	20.0	2.45-008
30.0	2.45-008	30.0	2.45-008	30.0	2.45-008	30.0	2.45-008
40.0	2.50-008	40.0	2.50-008	40.0	2.50-008	40.0	2.50-008
50.0	2.53-008	50.0	2.53-008	50.0	2.53-008	50.0	2.53-008
60.0	2.52-008	60.0	2.52-008	60.0	2.52-008	60.0	2.52-008
70.0	2.50-008	70.0	2.50-008	70.0	2.50-008	70.0	2.50-008
80.0	2.45-008	80.0	2.45-008	80.0	2.45-008	80.0	2.45-008
90.0	2.44-008	90.0	2.44-008	90.0	2.44-008	90.0	2.44-008
100.0	2.44-008						

WHITE AND WOODS(1955), BE-WW(1)	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
LORENZ RATIO	TEMP RATIO	TEMP RATIO
10.0	3.30-008	1.10-006
20.0	2.75-008	1.10-006
30.0	3.30-008	1.10-006
40.0	3.05-008	1.11-006
50.0	3.00-008	1.11-006
60.0	3.00-008	1.11-006
70.0	3.00-008	1.11-006
80.0	3.00-008	1.11-006
90.0	3.00-008	1.11-006
100.0	3.00-008	1.11-006

WHITE AND WOODS(1955), BE-WW(2)	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
LORENZ RATIO	TEMP RATIO	TEMP RATIO
10.0	3.05-008	1.15-006
20.0	2.88-008	1.15-006
30.0	3.45-008	1.15-006
40.0	3.16-008	1.16-006
50.0	3.16-008	1.16-006
60.0	3.20-008	1.24-006
70.0	3.28-008	1.28-006
80.0	3.31-006	1.28-006
90.0	3.35-006	1.28-006
100.0	3.40-006	1.28-006

MAGNESIUM

SEE APPENDIX I - MAGNESIUM

Aluminum

AMUNOSEN, MYHRE, AND SALTER (1972), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.44-008	TEMP	2.0	2.44-008	TEMP	2.3
	2.0	2.41-008		3.0	2.41-008		2.47-008
	3.0	2.41-008		3.0	2.41-008		2.36-008
	4.0	2.37-008		4.0	2.37-008		1.77-008

AMUNOSEN, MYHRE, AND SALTER (1972), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.42-008	TEMP	2.0	2.42-008	TEMP	2.0
	3.0	2.38-008		3.0	2.38-008		1.50-008
	4.0	2.30-008		4.0	2.30-008		1.54-008

ANDREWS, WEBBER, AND SPOHR (1951), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.44-008	TEMP	2.0	2.44-008	TEMP	2.0
	3.0	2.44-008		3.0	2.44-008		1.90-008
	4.0	2.43-008		4.0	2.43-008		1.98-008

ANDREWS, WEBBER, AND SPOHR (1951), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	4.2	2.34-008	TEMP	4.2	2.34-008	TEMP	4.2
	14.5	1.80-008		14.5	1.80-008		1.66-008
	20.4	1.40-008		20.4	1.40-008		1.43-008

FENTON, ROGERS, AND WOODS (1963), AL-FRW

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	4.2	2.41-008	TEMP	4.2	2.41-008	TEMP	4.2
	14.5	1.88-008		14.5	1.88-008		1.55-008
	20.4	1.57-008		20.4	1.57-008		1.10-008

FENTON, ROGERS, AND WOODS (1963), AL-FRW

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.3	2.43-008	TEMP	2.3	2.43-008	TEMP	2.3
	5.0	2.23-008		5.0	2.23-008		2.01-008
	10.0	1.55-008		10.0	1.55-008		1.65-008
	15.0	1.10-008		15.0	1.10-008		1.31-008
	20.0	8.90-009		20.0	8.90-009		2.24-008
	25.0	7.90-009		25.0	7.90-009		8.59-009
	30.0	8.70-009		30.0	8.70-009		1.22-008
	33.4	9.80-009		33.4	9.80-009		

FENTON, ROGERS, AND WOODS (1963), AL-FRW

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.44-008	TEMP	2.0	2.44-008	TEMP	2.0
	3.0	2.41-008		3.0	2.41-008		1.77-008
	4.0	2.37-008		4.0	2.37-008		1.24-008

AMUNOSEN, MYHRE, AND SALTER (1972), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.42-008	TEMP	2.0	2.42-008	TEMP	2.0
	3.0	2.38-008		3.0	2.38-008		1.50-008
	4.0	2.30-008		4.0	2.30-008		1.54-008

ANDREWS, WEBBER, AND SALTHER (1972), AL-AWS

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.44-008	TEMP	2.0	2.44-008	TEMP	2.0
	3.0	2.44-008		3.0	2.44-008		1.90-008
	4.0	2.43-008		4.0	2.43-008		1.98-008

MOORE, MCELROY, AND BARSONI (1966), AL-MMB

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	100.0	1.16-008	TEMP	100.0	1.16-008	TEMP	4.4-007
	120.0	1.30-008		120.0	1.30-008		6.66-007
	140.0	1.43-008		140.0	1.43-008		9.01-007
	160.0	1.55-008		160.0	1.55-008		1.13-006
	180.0	1.68-008		180.0	1.68-008		1.37-006
	200.0	1.80-008		200.0	1.80-008		1.60-006
	220.0	1.90-008		220.0	1.90-008		1.83-006
	240.0	1.99-008		240.0	1.99-008		2.06-006
	260.0	2.07-008		260.0	2.07-008		2.25+000
	280.0	2.14-008		280.0	2.14-008		2.55+000
	300.0	2.18-008		300.0	2.18-008		2.85+000

POWELL, HALL, AND RODGER (1960), AL-PHRC SC

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	2.0	2.48-008	TEMP	2.0	2.48-008	TEMP	2.0
	20.0	2.01-008		20.0	2.01-008		2.01-008
	40.0	1.65-008		40.0	1.65-008		4.00-008
	60.0	1.31-008		60.0	1.31-008		6.00-008
	80.0	1.24-008		80.0	1.24-008		8.00-008

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	
TEMP	85.9	8.27-008	TEMP	85.9	8.27-008	TEMP	8.59-009

Aluminum (cont.)

POWELL, HALL, AND ROOER(1960), AL-PHR(1100-0)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
2•0	2•44-008	2•0	2•44-008
20•0	2•36-008	20•0	2•36-008
40•0	2•10-008	40•0	2•10-008
60•0	1•78-008	60•0	1•78-008
80•0	1•59-008	80•0	1•59-008
100•0	1•55-008	100•0	1•55-008

POWELL, TYF, AND WOODMAN(1965), AL-PTW

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
12•1	1•49-008	-150•0	1•49-008
17•1	1•74-008	-100•0	1•74-008
22•1	1•31-008	-50•0	1•91-008
273•1	2•09-008	0•0	2•09-008

Aluminum Alloys

HUST AND SPARKS(1971C), AL-ALLOYS-HS(T86)

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	RATIO	TEMP	RATIO	TEMP	RATIO
5.0	2.48-008	5.0	2.48-008	5.0	2.44-008
6.0	2.50-008	6.0	2.50-008	6.0	2.46-008
7.0	2.51-008	7.0	2.51-008	7.0	2.48-008
8.0	2.51-008	8.0	2.51-008	8.0	2.50-008
9.0	2.51-008	9.0	2.51-008	9.0	2.50-008
10.0	2.51-008	10.0	2.51-008	10.0	2.50-008
12.0	2.51-008	12.0	2.51-008	12.0	2.50-008
14.0	2.51-008	14.0	2.51-008	14.0	2.49-008
16.0	2.53-008	16.0	2.53-008	16.0	2.49-008
18.0	2.54-008	18.0	2.54-008	18.0	2.48-008
20.0	2.54-008	20.0	2.54-008	20.0	2.48-008
25.0	2.55-008	25.0	2.55-008	25.0	2.45-008
30.0	2.54-008	30.0	2.54-008	30.0	2.41-008
35.0	2.51-008	35.0	2.51-008	35.0	2.35-008
40.0	2.48-008	40.0	2.48-008	40.0	2.28-008
45.0	2.44-008	45.0	2.44-008	45.0	2.21-008
50.0	2.40-008	50.0	2.40-008	50.0	2.14-008
55.0	2.36-008	55.0	2.36-008	55.0	2.08-008
60.0	2.32-008	60.0	2.32-008	60.0	2.03-008
65.0	2.29-008	65.0	2.29-008	65.0	1.98-008
70.0	2.26-008	70.0	2.26-008	70.0	1.94-008
75.0	2.24-008	75.0	2.24-008	75.0	1.91-008
80.0	2.22-008	80.0	2.22-008	80.0	1.89-008
85.0	2.20-008	85.0	2.20-008	85.0	1.87-008
90.0	2.18-008	90.0	2.18-008	90.0	1.85-008
95.0	2.17-008	95.0	2.17-008	95.0	1.84-008
100.0	2.16-008	100.0	2.16-008	100.0	1.84-008
110.0	2.15-008	110.0	2.15-008	110.0	1.83-008
120.0	2.14-008	120.0	2.14-008	120.0	1.84-008
130.0	2.14-008	130.0	2.14-008	130.0	1.86-008
140.0	2.14-008	140.0	2.14-008	140.0	1.88-008
150.0	2.15-008	150.0	2.15-008	150.0	1.91-008
160.0	2.15-008	160.0	2.16-008	160.0	1.94-008
170.0	2.17-008	170.0	2.17-008	170.0	1.97-008
180.0	2.18-008	180.0	2.18-008	180.0	2.01-008
190.0	2.20-008	190.0	2.20-008	190.0	2.04-008
200.0	2.21-008	200.0	2.21-008	200.0	2.06-008
220.0	2.24-008	220.0	2.24-008	220.0	2.12-008
240.0	2.28-008	240.0	2.28-008	240.0	2.16-008
260.0	2.30-008	260.0	2.30-008	260.0	2.19-008
280.0	2.33-008	280.0	2.33-008	280.0	2.23-008

HUST, WEITZEL, AND POWELL(1971), AL ALLOYS-HWP

POWELL, HALL, AND RODER(1960), AL ALLOYS- PHR(L/L)

TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
6.0	2.47-008	6.0	2.47-008	2.0	2.49-008	2.0
7.0	2.50-008	7.0	2.50-008	2.0	2.37-008	2.0
8.0	2.52-008	8.0	2.52-008	4.0	2.17-008	4.0
9.0	2.52-008	9.0	2.52-008	6.0	1.20-008	6.0
10.0	2.52-008	10.0	2.52-008	8.0	1.65-008	8.0
12.0	2.52-008	12.0	2.52-008	10.0	1.61-008	10.0
14.0	2.52-008	14.0	2.52-008	16.0	2.53-008	16.0
16.0	2.53-008	18.0	2.53-008	20.0	2.53-008	20.0
20.0	2.53-008	25.0	2.53-008	30.0	2.51-008	30.0
30.0	2.51-008	35.0	2.49-008	40.0	2.45-008	45.0
40.0	2.44-008	45.0	2.42-008	50.0	2.38-008	55.0
50.0	2.38-008	55.0	2.35-008	60.0	2.31-008	65.0
60.0	2.31-008	65.0	2.29-008	70.0	2.26-008	75.0
70.0	2.26-008	75.0	2.24-008	80.0	2.22-008	85.0
80.0	2.22-008	90.0	2.19-008	95.0	2.18-008	100.0
90.0	2.19-008	100.0	2.18-008	110.0	2.17-008	120.0
110.0	2.11-008	120.0	2.11-008	130.0	2.11-008	140.0
130.0	2.11-008	140.0	2.18-008	150.0	2.20-008	160.0
160.0	2.20-008	170.0	2.21-018	170.0	2.24-008	180.0
180.0	2.26-008	190.0	2.26-008	190.0	2.28-008	200.0
190.0	2.28-008	200.0	2.31-008	200.0	2.36-008	220.0
200.0	2.31-008	220.0	2.36-008	220.0	2.40-008	240.0
220.0	2.36-008	240.0	2.40-008	260.0	2.44-008	260.0
240.0	2.40-008	280.0	2.47-008	280.0	2.47-008	280.0

POWELL, HALL, AND RODER(1960), AL ALLOYS- PHR(5154)

TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
6.0	2.47-008	6.0	2.47-008	2.0	2.49-008	2.0
7.0	2.50-008	7.0	2.50-008	2.0	2.37-008	2.0
8.0	2.52-008	8.0	2.52-008	4.0	2.17-008	4.0
9.0	2.52-008	9.0	2.52-008	6.0	1.20-008	6.0
10.0	2.52-008	10.0	2.52-008	8.0	1.65-008	8.0
12.0	2.52-008	12.0	2.52-008	10.0	1.61-008	10.0
14.0	2.52-008	14.0	2.52-008	16.0	2.53-008	16.0
16.0	2.53-008	18.0	2.53-008	20.0	2.53-008	20.0
20.0	2.53-008	25.0	2.53-008	30.0	2.51-008	35.0
30.0	2.51-008	35.0	2.49-008	40.0	2.45-008	45.0
40.0	2.44-008	45.0	2.42-008	50.0	2.38-008	55.0
50.0	2.38-008	55.0	2.35-008	60.0	2.31-008	65.0
60.0	2.31-008	65.0	2.29-008	70.0	2.26-008	75.0
70.0	2.26-008	75.0	2.24-008	80.0	2.22-008	85.0
80.0	2.22-008	90.0	2.19-008	95.0	2.18-008	100.0
90.0	2.19-008	100.0	2.18-008	110.0	2.17-008	120.0
110.0	2.11-008	120.0	2.11-008	130.0	2.11-008	140.0
130.0	2.11-008	140.0	2.18-008	150.0	2.20-008	160.0
160.0	2.20-008	170.0	2.21-018	170.0	2.24-008	180.0
180.0	2.26-008	190.0	2.26-008	190.0	2.28-008	200.0
190.0	2.28-008	200.0	2.31-008	200.0	2.36-008	220.0
200.0	2.31-008	220.0	2.36-008	220.0	2.40-008	240.0
220.0	2.36-008	240.0	2.40-008	260.0	2.44-008	260.0
240.0	2.40-008	280.0	2.47-008	280.0	2.47-008	280.0

TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
6.0	2.49-008	2.0	2.37-008	2.0	2.18-008	2.0
7.0	2.50-008	2.0	2.37-008	2.0	2.18-008	2.0
8.0	2.52-008	4.0	2.17-008	4.0	2.60-008	4.0
9.0	2.52-008	6.0	1.20-008	6.0	2.61-008	6.0
10.0	2.52-008	8.0	1.20-008	8.0	2.64-008	8.0
12.0	2.52-008	10.0	1.20-008	10.0	2.61-008	10.0
14.0	2.52-008	12.0	1.20-008	12.0	2.64-008	12.0
16.0	2.52-008	14.0	1.20-008	14.0	2.64-008	14.0
17.0	2.2-008	16.0	2.2-008	16.0	2.2-008	17.0
18.0	2.26-008	18.0	2.26-008	18.0	2.26-008	19.0
19.0	2.28-008	19.0	2.28-008	19.0	2.28-008	200.
200.	2.31-008	200.	2.31-008	200.	2.35-008	200.
220.0	2.36-008	220.0	2.36-008	220.0	2.41-008	240.0
240.0	2.40-008	240.0	2.40-008	240.0	2.46-008	260.0
260.0	2.44-008	260.0	2.44-008	260.0	2.46-008	280.0
280.0	2.47-008	280.0	2.47-008	280.0	2.47-008	280.0

POWELL, HALL, AND RODER(1960), AL ALLOYS- PHR(5086)

TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
2.0	2.49-008	2.0	2.49-008	2.0	2.49-008	2.0
2.0	2.18-008	2.0	2.18-008	2.0	2.18-008	2.0
2.0	2.60-008	2.0	2.60-008	2.0	2.60-008	2.0
2.0	2.51-008	4.0	2.51-008	4.0	2.51-008	4.0
4.0	1.20-008	6.0	1.20-008	6.0	2.64-008	6.0
6.0	1.20-008	8.0	1.20-008	8.0	2.64-008	8.0
8.0	1.20-008	10.0	1.20-008	10.0	2.64-008	10.0
10.0	1.20-008	12.0	1.20-008	12.0	2.64-008	12.0
12.0	1.20-008	14.0	1.20-008	14.0	2.64-008	14.0
14.0	1.20-008	16.0	2.2-008	16.0	2.2-008	17.0
17.0	2.2-008	18.0	2.2-008	18.0	2.2-008	19.0
19.0	2.26-008	190.0	2.28-008	190.0	2.28-008	200.
200.	2.31-008	200.	2.31-008	200.	2.35-008	200.
220.0	2.36-008	220.0	2.36-008	220.0	2.41-008	240.0
240.0	2.40-008	240.0	2.40-008	240.0	2.46-008	260.0
260.0	2.44-008	260.0	2.44-008	260.0	2.46-008	280.0
280.0	2.47-008	280.0	2.47-008	280.0	2.47-008	280.0

POWELL, HALL, AND RODER(1960), AL ALLOYS- PHR(5052)

LORENZ RATIO

TEMP RATIO

Lead and Tin

LEES(1908), Pb AND Sn-L (PB)

	LORENZ RATIO	TEMP 103•0	TEMP 111•0	TEMP 123•0	TEMP 148•0	TEMP 173•0	TEMP 198•0	TEMP 223•0	TEMP 248•0	TEMP 273•0	TEMP 291•0
		2•55-008	2•53-008	2•52-008	2•54-008	2•51-008	2•52-008	2•51-008	2•51-008	2•53-008	2•51-008
		103•0	113•0	123•0	148•0	173•0	198•0	223•0	248•0	273•0	291•0
		2•55-008	2•53-008	2•52-008	2•54-008	2•51-008	2•52-008	2•51-008	2•51-008	2•53-008	2•51-008
		103•0	113•0	123•0	148•0	173•0	198•0	223•0	248•0	273•0	291•0

LEES(1908), Pb AND Sn-L (Sn)

	LORENZ RATIO	TEMP 103•0	TEMP 111•0	TEMP 123•0	TEMP 148•0	TEMP 173•0	TEMP 198•0	TEMP 223•0	TEMP 248•0	TEMP 273•0	TEMP 291•0
		2•48-008	2•49-008	2•53-008	2•52-008	2•51-008	2•52-008	2•53-008	2•53-008	2•53-008	2•47-008
		103•0	113•0	123•0	148•0	173•0	198•0	223•0	248•0	273•0	291•0
		2•48-008	2•49-008	2•53-008	2•52-008	2•51-008	2•52-008	2•53-008	2•53-008	2•53-008	2•49-008
		103•0	113•0	123•0	148•0	173•0	198•0	223•0	248•0	273•0	291•0

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
6.0	2.35-008	6.0	2.35-008	7.0	2.16-008	7.0
7.0	2.33-008	7.0	2.44-008	8.0	2.15-008	8.0
8.0	2.39-008	8.0	2.33-008	9.0	2.17-008	9.0
9.0	2.33-008	9.0	2.33-008	10.0	2.16-008	10.0
10.0	2.37-008	10.0	2.27-008	12.0	2.07-008	12.0
12.0	2.15-008	12.0	2.15-008	14.0	1.93-008	14.0
14.0	2.04-008	14.0	2.04-008	16.0	1.79-008	16.0
16.0	1.93-008	16.0	1.93-008	18.0	1.69-008	18.0
18.0	1.84-008	18.0	1.84-008	20.0	1.61-008	20.0
20.0	1.76-008	20.0	1.76-008	25.0	1.54-008	25.0
25.0	1.64-008	25.0	1.64-008	30.0	1.53-008	30.0
30.0	1.50-008	30.0	1.60-008	35.0	1.55-008	35.0
35.0	1.41-008	35.0	1.61-008	40.0	1.59-008	40.0
40.0	1.43-008	40.0	1.63-008	45.0	1.53-008	45.0
45.0	1.47-008	45.0	1.67-008	50.0	1.68-008	50.0
50.0	1.61-008	50.0	1.71-008	55.0	1.73-008	55.0
55.0	1.77-008	55.0	1.77-008	60.0	1.79-008	60.0
60.0	1.82-008	60.0	1.82-008	65.0	1.85-008	65.0
65.0	1.87-008	65.0	1.87-008	70.0	1.90-008	70.0
70.0	1.92-008	70.0	1.92-008	75.0	1.95-008	75.0
75.0	1.97-008	75.0	1.97-008	80.0	2.00-008	80.0
80.0	2.01-008	80.0	2.01-008	85.0	2.05-008	85.0
85.0	2.05-008	85.0	2.05-008	90.0	2.09-008	90.0
90.0	2.08-008	90.0	2.08-008	95.0	2.12-008	95.0
95.0	2.11-008	95.0	2.11-008	100.0	2.15-008	100.0
100.0	2.14-008	100.0	2.14-008	110.0	2.20-008	110.0
110.0	2.19-008	110.0	2.19-008	120.0	2.23-008	120.0
120.0	2.22-008	120.0	2.22-008	130.0	2.26-008	130.0
130.0	2.26-008	130.0	2.26-008	140.0	2.28-008	140.0
140.0	2.29-008	140.0	2.29-008	150.0	2.30-008	150.0
150.0	2.32-008	150.0	2.32-008	160.0	2.31-008	160.0
160.0	2.33-008	160.0	2.33-008	170.0	2.32-008	170.0
170.0	2.33-008	170.0	2.33-008	180.0	2.34-008	180.0
				190.0	2.36-008	190.0
				200.0	2.37-008	200.0
				220.0	2.39-008	220.0
				240.0	2.40-008	240.0
				260.0	2.41-008	260.0
				280.0	2.41-008	280.0

HUST AND SPARKS(1971), AU-HS(3)

	LORENZ TEMP	LORENZ RATIO	TEMP RATIO
7.0	2.21-008	7.0	2.21-008
8.0	1.99-008	8.0	1.99-008
9.0	1.90-008	9.0	1.80-008
10.0	1.84-008	10.0	1.84-008
12.0	1.70-008	12.0	1.70-008
14.0	1.53-008	14.0	1.53-008
16.0	1.41-008	16.0	1.61-008
18.0	1.34-008	18.0	1.34-008
20.0	1.32-008	20.0	1.32-008
25.0	1.36-008	25.0	1.36-008
30.0	1.42-008	30.0	1.42-008
35.0	1.46-008	35.0	1.46-008
40.0	1.51-008	40.0	1.51-008
45.0	1.56-008	45.0	1.56-008
50.0	1.62-008	50.0	1.62-008
55.0	1.69-008	55.0	1.69-008
60.0	1.75-008	60.0	1.75-008
65.0	1.82-008	65.0	1.82-008
70.0	1.88-008	70.0	1.88-008
75.0	1.93-008	75.0	1.93-008
80.0	1.98-008	80.0	1.98-008
85.0	2.02-008	85.0	2.02-008
90.0	2.06-008	90.0	2.06-008
95.0	2.10-008	95.0	2.10-008
100.0	2.13-008	100.0	2.13-008
110.0	2.17-008	110.0	2.17-008
120.0	2.21-008	120.0	2.21-008
130.0	2.24-008	130.0	2.24-008
140.0	2.27-008	140.0	2.27-008
150.0	2.29-008	150.0	2.29-008
160.0	2.31-008	160.0	2.31-008
170.0	2.33-008	170.0	2.33-008

Gold Cobalt Alloy

POWELL, BUNCH, AND GIBSON(1960), AU CO ALLOY-PBG

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
4•0	3•01-008	4•0	3•01-008	
10•0	4•90-008	10•0	4•90-008	
15•0	5•20-008	15•0	5•20-008	
20•0	5•08-008	20•0	5•08-008	
30•0	4•51-008	30•0	4•51-008	
40•0	4•02-008	40•0	4•02-008	
50•0	3•70-008	50•0	3•70-008	
60•0	3•45-008	60•0	3•45-008	
70•0	3•27-008	70•0	3•27-008	
80•0	3•12-008	80•0	3•12-008	
90•0	3•00-008	90•0	3•00-008	

FENTON, ROGERS, AND WOODS(1963), AG-FRW (1)

MALM AND WOODS(1966), AG-NW

	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
2•2	2•43-008	2•2	2•4-008	2•0	2•45-008	2•0	4•90-007	1•00-001
4•0	2•34-008	4•0	2•34-008	3•0	2•51-008	3•0	5•03-007	1•50-001
8•0	1•75-008	8•0	1•75-008	4•0	2•53-008	4•0	5•10-007	2•03-001
10•9	1•31-008	10•9	1•31-008	6•0	2•93-008	6•0	5•13-007	3•50-001
FENTON, ROGERS, AND WOODS(1963), AG-FRW (2)								
TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
2•0	2•41-008	2•0	2•41-008	10•0	3•57-008	8•0	5•60-001	5•60-001
4•0	2•25-008	4•0	2•25-008	10•0	3•56-008	10•0	5•08-007	7•00-001
8•0	1•58-008	8•0	1•58-008	12•0	3•81-008	12•0	5•08-007	9•00-001
14•0	1•01-008	14•0	1•01-008	14•0	3•89-008	14•0	5•07-007	1•02-000
16•5	1•00-008	16•5	1•00-008	16•0	3•65-008	16•0	5•08-007	1•15-000
MALM AND WOODS(1966), AG-NW								
KANNULUIK(1933), AG-K(1)								
TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
273•1	2•31-008	0•0	2•31-008	2•0	2•48-008	2•0	1•72-007	2•88-001
194•6	2•24-008	-78•5	2•24-008	3•0	2•68-008	3•0	1•73-007	4•6-001
90•1	1•80-008	-183•0	1•80-008	4•0	2•72-008	4•0	1•73-007	6•30-001
KANNULUIK(1933), AG-K(2)								
TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
273•1	2•31-008	0•0	2•31-008	10•0	2•84-008	8•0	1•71-007	1•00-000
194•6	2•24-008	-78•5	2•24-008	10•0	2•93-008	8•0	1•70-007	1•38-000
90•1	1•62-008	-183•0	1•62-008	12•0	3•26-008	12•0	1•69-007	1•69-000
MALM AND WOODS(1966), AG-NW								
LEES(1908), AG-L								
TEMP	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO	LORENZ RATIO	TEMP RATIO
10•4•0	2•04-008	10•3•0	2•04-008	2•0	2•64-008	2•0	9•60-008	5•49-001
11•1•0	2•09-008	11•3•0	2•09-008	3•0	2•75-008	3•0	9•50-008	8•70-001
12•0•0	2•13-008	12•3•0	2•13-008	4•0	2•88-008	4•0	9•44-008	1•20-000
14•0•0	2•24-008	14•8•0	2•24-008	6•0	2•88-008	6•0	9•30-008	1•86-000
17•3•0	2•29-008	17•3•0	2•29-008	8•0	2•90-008	8•0	9•25-008	2•51-000
19•8•0	2•34-008	19•8•0	2•34-008	10•0	3•03-008	10•0	9•22-008	3•31-000
22•3•0	2•36-008	22•3•0	2•36-008	12•0	2•82-008	12•0	9•23-008	3•71-000
24•0•0	2•35-008	24•8•0	2•35-008	14•0	2•76-008	14•0	9•29-008	4•16-000
27•3•0	2•33-008	27•3•0	2•33-008	16•0	2•74-008	16•0	9•46-008	4•6-000
29•1•0	2•33-008	29•1•0	2•33-008	18•0	2•54-008	18•0	9•56-008	4•79-000
				20•0	2•55-008	20•0	9•75-008	5•24-000
				22•0	2•46-008	22•0	1•01-007	5•36-000

MALM AND WOODS(1966) • AG-MW

TEMP	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
2•0	2•72-0•08	2•0	1•14-0•08	4•78-0•00
3•0	2•52-0•08	3•0	1•12-0•08	6•75-0•00
4•0	2•66-0•08	4•0	1•12-0•08	9•54-0•00
6•0	2•73-0•08	6•0	1•11-0•08	1•48-0•01
8•0	2•62-0•08	8•0	1•11-0•08	1•90-0•01
10•0	2•442-0•08	10•0	1•11-0•08	2•18-0•01
12•0	2•26-0•08	12•0	1•13-0•08	2•40-0•01
14•0	2•08-0•08	14•0	1•16-0•08	2•51-0•01

Copper

LEE(1908), CU-L

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
103.0	1.85-008	103.0	1.85-008	103.0	1.85-008	103.0	1.83-008	5.0	1.83-008
113.0	1.92-008	113.0	1.92-008	113.0	1.92-008	113.0	1.91-008	5.0	1.51-008
123.0	1.97-008	123.0	1.97-008	123.0	1.97-008	123.0	1.97-008	10.0	1.51-008
148.0	2.07-008	148.0	2.07-008	148.0	2.07-008	148.0	2.07-008	15.0	1.07-008
173.0	2.17-008	173.0	2.17-008	173.0	2.17-008	173.0	2.17-008	20.0	8.80-009
198.0	2.23-008	198.0	2.23-008	198.0	2.23-008	198.0	2.23-008	25.0	8.40-009
248.0	2.26-008	248.0	2.26-008	248.0	2.26-008	248.0	2.26-008	35.0	9.30-009
273.0	2.30-008	273.0	2.30-008	273.0	2.30-008	273.0	2.30-008	45.0	1.04-008
291.0	2.32-008	291.0	2.32-008	291.0	2.32-008	291.0	2.32-008	55.0	1.15-008
								55.0	1.15-008
								65.0	1.26-008
								75.0	1.37-008
								80.0	1.42-008
									80.0

MOORE, MCILROY, AND GRAVES(1967), CU-MMG

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
85.0	1.55-008	85.0	1.48-007	85.0	1.48-007	85.0	1.52-007	5.0	2.36-008
90.0	1.59-008	90.0	1.52-007	90.0	1.52-007	90.0	1.55-008	5.0	1.85-008
100.0	1.68-008	100.0	1.50-007	100.0	1.50-007	100.0	1.65-008	10.0	1.20-008
110.0	1.76-008	110.0	1.48-007	110.0	1.48-007	110.0	1.50-008	15.0	1.20-008
120.0	1.83-008	120.0	1.45-007	120.0	1.48-007	120.0	1.50-008	20.0	9.20-009
130.0	1.89-008	130.0	1.42-007	130.0	1.45-007	130.0	1.50-009	25.0	8.50-009
140.0	1.96-008	140.0	1.41-007	140.0	1.44-007	140.0	1.50-009	30.0	8.90-009
150.0	2.00-008	150.0	1.39-007	150.0	1.42-007	150.0	1.50-009	35.0	9.90-009
175.0	2.10-008	175.0	1.36-007	175.0	1.39-007	175.0	1.56-008	40.0	1.06-008
220.0	2.16-008	220.0	1.05-006	220.0	1.05-006	220.0	1.13-008	45.0	1.13-008
250.0	2.22-008	250.0	1.22-006	250.0	1.22-006	250.0	1.20-008	50.0	1.20-008
275.0	2.26-008	275.0	1.39-006	275.0	1.39-006	275.0	1.27-008	55.0	1.27-008
300.0	2.31-008	300.0	1.33-006	300.0	1.33-006	300.0	1.33-008	60.0	1.33-008
								65.0	1.39-008
								70.0	1.45-008
								75.0	1.51-008
								80.0	1.57-008
								85.0	1.61-008
								90.0	1.67-008
								95.0	1.71-008
								100.0	1.75-008

POWELL, RODER, AND HALL(1959), CU-PRH(ANN)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
10.0	2.04-008	10.0	2.04-008	10.0	2.04-008	10.0	2.04-008	5.0	1.83-008
15.0	1.92-008	15.0	1.92-008	15.0	1.92-008	15.0	1.92-008	10.0	1.51-008
20.0	1.75-008	20.0	1.75-008	20.0	1.75-008	20.0	1.75-008	15.0	1.20-008
25.0	1.56-008	25.0	1.56-008	25.0	1.56-008	25.0	1.56-008	20.0	9.20-009
35.0	1.28-008	35.0	1.28-008	35.0	1.28-008	35.0	1.28-008	25.0	8.50-009
45.0	1.23-008	45.0	1.23-008	45.0	1.23-008	45.0	1.23-008	30.0	7.80-009
55.0	1.28-008	55.0	1.28-008	55.0	1.28-008	55.0	1.28-008	35.0	7.10-009
60.0	1.32-008	60.0	1.32-008	60.0	1.32-008	60.0	1.32-008	40.0	6.40-009
65.0	1.36-008	65.0	1.36-008	65.0	1.36-008	65.0	1.36-008	45.0	6.10-009
70.0	1.39-008	70.0	1.39-008	70.0	1.39-008	70.0	1.39-008	50.0	5.80-009
75.0	1.44-008	75.0	1.44-008	75.0	1.44-008	75.0	1.44-008	55.0	5.50-009
80.0	1.46-008	80.0	1.46-008	80.0	1.46-008	80.0	1.46-008	60.0	5.20-009
85.0	1.51-008	85.0	1.51-008	85.0	1.51-008	85.0	1.51-008	65.0	5.00-009

WHITE AND TAINSH(WT), CU-WT

POWELL, RODER, AND HALL(1959), CU-PRH(AN)

LORENZ
RATIO

TEMP

LORENZ
RATIO

Copper Alloys (German Silver and Brass)

ALLEN AND MENDOZA(1947), CU ALLOYS-AM

	LORENZ RATIO	ELECTRICAL RESISTIVITY	TEMP	TEMP	LORENZ RATIO	ELECTRICAL RESISTIVITY	TEMP	TEMP	LORENZ RATIO
TEMP	1•4	3•50-009	1•4	3•27-006	1•50-003	1•50-003	103•0	103•0	6•51-008
	2•0	7•19-009	2•0	3•27-006	4•40-003	4•40-003	113•0	113•0	6•06-008
	3•0	1•56-008	3•0	3•27-006	1•43-002	1•43-002	123•0	123•0	5•69-008
	4•0	2•80-008	4•0	3•27-006	3•43-002	3•43-002	148•0	148•0	4•93-008
 BERMAN(1951) • CUALLOYS-B (G-AG)									
	LORENZ RATIO	ELECTRICAL RESISTIVITY	TEMP	TEMP	LORENZ RATIO	ELECTRICAL RESISTIVITY	TEMP	TEMP	LORENZ RATIO
TEMP	4•0	9•18-008	4•0	4•42-005	8•31-003	3•44-003	103•0	103•0	5•13-008
	10•0	1•53-007	10•0	4•42-005	3•44-002	3•44-002	113•0	113•0	4•82-008
	20•0	1•08-007	20•0	2•43-005	8•89-002	8•89-002	123•0	123•0	4•56-008
	50•0	1•51-007	50•0	4•45-005	1•70-001	1•70-001	148•0	148•0	4•00-008
	81•2	1•07-007	81•2	4•55-005	1•90-001	1•90-001	173•0	173•0	3•61-008

LEES(1908), CU ALLOYS-L (PTD)

	LORENZ RATIO	ELECTRICAL CONDUCTIVITY	TEMP	TEMP	LORENZ RATIO	ELECTRICAL CONDUCTIVITY	TEMP	TEMP	LORENZ RATIO
TEMP	4•47-008	4•42-005	4•42-005	4•42-005	4•42-005	4•42-005	103•0	103•0	5•13-008
	7•29-008	4•0	2•65-005	2•65-005	6•75-003	6•75-003	113•0	113•0	4•82-008
	10•0	9•40-008	10•0	2•65-005	2•65-005	2•65-005	123•0	123•0	4•56-008
	20•0	7•96-008	20•0	2•66-005	2•66-005	2•66-005	148•0	148•0	4•00-008
	50•0	7•96-008	50•0	2•69-005	1•48-001	1•48-001	173•0	173•0	3•38-008
	81•2	5•75-008	81•2	2•75-005	1•70-001	1•70-001	223•0	223•0	3•21-008

KARWEIL AND SCHAFER(1939), CU ALLOYS-KS(SB)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
TEMP	5•0	5•40-008	5•0	5•40-008	103•0	103•0	2•78-008	103•0	2•78-008
	10•0	6•40-008	10•0	6•40-008	113•0	113•0	2•71-008	113•0	2•71-008
	15•0	8•20-008	15•0	8•20-008	123•0	123•0	2•65-008	123•0	2•65-008
	20•0	7•20-008	20•0	7•20-008	148•0	148•0	2•57-008	148•0	2•57-008

KARWEIL AND SCHAFER(1939), CU ALLOYS-KS(NS)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
TEMP	3•0	4•10-008	3•0	4•10-008	198•0	198•0	2•51-008	198•0	2•51-008
	5•0	4•90-008	5•0	4•90-008	223•0	223•0	2•47-008	223•0	2•47-008
	10•0	7•90-008	10•0	7•90-008	248•0	248•0	2•46-008	248•0	2•46-008
	15•0	9•40-008	15•0	9•40-008	273•0	273•0	2•45-008	273•0	2•45-008
	20•0	9•90-008	20•0	9•90-008	291•0	291•0	2•45-008	291•0	2•45-008

Zinc and Cadmium

GOENS AND GRUNENSEN(1932) • ZN AND CO-GG(ZN-PERPENDICULAR)

	LORENZ RATIO	TEMP	LORENZ RATIO									
TEMP	2•56-008	293•2	2•56-008	103•0	2•39-008	103•0	2•39-008	103•0	2•39-008	103•0	2•39-008	
	83•2	2•04-008	93•2	2•04-008	113•0	2•41-008	113•0	2•41-008	113•0	2•41-008	113•0	2•41-008
	21•2	1•47-008	83•2	2•04-008	123•0	2•42-008	123•0	2•42-008	123•0	2•42-008	123•0	2•42-008
			21•2	1•47-008	148•0	2•43-008	148•0	2•43-008	148•0	2•43-008	148•0	2•43-008
LORENZ RATIO					173•0	2•43-008	173•0	2•43-008	173•0	2•43-008	173•0	2•43-008
TEMP	2•47-008	293•2	2•47-008	198•0	2•42-008	198•0	2•42-008	198•0	2•42-008	198•0	2•42-008	
	83•2	1•90-008	293•2	2•47-008	223•0	2•41-008	223•0	2•41-008	223•0	2•41-008	223•0	2•41-008
	21•2	9•70-009	83•2	1•90-008	248•0	2•40-008	248•0	2•40-008	248•0	2•40-008	248•0	2•40-008
			21•2	9•70-009	273•0	2•40-008	273•0	2•40-008	273•0	2•40-008	273•0	2•40-008
LORENZ RATIO					291•0	2•39-008	291•0	2•39-008	291•0	2•39-008	291•0	2•39-008

LEES(1908) • ZN AND CO-L(ZN)

	LORENZ RATIO	TEMP	LORENZ RATIO									
TEMP	2•29-008	103•0	2•29-008	103•0	2•29-008	113•0	2•25-008	113•0	2•25-008	113•0	2•25-008	
	113•0	2•25-008	123•0	2•30-008	123•0	2•30-008	123•0	2•30-008	123•0	2•30-008	123•0	2•30-008
	123•0	2•30-008	148•0	2•33-008	148•0	2•33-008	148•0	2•33-008	148•0	2•33-008	148•0	2•33-008
	148•0	2•33-008	173•0	2•39-008	173•0	2•39-008	173•0	2•39-008	173•0	2•39-008	173•0	2•39-008
	173•0	2•41-008	198•0	2•41-008	198•0	2•41-008	198•0	2•41-008	198•0	2•41-008	198•0	2•41-008
LORENZ RATIO					223•0	2•40-008	223•0	2•40-008	223•0	2•40-008	223•0	2•40-008
TEMP	2•44-008	248•0	2•44-008	248•0	2•44-008	248•0	2•44-008	248•0	2•44-008	248•0	2•44-008	
	83•2	2•45-008	273•0	2•45-008	273•0	2•45-008	273•0	2•45-008	273•0	2•45-008	273•0	2•45-008
	21•2	1•60-008	291•0	2•43-008	291•0	2•43-008	291•0	2•43-008	291•0	2•43-008	291•0	2•43-008
LORENZ RATIO												

GOENS AND GRUNENSEN(1932) • ZN AND CO-GG(CO-PARALLEL)

	LORENZ RATIO	TEMP	LORENZ RATIO									
TEMP	2•31-008	293•2	2•31-008	293•2	2•31-008	83•2	2•37-008	83•2	2•37-008	83•2	2•44-008	
	83•2	2•20-008	21•2	1•60-008	21•2	1•60-008	21•2	1•60-008	21•2	1•60-008	21•2	1•60-008
LORENZ RATIO					248•0	2•44-008	248•0	2•44-008	248•0	2•44-008	248•0	2•44-008
TEMP	2•21-008	21•2	1•27-008	21•2	1•27-008	83•2	2•21-008	83•2	1•27-008	83•2	1•27-008	

GOENS AND GRUNENSEN(1932) • ZN AND CO-GG(ZN-PERPENDICULAR)

Scandium and Yttrium

ARAJA AND COLVIN(1964), SC AND Y-AC

	LORENZ		LORENZ
TEMP	RATIO	TEMP	RATIO
4.2	7.00-008	4.2	7.00-008
20.0	6.00-008	20.0	6.00-008
40.0	4.20-008	40.0	4.20-008
60.0	3.30-008	60.0	3.30-008
80.0	3.10-008	80.0	3.10-008
120.0	3.15-008	120.0	3.15-008
160.0	3.30-008	160.0	3.30-008
200.0	3.60-008	200.0	3.60-008
240.0	3.85-008	240.0	3.85-008
280.0	4.35-008	280.0	4.35-008
300.0	4.65-008	300.0	4.65-008

TAMARIN, CHUPRIKOV, AND SHALYT(1969), SC AND Y- TCS(PARALLEL)

	LORENZ		LORENZ
TEMP	RATIO	TEMP	RATIO
5.0	4.20-008	5.0	4.20-008
10.0	4.70-008	10.0	4.70-008
17.0	5.00-008	17.0	5.00-008
20.0	4.90-008	20.0	4.90-008
30.0	4.50-008	30.0	4.50-008
40.0	4.10-008	40.0	4.10-008
50.0	3.95-008	50.0	3.95-008
60.0	3.90-008	60.0	3.90-008
70.0	3.80-008	70.0	3.80-008
80.0	3.77-008	80.0	3.77-008
90.0	3.75-008	90.0	3.75-008
100.0	3.72-008	100.0	3.72-008
110.0	3.70-008	110.0	3.70-008
120.0	3.68-008	120.0	3.68-008
130.0	3.67-008	130.0	3.67-008
140.0	3.66-008	140.0	3.66-008
150.0	3.65-008	150.0	3.65-008

TAMARIN, SHUPRIKOV, AND SHALYT(1969), SC AND Y- TCS PERPENDICULAR)

	LORENZ		LORENZ
TEMP	RATIO	TEMP	RATIO
5.0	3.00-008	5.0	3.00-008
10.0	3.05-008	10.0	3.05-008
14.0	3.10-008	14.0	3.10-008
20.0	3.00-008	20.0	3.00-008
30.0	2.47-008	30.0	2.47-008
40.0	2.00-008	40.0	2.00-008
50.0	1.95-008	50.0	1.95-008
60.0	1.90-008	60.0	1.90-008
70.0	1.95-008	70.0	1.95-008
80.0	2.00-008	80.0	2.00-008
100.0	2.03-008	100.0	2.03-008
120.0	2.47-008	120.0	2.47-008
140.0	2.50-008	140.0	2.50-008
150.0	2.53-008	150.0	2.53-008

Titanium, Hafnium, and Zirconium

KEMP, KLEMENS, AND WHITE (1956), Ti+HF, AND ZR-KKW (Ti)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
3•3	2•05-008	3•3	2•95-008
10•0	3•38-008	10•0	3•58-008
20•0	4•30-008	20•0	4•30-008
30•0	4•62-008	30•0	4•62-008
40•0	4•62-008	40•0	4•62-008
50•0	4•67-008	50•0	4•44-008
60•0	4•81-008	60•0	4•44-008
70•0	4•80-008	70•0	4•40-008
80•0	4•38-008	80•0	4•38-008
90•0	4•38-008	90•0	4•38-008

KEMP, KLEMENS, AND WHITE (1956), Ti+HF, AND ZR-KKKWW (ZR)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
2•1	2•39-008	2•1	2•39-008
10•0	2•39-008	10•0	2•39-008
20•0	2•46-008	20•0	2•46-008
30•0	2•45-008	30•0	2•45-008
40•0	2•41-008	40•0	2•31-008
50•0	2•27-008	50•0	2•27-008
60•0	2•33-008	60•0	2•33-008
70•0	2•55-008	70•0	2•45-008
80•0	2•57-008	80•0	2•57-008
90•0	2•68-008	90•0	2•68-008

WHITE AND WOODS (1957A), Ti+HF, AND ZR - WW(HF)

TEMP	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	TEMP	THermal CONDUCTIVITY
10•0	4•23-008	10•0	4•23-006	10•0	1•00-001
15•0	4•27-008	15•0	4•27-006	15•0	1•50-001
20•0	3•92-008	20•0	4•36-006	18•0	1•80-001
30•0	3•64-008	30•0	4•97-006	22•0	2•20-001
40•0	3•32-008	40•0	5•53-006	24•0	2•40-001
50•0	3•20-008	50•0	6•53-006	24•5	2•45-001
75•0	3•17-008	75•0	9•43-006	52•0	2•52-001
90•0	3•24-008	90•0	1•12-005	60•0	2•60-001

HUST, WEITZEL, AND POWELL(1971) • Ti • ALLOYS		— HWP	
TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
7•0	1•29-007	7•0	1•29-007
8•0	1•32-007	8•0	1•32-007
9•0	1•34-007	9•0	1•34-007
10•0	1•36-007	10•0	1•36-007
12•0	1•38-007	12•0	1•38-007
14•0	1•38-007	14•0	1•38-007
16•0	1•37-007	16•0	1•37-007
18•0	1•35-007	18•0	1•35-007
20•0	1•33-007	20•0	1•33-007
25•0	1•27-007	25•0	1•27-007
30•0	1•20-007	30•0	1•20-007
35•0	1•14-007	35•0	1•14-007
40•0	1•08-007	40•0	1•08-007
45•0	1•03-007	45•0	1•03-007
50•0	9•86-008	50•0	9•86-008
55•0	9•43-008	55•0	9•43-008
60•0	9•04-008	60•0	9•04-008
65•0	8•69-008	65•0	8•69-008
70•0	8•36-008	70•0	8•36-008
75•0	8•06-008	75•0	8•06-008
80•0	7•78-008	80•0	7•78-008
85•0	7•53-008	85•0	7•53-008
90•0	7•30-008	90•0	7•30-008
95•0	7•08-008	95•0	7•08-008
100•0	6•89-008	100•0	6•89-008
110•0	6•54-008	110•0	6•54-008
120•0	6•25-008	120•0	6•25-008
130•0	6•00-008	130•0	6•00-008
140•0	5•78-008	140•0	5•78-008
150•0	5•60-008	150•0	5•60-008
160•0	5•45-008	160•0	5•45-008
170•0	5•32-008	170•0	5•32-008
180•0	5•20-008	180•0	5•20-008
190•0	5•11-008	190•0	5•11-008
200•0	5•02-008	200•0	5•02-008
220•0	4•88-008	220•0	4•88-008
240•0	4•76-008	240•0	4•76-008
260•0	4•66-008	260•0	4•66-008
280•0	4•56-008	280•0	4•56-008
300•0	4•45-008	300•0	4•45-008

Tungsten

BACILLUNO (1967), w = B

	LORENZ RATIO	TEMP 100•0	ELECTRICAL RESISTIVITY 1.23•-006	Thermal CONDUCTIVITY 1.91•-000
100•0	2.79•-008	150•0	2.35•-006	1.78•-000
150•0	2.79•-008	150•0	2.35•-006	1.71•-000
200•0	2.99•-008	200•0	3.50•-006	1.67•-000
250•0	3.14•-008	250•0	4.70•-006	1.63•-000
300•0	3.12•-008	300•0	5.75•-006	1.63•-000

DIEHAAS AND OENOBLE (1938), w = 00

	LORENZ RATIO	TEMP 77•4	ELECTRICAL RESISTIVITY 5.59•-007	Thermal CONDUCTIVITY 2.60•-000
74•3	1.81•-008	74•3	4.99•-007	2.69•-000
71•3	1.74•-008	71•3	4.46•-007	2.77•-000
68•2	1.66•-008	68•2	3.94•-007	2.87•-000
65•8	1.60•-008	65•8	3.56•-007	2.95•-000
63•5	1.56•-008	63•5	3.23•-007	3.07•-000
60•6	9.82•-009	50•6	1.42•-007	3.50•-000
90•2	2.17•-008	90•2	8.57•-007	2.43•-000
85•1	2.04•-008	85•1	8.04•-007	2.47•+000
74•9	1.82•-008	74•9	5•11•-007	2.68•-000
69•8	1.71•-008	69•8	4•23•-007	2.82•-000
65•2	1.58•-008	65•2	3•41•-007	2.96•-000
20•4	1.07•-008	20•4	4•20•-009	5.22•-001
20•4	1.05•-008	20•4	4•10•-009	6.69•+001
17•5	1.18•-008	17•5	3•10•-009	

KANNULUIK (1933), w = K

	LORENZ RATIO	TEMP 0•0	ELECTRICAL RESISTIVITY 3.04•-008	Thermal CONDUCTIVITY 78•5
273•1	3.04•-008	194•6	2.80•-008	-183•0
90•1	1.91•-008			

KANNULUIK (1933), w = K

	LORENZ RATIO	TEMP 0•0	ELECTRICAL RESISTIVITY 3.06•-008	Thermal CONDUCTIVITY 78•5
273•1	3.06•-008	194•6	2.90•-008	-183•0
90•1	2.00•-008			

MOORE, MCELROY, AND BARSONI (1966), w = MMB

	LORENZ RATIO	TEMP 80•0	ELECTRICAL RESISTIVITY 1.72•-008	Thermal CONDUCTIVITY 2.33•-008
100•0	1.91•-000	100•0	1.00•-000	1.00•-000
150•0	2.79•-008	120•0	2.54•-008	120•0
200•0	3.50•-006	140•0	2.74•-008	140•0
250•0	4.70•-006	160•0	2.88•-008	160•0
300•0	5.75•-006	180•0	2.99•-008	180•0

DEHAAS AND OENOBLE (1938), w = 00

	LORENZ RATIO	TEMP 80•0	ELECTRICAL RESISTIVITY 1.72•-008	Thermal CONDUCTIVITY 2.33•-008
100•0	1.91•-000	100•0	1.00•-000	1.00•-000
150•0	2.69•-000	120•0	2.65•-008	120•0
200•0	3.46•-007	140•0	2.83•-008	140•0
250•0	4.46•-007	160•0	2.95•-008	160•0
300•0	5.56•-007	180•0	3.03•-008	180•0

MOORE, MCELROY, AND BARSONI (1966), w = MMB

	LORENZ RATIO	TEMP 80•0	ELECTRICAL RESISTIVITY 1.72•-008	Thermal CONDUCTIVITY 2.33•-008
100•0	1.91•-000	100•0	1.00•-000	1.00•-000
150•0	2.69•-000	120•0	2.65•-008	120•0
200•0	3.46•-007	140•0	2.83•-008	140•0
250•0	4.46•-007	160•0	2.95•-008	160•0
300•0	5.56•-007	180•0	3.03•-008	180•0

WHITE AND WOODS (1957B), w = ww

	LORENZ RATIO	TEMP 100•0	ELECTRICAL RESISTIVITY 2.40•-008	Thermal CONDUCTIVITY 3.00•-008
100•0	1.91•-000	100•0	1.00•-000	1.00•-000
150•0	2.69•-000	120•0	2.65•-008	120•0
200•0	3.46•-007	140•0	2.83•-008	140•0
250•0	4.46•-007	160•0	2.95•-008	160•0
300•0	5.56•-007	180•0	3.03•-008	180•0

WHITE AND WOODS (1957B), w = ww

	LORENZ RATIO	TEMP 100•0	ELECTRICAL RESISTIVITY 2.47•-008	Thermal CONDUCTIVITY 3.07•-008
100•0	1.91•-000	100•0	1.00•-000	1.00•-000
150•0	2.69•-000	120•0	2.65•-008	120•0
200•0	3.46•-007	140•0	2.83•-008	140•0
250•0	4.46•-007	160•0	2.95•-008	160•0
300•0	5.56•-007	180•0	3.03•-008	180•0

WAGNER, GARLANO, AND BOWERS(1971), W- wGB

	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO
1•5	2.43•008	1•5	2.43•008	1•5	2.39•008	2•0	2.07•008	1•5
2•0	2.39•008	2•0	2.33•008	2•5	2.33•008	2•0	2.07•008	2•0
2•5	2.33•008	2•5	2.29•008	3•0	2.29•008	2•5	1.93•008	2•5
3•0	2.29•008	3•0	2.29•008	3•5	2.20•008	3•0	1.80•008	3•0
3•5	2.20•008	3•5	2.15•008	4•0	2.15•008	3•5	1.60•008	3•5
4•0	2.15•008	4•5	2.09•008	4•5	2.09•008	4•0	1.50•008	4•0
4•5	2.09•008	5•0	2.00•008	5•5	1.91•008	5•0	1.38•008	4•5
5•0	1.91•008	5•5	1.91•008	6•0	1.82•008	5•5	1.27•008	5•0
5•5	1.91•008	6•0	1.82•008	6•0	1.82•008	5•5	1.17•008	5•5
6•0	1.82•008	6•0	1.82•008	6•0	1.82•008	6•0	1.09•008	6•0

WAGNER, GARLANO, AND BOWERS(1971), W- wGB

	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO
1•5	2.32•008	1•5	2.32•008	2•0	2.21•008	2•5	2.10•008	1•5
2•0	2.21•008	2•5	2.10•008	3•0	1.97•008	3•5	1.82•008	2•0
2•5	2.10•008	3•0	1.97•008	3•5	1.82•008	4•0	1.69•008	2•5
3•0	1.97•008	3•5	1.82•008	4•0	1.69•008	4•5	1.56•008	3•0
3•5	1.82•008	4•0	1.69•008	4•5	1.58•008	5•0	1.45•008	3•5
4•0	1.69•008	4•5	1.58•008	5•0	1.50•008	5•5	1.35•008	4•5
4•5	1.58•008	5•0	1.50•008	5•5	1.40•008	6•0	1.22•008	5•0
5•0	1.50•008	5•5	1.40•008	6•0	1.33•008	6•0	1.13•008	5•5
5•5	1.40•008	6•0	1.33•008	6•0	1.33•008	6•0	9.70•009	6•0
6•0	1.33•008	6•0	1.33•008	6•0	1.33•008	6•0	9.10•009	6•0

WAGNER, GARLANO, AND BOWERS(1971), W- wGB

	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO
1•5	2.28•008	1•5	2.28•008	2•0	2.12•008	2•5	1.98•008	1•5
2•0	2.12•008	2•5	1.98•008	3•0	1.80•008	3•5	1.62•008	2•0
2•5	1.98•008	3•0	1.80•008	3•5	1.62•008	4•0	1.43•008	2•5
3•0	1.80•008	3•5	1.62•008	4•0	1.43•008	4•5	1.25•008	3•0
3•5	1.62•008	4•0	1.43•008	4•5	1.21•008	5•0	1.07•008	3•5
4•0	1.53•008	4•5	1.41•008	5•0	1.32•008	5•5	1.22•008	4•0
4•5	1.41•008	5•0	1.32•008	5•5	1.22•008	6•0	1.13•008	4•5
5•0	1.32•008	5•5	1.22•008	6•0	1.18•008	6•0	9.70•009	6•0
5•5	1.22•008	6•0	1.18•008	6•0	1.18•008	6•0	9.70•009	6•0

WAGNER, GARLANO, AND BOWERS(1971), W- wGB

	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO	LORENZ TEMP	LORENZ RATIO
1•5	2.43•008	1•5	2.39•008	2•0	2.39•008	2•5	2.09•008	1•5
2•0	2.39•008	2•5	2.29•008	3•0	2.29•008	3•5	1.80•008	2•5
2•5	2.29•008	3•0	2.20•008	3•5	2.15•008	4•0	1.60•008	3•5
3•0	2.20•008	3•5	2.09•008	4•0	2.09•008	4•5	1.38•008	4•5
3•5	2.09•008	4•0	2.00•008	4•5	1.91•008	5•0	1.27•008	5•0
4•0	2.00•008	4•5	1.91•008	5•0	1.82•008	5•5	1.17•008	5•5
4•5	1.91•008	5•0	1.82•008	5•5	1.82•008	6•0	1.09•008	6•0

Molybdenum

BACKLUNO(1967), MO - B

TEMP	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
100.0	1.74-008	100.0	1.05-006	1.66+000
150.0	2.24-008	150.0	2.30-006	1.46+000
200.0	2.36-008	200.0	3.35-006	1.41+000
250.0	2.53-008	250.0	4.55-006	1.39+000
300.0	2.54-008	300.0	5.60-006	1.36+000

KANNULUIK(1933), MO- K(1)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
273.1	2.63-008	0.0	2.63-008
194.6	2.38-008	-78.5	2.38-008
90.1	1.94-008	-183.0	1.94-008

KANNULUIK(1933), MO- K(2)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
273.1	2.61-008	0.0	2.61-008
194.6	2.35-008	-78.5	2.35-008
90.1	1.76-008	-183.0	1.76-008

Chromium

GOFF (1970), CR-G

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
2.0	2.44-008	2.0	2.44-008	2.0	2.44-008
5.0	2.44-008	5.0	2.44-008	5.0	2.44-008
10.0	2.47-008	10.0	2.47-008	10.0	2.47-008
20.0	2.47-008	20.0	2.47-008	20.0	2.47-008
50.0	1.90-008	50.0	1.90-008	50.0	1.90-008
100.0	2.75-008	100.0	2.75-008	100.0	2.75-008
200.0	4.05-008	200.0	4.05-008	200.0	4.05-008

HARPER, KEMP, KLEMENS, TAINSH, AND WHITE (1957), CR-HKKTW

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
2.0	2.44-008	2.0	2.44-008	2.0	2.44-008
5.0	2.44-008	5.0	2.44-008	5.0	2.44-008
10.0	2.47-008	10.0	2.47-008	10.0	2.47-008
20.0	2.47-008	20.0	2.47-008	20.0	2.47-008
50.0	1.90-008	50.0	1.90-008	50.0	1.90-008
100.0	2.75-008	100.0	2.75-008	100.0	2.75-008
200.0	4.05-008	200.0	4.05-008	200.0	4.05-008

HARPER, KEMP, KLEMENS, TAINSH, AND WHITE (1957), CR-HKKTW

ELECTRICAL RESISTIVITY		THERMAL CONDUCTIVITY		ELECTRICAL RESISTIVITY	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.0	2.45-007	5.0	1.66-008	5.0	2.36-008
10.0	2.32-008	10.0	1.92-007	10.0	2.28-008
15.0	2.27-008	15.0	1.85-008	15.0	2.00-008
20.0	2.24-008	20.0	1.82-008	20.0	1.69-008
25.0	2.21-008	25.0	1.79-008	25.0	1.49-008
30.0	2.18-008	30.0	1.76-008	30.0	1.24-008
35.0	2.16-008	35.0	1.74-008	35.0	1.04-008
45.0	1.73-008	45.0	1.66-008	45.0	8.30-008
55.0	1.66-008	55.0	1.62-000	55.0	7.10-008
75.0	1.83-008	75.0	1.55-007	75.0	4.42-008
100.0	2.49-008	100.0	1.75-006	100.0	1.55-006
150.0	4.12-008	150.0	4.75-006	150.0	3.98-008

HARPER, KEMP, KLEMENS, TAINSH, AND WHITE (1957), CR-HKKTW

ELECTRICAL RESISTIVITY		THERMAL CONDUCTIVITY		ELECTRICAL RESISTIVITY	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.0	2.45-008	5.0	1.66-008	5.0	2.36-008
15.0	1.92-008	15.0	1.66-007	15.0	1.96-008
20.0	1.73-008	20.0	1.46-000	20.0	1.78-008
25.0	1.66-008	25.0	1.36-000	25.0	1.81-008
30.0	1.55-008	30.0	1.26-000	30.0	1.86-008
35.0	1.60-008	35.0	1.17-000	35.0	1.91-008
45.0	1.44-008	45.0	1.08-000	45.0	2.11-008
55.0	1.42-008	55.0	9.01-007	55.0	2.22-000
75.0	1.73-008	75.0	8.91-007	75.0	2.24-000
100.0	2.39-008	100.0	8.68-006	100.0	2.24-000
150.0	4.06-008	150.0	4.68-006	150.0	3.04-000

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.0	2.38-008	5.0	1.25-007	5.0	1.25-007
15.0	2.28-008	15.0	1.28-007	15.0	1.28-007
20.0	2.18-008	20.0	1.32-007	20.0	1.32-007
25.0	2.07-008	25.0	1.42-007	25.0	1.42-007
30.0	1.95-008	30.0	1.54-007	30.0	1.54-007
35.0	1.91-008	35.0	1.70-007	35.0	1.70-007
45.0	1.62-008	45.0	2.35-007	45.0	2.35-007
55.0	1.63-008	55.0	2.45-007	55.0	2.45-007
75.0	1.78-008	75.0	1.62-006	75.0	1.62-006
100.0	2.08-008	100.0	1.52-006	100.0	1.52-006
150.0	2.00-008	150.0	1.50-006	150.0	1.50-006

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.0	2.36-008	5.0	9.00-008	5.0	9.00-008
15.0	2.28-008	15.0	9.30-008	15.0	9.30-008
20.0	2.00-008	20.0	9.70-008	20.0	9.70-008
25.0	1.69-008	25.0	1.07-007	25.0	1.07-007
30.0	1.49-008	30.0	1.19-007	30.0	1.19-007
35.0	1.44-008	35.0	1.35-000	35.0	1.35-000
45.0	1.46-008	45.0	3.98-000	45.0	3.98-000
55.0	1.47-008	55.0	3.10-007	55.0	3.10-007
75.0	1.76-008	75.0	7.10-007	75.0	7.10-007
100.0	2.42-008	100.0	1.07-007	100.0	1.07-007
150.0	3.98-008	150.0	4.59-006	150.0	4.59-006

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.0	2.20-008	5.0	5.50-008	5.0	5.50-008
15.0	1.96-008	15.0	5.80-008	15.0	5.80-008
20.0	1.78-008	20.0	6.20-008	20.0	6.20-008
25.0	1.68-008	25.0	7.20-008	25.0	7.20-008
30.0	1.55-008	30.0	8.40-008	30.0	8.40-008
35.0	1.43-008	35.0	9.52-000	35.0	9.52-000
45.0	1.36-008	45.0	1.00-007	45.0	1.00-007
55.0	1.40-008	55.0	1.65-007	55.0	1.65-007
75.0	1.73-008	75.0	2.75-007	75.0	2.75-007
100.0	2.46-008	100.0	4.13-008	100.0	4.13-008
150.0	4.13-008	150.0	4.56-006	150.0	4.56-006

Chromium (cont.)

MOORE, WILLIAMS, AND MCELROY(1968), CR- MWM(68)

	LORENZ TEMP	TEMP	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
	RATIO			
90.0	2.49-008	90.0	1.49-008	1.48-002
100.0	2.71-008	100.0	1.93-008	1.40-002
110.0	2.91-008	110.0	2.41-008	1.34-002
120.0	3.16-008	120.0	2.93-008	1.29-002
130.0	3.39-008	130.0	3.52-008	1.25-002
140.0	3.62-008	140.0	4.19-008	1.21-002
150.0	3.81-008	150.0	4.87-008	1.17-002
160.0	3.93-008	160.0	5.51-008	1.14-002
170.0	4.02-008	170.0	6.17-008	1.11-002
180.0	4.09-008	180.0	6.83-008	1.08-002
190.0	4.13-008	190.0	7.48-008	1.05-002
200.0	4.11-008	200.0	8.13-008	1.02-002
210.0	4.21-008	210.0	8.80-008	1.00-002
220.0	4.23-008	220.0	9.46-009	9.83-001
320.0	3.05-008	320.0	1.01-007	9.64-001
240.0	4.24-008	240.0	1.08-007	9.47-001
250.0	4.22-008	250.0	1.13-007	9.32-001
260.0	4.20-008	260.0	1.19-007	9.17-001
270.0	4.15-008	270.0	1.24-007	9.05-001
280.0	4.10-008	280.0	1.28-007	8.93-001
290.0	4.02-008	290.0	1.32-007	8.82-001
300.0	3.93-008	300.0	1.35-007	8.75-001

MOORE, WILLIAMS, AND MCELROY(1967), CR- MWM(67)

	LORENZ TEMP	TEMP	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
	RATIO			
90.0	2.31-008	90.0	1.23-008	1.70-002
100.0	2.60-008	100.0	1.63-008	1.59-002
120.0	3.12-008	120.0	2.60-008	1.44-002
140.0	3.56-008	140.0	3.76-008	1.33-002
160.0	3.88-008	160.0	5.00-008	1.24-002
180.0	4.11-008	180.0	6.31-008	1.17-002
200.0	4.19-008	200.0	7.55-008	1.11-002
220.0	4.24-008	220.0	8.79-008	1.06-002
240.0	4.25-008	240.0	1.00-007	1.02-002
260.0	4.20-008	260.0	1.11-007	9.84-001
280.0	4.11-008	280.0	1.20-007	9.57-001
300.0	3.96-008	300.0	1.27-007	9.35-001

MOORE, WILLIAMS, AND MCELROY(1967), CR- MWM(67)

	LORENZ TEMP	TEMP	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
	RATIO			
90.0	2.44-008	90.0	1.45-008	1.52-002
100.0	2.70-008	100.0	1.86-008	1.45-002
120.0	3.20-008	120.0	2.86-008	1.34-002
140.0	3.64-008	140.0	4.05-008	1.26-002
160.0	3.92-008	160.0	5.30-008	1.19-002
180.0	4.10-008	180.0	6.57-008	1.12-002
200.0	4.17-008	200.0	7.83-008	1.07-002
220.0	4.22-008	220.0	9.10-008	1.02-002
240.0	4.22-008	240.0	1.03-007	9.83-001
260.0	4.18-008	260.0	1.14-007	9.54-001
280.0	4.08-008	280.0	1.23-007	9.30-001
300.0	3.93-008	300.0	1.29-007	9.15-001

Manganese and Rhenium

REEDMANN(1935), MN AND RE- R

	LORENZ RATIO	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY
TEMP	79•6	4.85-008	10•0	3.03-005	1.60-002	10•0	2.91-008	10•0	4.69-007
LORENZ RATIO	83•2	1.03-007	20•0	8.60-005	2.40-002	15•0	2.72-008	15•0	4.69-007
91•5	6.39-008					20•0	2.77-008	20•0	4.85-007

WHITE AND WOODS (1957C), MN AND RE - WW(MN)

	LORENZ RATIO	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY
TEMP	5•0	3.18-008	5•0	1.57-005	1.00-002	10•0	4.00-002	10•0	4.00-002
LORENZ RATIO	10•0	4.85-008	10•0	3.03-005	1.60-002	15•0	2.91-008	15•0	4.69-007
20•0	2.03-007	20•0	8.60-005	2.40-002	3.00-004	3.00-002	1.95-008	1.95-008	1.39-008
30•0	1.11-007	30•0	1.11-004	1.30-004	3.00-002	4.00-002	1.20-008	1.20-008	1.39-008
40•0	1.14-007	40•0	1.30-004	1.30-004	4.00-002	5.00-002	1.29-008	1.29-008	2.99-008
50•0	1.25-007	50•0	1.56-004	1.56-004	4.50-002	5.50-002	3.00-008	3.00-008	1.44-007
60•0	1.17-007	60•0	1.56-004	1.56-004	5.00-002	5.50-002	4.00-008	4.00-008	4.23-007
70•0	1.11-007	70•0	1.56-004	1.56-004	5.50-002	5.50-002	4.50-008	4.50-008	7.63-007
80•0	1.07-007	80•0	1.56-004	1.56-004	5.50-002	5.50-002	5.00-008	5.00-008	9.50-001

POWELL, TYE, AND WOODMAN(1963), RE-PTW(RE)

	LORENZ RATIO	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY
TEMP	83•0	2.17-008	123•0	2.64-008	173•0	2.44-008	173•0	2.17-008	83•0
LORENZ RATIO	123•0	2.64-008	173•0	2.44-008	223•0	3.04-008	223•0	2.64-008	123•0
223•0	3.04-008					223•0	3.04-008	223•0	2.64-008
273•0	3.09-008					273•0	3.09-008	273•0	2.64-008
293•0	3.08-008					293•0	3.08-008	293•0	2.64-008

WHITE AND WOODS (1957B), MN AND RE - WW(RE1)

	LORENZ RATIO	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY
TEMP	10•0	3.07-008	10•0	1.07-007	7.87-007	10•0	2.17-008	10•0	3.30-001
LORENZ RATIO	15•0	2.78-008	15•0	7.87-007	8.03-007	15•0	2.64-008	15•0	5.30-001
20•0	2.69-008	20•0	7.87-007	7.87-007	9.17-007	20•0	6.70-001	20•0	6.70-001
30•0	2.59-008	30•0	7.87-007	7.87-007	7.87-007	30•0	7.50-001	30•0	7.50-001
40•0	2.45-008	40•0	7.87-007	7.87-007	7.87-007	40•0	6.20-001	40•0	6.20-001
50•0	2.19-008	50•0	7.87-007	7.87-007	7.87-007	50•0	5.30-001	50•0	5.30-001
75•0	2.15-008	75•0	7.87-007	7.87-007	7.87-007	75•0	4.79-001	75•0	4.79-001
100•0	2.04-008	100•0	7.87-007	7.87-007	7.87-007	100•0	4.10-001	100•0	4.10-001

WHITE AND WOODS (1957B), MN AND RE - WW(RE2)

	LORENZ RATIO	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY	TEMP	RESISTIVITY
TEMP	79•6	6.41-008	79•6	6.41-008	79•6	6.41-008	10•0	10•0	6.00-001
LORENZ RATIO	83•2	6.41-008	83•2	6.41-008	83•2	6.41-008	15•0	15•0	4.69-007
91•5	6.39-008						20•0	20•0	8.70-001
		91•5	6.39-008				30•0	30•0	1.06+000
							40•0	40•0	5.99-007
							50•0	50•0	1.02+000
							75•0	75•0	9.00-001
							100•0	100•0	7.50-001
								100•0	6.49-006
									6.70-001
									6.00-001

Iron

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	2•49-008	100•0	3•50-006	7•10-001
150•0	2•71-008	150•0	5•50-006	7•40-001
200•0	2•74-008	200•0	7•50-006	7•30-001
250•0	2•44-008	250•0	1•00-005	7•10-001
280•0	2•90-008	280•0	1•16-005	7•00-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	2•31-008	100•0	3•80-006	6•60-001
150•0	2•75-008	150•0	5•90-006	7•00-001
200•0	2•78-008	200•0	8•00-006	6•95-001
250•0	2•93-008	250•0	1•06-005	6•90-001
280•0	2•98-008	280•0	1•21-005	6•90-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	2•31-008	100•0	3•80-006	6•60-001
150•0	2•75-008	150•0	5•90-006	7•00-001
200•0	2•78-008	200•0	8•00-006	6•95-001
250•0	2•93-008	250•0	1•06-005	6•90-001
280•0	2•98-008	280•0	1•21-005	6•90-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	2•44-008	100•0	5•80-006	4•90-001
150•0	2•99-008	150•0	8•00-006	5•60-001
200•0	3•44-008	200•0	1•03-005	5•90-001
250•0	3•07-008	250•0	1•30-005	5•90-001
280•0	3•11-008	280•0	1•45-005	6•00-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	3•44-008	100•0	8•70-006	3•95-001
150•0	3•46-008	150•0	1•08-005	4•80-001
200•0	3•42-008	200•0	1•30-005	5•10-001
250•0	3•35-008	250•0	1•55-005	5•40-001
280•0	3•34-008	280•0	1•70-005	5•50-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	4•51-008	100•0	4•29-008	100•0
150•0	5•00-008	150•0	5•37-008	150•0
200•0	5•97-008	200•0	6•81-008	200•0
250•0	3•81-008	250•0	3•73-008	250•0
280•0	3•73-008	280•0	2•86-005	3•65-001

BACKLUNO(1961), FE-B

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity
100•0	1•64-008	100•0	1•10-006	1•17-000
150•0	2•13-008	150•0	3•30-006	9•70-001
200•0	2•43-008	200•0	5•40-006	9•00-001
250•0	2•62-008	250•0	7•80-006	8•40-001
280•0	2•69-008	280•0	9•30-006	8•10-001

HUSTI, POWELL, AND WEITZEL (1970), FE - HPW(2)

HUSTI, POWELL, AND WEITZEL (1970), FE - HPW(2A)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
6.0	2.50-008	6.0	2.50-008	6.0	2.52-008	6.0	2.52-008	6.0
7.0	2.51-008	7.0	2.51-008	7.0	2.52-008	7.0	2.53-008	7.0
8.0	2.49-008	8.0	2.49-008	8.0	2.50-008	8.0	2.54-008	8.0
9.0	2.49-008	9.0	2.49-008	9.0	2.50-008	9.0	2.55-008	9.0
10.0	2.50-008	10.0	2.50-008	10.0	2.50-008	10.0	2.54-008	10.0
12.0	2.50-008	12.0	2.50-008	12.0	2.50-008	12.0	2.54-008	12.0
14.0	2.50-008	14.0	2.50-008	14.0	2.50-008	14.0	2.54-008	14.0
16.0	2.48-008	16.0	2.48-008	16.0	2.50-008	16.0	2.54-008	16.0
18.0	2.46-008	18.0	2.46-008	18.0	2.46-008	18.0	2.54-008	18.0
20.0	2.44-008	20.0	2.44-008	20.0	2.44-008	20.0	2.54-008	20.0
25.0	2.36-008	25.0	2.36-008	25.0	2.36-008	25.0	2.49-008	25.0
30.0	2.25-008	30.0	2.25-008	30.0	2.40-008	30.0	2.49-008	30.0
35.0	2.13-008	35.0	2.13-008	35.0	2.31-008	35.0	2.31-008	35.0
40.0	2.01-008	40.0	2.01-008	40.0	2.21-008	40.0	2.21-008	40.0
45.0	1.91-008	45.0	1.91-008	45.0	2.11-008	45.0	2.11-008	45.0
50.0	1.83-008	50.0	1.83-008	50.0	2.04-008	50.0	2.11-008	50.0
55.0	1.77-008	55.0	1.77-008	55.0	1.98-008	55.0	2.04-008	55.0
60.0	1.74-008	60.0	1.74-008	60.0	1.94-008	60.0	1.98-008	60.0
65.0	1.72-008	65.0	1.72-008	65.0	1.92-008	65.0	1.92-008	65.0
70.0	1.72-008	70.0	1.72-008	70.0	1.91-008	70.0	1.91-008	70.0
75.0	1.73-008	75.0	1.73-008	75.0	1.91-008	75.0	1.91-008	75.0
80.0	1.75-008	80.0	1.75-008	80.0	1.91-008	80.0	1.91-008	80.0
85.0	1.78-008	85.0	1.78-008	85.0	1.93-008	85.0	1.93-008	85.0
90.0	1.81-008	90.0	1.81-008	90.0	1.95-008	90.0	1.95-008	90.0
95.0	1.84-008	95.0	1.84-008	95.0	1.97-008	95.0	1.97-008	95.0
100.0	1.88-008	100.0	1.88-008	100.0	1.99-008	100.0	1.99-008	100.0
110.0	1.96-008	110.0	1.96-008	110.0	2.05-008	110.0	2.05-008	110.0
120.0	2.04-008	120.0	2.04-008	120.0	2.10-008	120.0	2.10-008	120.0
130.0	2.12-008	130.0	2.12-008	130.0	2.16-008	130.0	2.16-008	130.0
140.0	2.19-008	140.0	2.19-008	140.0	2.22-008	140.0	2.22-008	140.0
150.0	2.26-008	150.0	2.26-008	150.0	2.28-008	150.0	2.28-008	150.0
160.0	2.32-008	160.0	2.32-008	160.0	2.33-008	160.0	2.33-008	160.0
170.0	2.37-008	170.0	2.37-008	170.0	2.38-008	170.0	2.38-008	170.0
180.0	2.42-008	180.0	2.42-008	180.0	2.42-008	180.0	2.42-008	180.0
190.0	2.46-008	190.0	2.46-008	190.0	2.46-008	190.0	2.46-008	190.0
200.0	2.49-008	200.0	2.49-008	200.0	2.50-008	200.0	2.50-008	200.0
220.0	2.56-008	220.0	2.56-008	220.0	2.56-008	220.0	2.56-008	220.0
240.0	2.63-008	240.0	2.63-008	240.0	2.60-008	240.0	2.60-008	240.0
260.0	2.69-008	260.0	2.69-008	260.0	2.64-008	260.0	2.64-008	260.0
280.0	2.74-008	280.0	2.74-008	280.0	2.68-008	280.0	2.68-008	280.0

Iron (cont.)

HUST, POWELL, AND WEITZEL (1970), FE - HPW(4)

HUST AND SPARKS(1970A), FE - HS

LORENZ		LORENZ		LORENZ		LORENZ	
TEMP	RATIO	TEMP	RATIO	TEMP	RATIO	TEMP	RATIO
6.0	2.49-008	6.0	2.44-008	6.0	2.51-008	6.0	2.51-008
7.0	2.52-008	7.0	2.52-008	7.0	2.52-008	7.0	2.52-008
8.0	2.52-008	8.0	2.52-008	8.0	2.52-008	8.0	2.52-008
9.0	2.52-008	9.0	2.52-008	9.0	2.52-008	9.0	2.52-008
10.0	2.52-008	10.0	2.52-008	10.0	2.52-008	10.0	2.52-008
12.0	2.51-008	12.0	2.51-008	12.0	2.53-008	12.0	2.53-008
14.0	2.51-008	14.0	2.51-008	14.0	2.53-008	14.0	2.53-008
16.0	2.50-008	16.0	2.50-008	16.0	2.53-008	16.0	2.53-008
18.0	2.50-008	18.0	2.50-008	18.0	2.53-008	18.0	2.53-008
20.0	2.49-008	20.0	2.49-008	20.0	2.52-008	20.0	2.52-008
25.0	2.45-008	25.0	2.44-008	25.0	2.48-008	25.0	2.48-008
30.0	2.46-008	30.0	2.46-008	30.0	2.49-008	30.0	2.49-008
35.0	2.45-008	35.0	2.45-008	35.0	2.50-008	35.0	2.50-008
40.0	2.44-008	40.0	2.44-008	40.0	2.50-008	40.0	2.50-008
45.0	2.45-008	45.0	2.45-008	45.0	2.51-008	45.0	2.51-008
50.0	2.47-008	50.0	2.47-008	50.0	2.54-008	50.0	2.54-008
55.0	2.42-008	55.0	2.42-008	55.0	2.58-008	55.0	2.58-008
60.0	2.48-008	60.0	2.48-008	60.0	1.95-008	60.0	1.95-008
65.0	2.46-008	65.0	2.46-008	65.0	1.92-008	65.0	1.92-008
70.0	2.45-008	70.0	2.45-008	70.0	1.91-008	70.0	1.91-008
75.0	2.46-008	75.0	2.46-008	75.0	1.91-008	75.0	1.91-008
80.0	2.47-008	80.0	2.47-008	80.0	1.92-008	80.0	1.92-008
85.0	2.48-008	85.0	2.48-008	85.0	1.93-008	85.0	1.93-008
90.0	2.49-008	90.0	2.49-008	90.0	1.95-008	90.0	1.95-008
95.0	2.52-008	95.0	2.52-008	95.0	1.97-008	95.0	1.97-008
100.0	2.52-008	100.0	2.52-008	100.0	1.99-008	100.0	1.99-008
110.0	2.00-008	110.0	2.00-008	110.0	2.04-008	110.0	2.04-008
120.0	2.06-008	120.0	2.06-008	120.0	2.10-008	120.0	2.10-008
130.0	2.11-008	130.0	2.12-008	130.0	2.16-008	130.0	2.16-008
140.0	2.18-008	140.0	2.18-008	140.0	2.22-008	140.0	2.22-008
150.0	2.24-008	150.0	2.24-008	150.0	2.28-008	150.0	2.28-008
160.0	2.29-008	160.0	2.29-008	160.0	2.33-008	160.0	2.33-008
170.0	2.34-008	170.0	2.34-008	170.0	2.38-008	170.0	2.38-008
180.0	2.39-008	180.0	2.39-008	180.0	2.43-008	180.0	2.43-008
190.0	2.44-008	190.0	2.44-008	190.0	2.47-008	190.0	2.47-008
200.0	2.48-008	200.0	2.48-008	200.0	2.51-008	200.0	2.51-008
220.0	2.55-008	220.0	2.55-008	220.0	2.58-008	220.0	2.58-008
240.0	2.60-008	240.0	2.60-008	240.0	2.62-008	240.0	2.62-008
260.0	2.65-008	260.0	2.65-008	260.0	2.65-008	260.0	2.65-008
280.0	2.68-008	280.0	2.68-008	280.0	2.68-008	280.0	2.68-008
300.0	2.71-008	300.0	2.71-008	300.0	2.71-008	300.0	2.71-008

KANNULUIK(1933), FE-KK

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.47-008	0.0	2.47-008	0.0
273.1	2.11-008	-78.5	2.11-008	-78.5
194.6	2.11-008	90.1	2.10-008	-183.0
90.1	1.20-008			1.60-008

KARWEIL AND SCHAFER(1939), FE-KS

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.70-008	5.0	2.70-008	5.0
10.0	2.40-008	10.0	2.40-008	10.0
15.0	2.40-008	15.0	2.40-008	15.0
20.0	2.40-008	20.0	2.40-008	20.0

KEMP, KLEMENS, AND WHITE(1956), FE-KK

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.49-008	5.0	2.49-008	5.0
15.0	2.53-008	15.0	2.53-008	15.0
25.0	2.40-008	25.0	2.40-008	25.0
35.0	2.04-008	35.0	2.04-008	35.0
45.0	1.71-008	45.0	1.71-008	45.0
55.0	1.52-008	55.0	1.52-008	55.0
65.0	1.51-008	65.0	1.51-008	65.0
75.0	1.57-008	75.0	1.57-008	75.0
90.0	1.70-008	90.0	1.70-008	90.0
115.0	1.98-008	115.0	1.98-008	115.0
130.0	2.16-008	130.0	2.16-008	130.0

KEMP, KLEMENS, AND TAINSH(1959), FE-KKT

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.53-008	4.0	2.53-008	4.0
10.0	2.49-008	10.0	2.49-008	10.0
20.0	2.40-008	20.0	2.40-008	20.0
26.0	1.95-008	26.0	1.95-008	26.0
32.0	1.62-008	32.0	1.62-008	32.0
40.0	1.33-008	40.0	1.33-008	40.0
50.0	1.11-008	50.0	1.11-008	50.0
60.0	1.26-008	60.0	1.26-008	60.0
70.0	1.31-008	70.0	1.31-008	70.0
90.0	1.50-008	90.0	1.50-008	90.0

KOLHAAS AND KIERSPE(1965), FE-KK

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.47-008	0.0	2.47-008	0.0
273.1	2.11-008	-78.5	2.11-008	-78.5
194.6	2.11-008	90.1	2.10-008	-183.0
90.1	1.20-008			1.60-008

LEES(1908), FE-L

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.70-008	5.0	2.70-008	5.0
10.0	2.40-008	10.0	2.40-008	10.0
15.0	2.40-008	15.0	2.40-008	15.0
20.0	2.40-008	20.0	2.40-008	20.0

MOORE, MCELROY, AND BARSONI(1966), FE-MMB

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.50-008	3.0	2.50-008	3.0
5.0	3.30-008	5.0	3.30-008	5.0
10.0	3.30-008	10.0	3.30-008	10.0
15.0	4.30-008	15.0	4.30-008	15.0
20.0	5.00-008	20.0	5.00-008	20.0

KOLHAAS AND KIERSPE(1965), FE-KK

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
TEMP	2.22-008	88.1	2.22-008	90.1
90.1	2.23-008	106.1	2.23-008	106.1
106.1	2.33-008	123.1	2.33-008	123.1
123.1	2.35-008	137.1	2.35-008	137.1
137.1	2.39-008	167.1	2.39-008	167.1
167.1	2.48-008	203.1	2.48-008	203.1
203.1	2.52-008	225.1	2.52-008	225.1
225.1	2.68-008	248.0	2.68-008	248.0
248.0	2.94-008	273.0	2.94-008	273.0
273.0	2.97-008	291.0	2.97-008	291.0
291.0	2.99-008			2.99-008

Stainless and Alloy Steels

BERMAN(1951), ALLOY AND STAINLESS STEELS -B

	LORENZ RATIO	ELECTRICAL RESISTIVITY	THERMAL CONDUCTIVITY
TEMP	3•12-008	4•40	2•63-003
4•00	4•18-008	10•0	4•80-005
10•0	5•04-008	20•0	4•92-005
20•0	5•27-008	50•0	5•03-005
50•0	5•40-008	81•2	5•22-005

ESTERMANN AND ZIMMERMANN(1952), ALLOY AND STAINLESS STEELS-E2(303)

	LORENZ RATIO	TEMP	TEMP
TEMP	3•00-008	4•2	3•00-008
4•2	3•30-008	10•1	3•30-008
10•1	4•90-008	20•4	4•90-008
20•4	5•40-008	20•6	5•40-008
20•6	5•90-008	56•9	5•90-008
56•9	5•80-008	58•0	5•80-008
58•0	5•60-008	59•7	5•60-008
59•7	5•70-008	63•3	5•70-008
63•3	66•8	66•8	5•50-008
66•8	77•0	77•0	5•60-008
77•0	77•6	77•6	2•50-008
77•6	4•3	4•3	2•85-008
4•3	10•1	10•1	3•40-008
10•1	19•4	19•4	5•20-008
19•4	58•3	58•3	5•70-008
58•3	59•5	59•5	5•70-008
59•5	77•6	77•8	5•40-008

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(NI)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	125•0	125•0
125•0	3•82-008	3•82-008	3•82-008
3•82-008	200•0	3•40-008	3•40-008
3•40-008	300•0	3•25-008	3•25-008

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(12315)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	3•96-008	3•96-008
3•96-008	200•0	3•52-008	3•52-008
3•52-008	300•0	3•34-008	3•34-008

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(4340)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	4•04-008	4•04-008
4•04-008	200•0	3•55-008	3•55-008
3•55-008	300•0	3•32-008	3•32-008

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(12515)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	4•16-008	4•16-008
4•16-008	200•0	3•65-008	3•65-008
3•65-008	300•0	3•40-008	3•40-008

ESTERMANN AND ZIMMERMANN(1952), ALLOY AND STAINLESS STEELS-E2(347)

	LORENZ RATIO	TEMP	TEMP
TEMP	4•3	2•80-008	4•3
4•3	14•0	4•30-008	14•0
14•0	58•5	5•60-008	58•5
58•5	63•2	5•40-008	63•2
63•2	63•3	5•50-008	63•3
63•3	70•7	5•40-008	70•7
70•7	76•2	5•30-008	76•2

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(1015)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	3•77-008	3•77-008
3•77-008	200•0	3•45-008	3•45-008
3•45-008	300•0	3•30-008	3•30-008

FLYNN(1971), ALLOY AND STAINLESS STEELS - F(HP49)

	LORENZ RATIO	TEMP	TEMP
TEMP	125•0	200•0	200•0
200•0	3•21-008	3•21-008	3•21-008
3•21-008	300•0	3•11-008	3•11-008

Stainless and Alloy Steels (cont.)

FLYNN(1971), ALLOY AND STAINLESS STEELS - F (HMB0)

	LORENZ RATIO	TEMP 3•20-008	LORENZ RATIO	TEMP 3•40-008		LORENZ RATIO	TEMP 5•74-008		LORENZ RATIO	TEMP 6•0	
12•0		125•0		200•0	3•40-008		7•0		5•83-008	7•0	5•83-008
200•0	3•30-008		200•0	3•40-008		8•0		5•95-008	8•0	5•96-008	
300•0	3•31-008		300•0	3•14-008		9•0		6•10-008	9•0	6•10-008	
						10•0		6•24-008	10•0	6•24-008	
						12•0		6•50-008	12•0	6•50-008	
						14•0		6•71-008	14•0	6•71-008	
						16•0		6•89-008	16•0	6•89-008	
						18•0		7•03-008	18•0	7•03-008	
						20•0		7•14-008	20•0	7•14-008	
						25•0		7•32-008	25•0	7•32-008	
						30•0		7•39-008	30•0	7•39-008	
						35•0		7•40-008	35•0	7•40-008	
						40•0		7•37-008	40•0	7•37-008	
						45•0		7•30-008	45•0	7•30-008	
						50•0		7•20-008	50•0	7•20-008	
						55•0		7•09-008	55•0	7•09-008	
						60•0		6•96-008	60•0	6•96-008	
						65•0		6•83-008	65•0	6•83-008	
						70•0		6•70-008	70•0	6•70-008	
						75•0		6•56-008	75•0	6•56-008	
						80•0		6•42-008	80•0	6•42-008	
						85•0		6•29-008	85•0	6•29-008	
						90•0		6•16-008	90•0	6•16-008	
						95•0		6•04-008	95•0	6•04-008	
						100•0		5•91-008	100•0	5•91-008	
						110•0		5•69-008	110•0	5•69-008	
						120•0		5•48-008	120•0	5•48-008	
						130•0		5•29-008	130•0	5•29-008	
						140•0		5•12-008	140•0	5•12-008	
						150•0		4•96-008	150•0	4•96-008	
						160•0		4•83-008	160•0	4•83-008	
						170•0		4•70-008	170•0	4•70-008	
						180•0		4•59-008	180•0	4•59-008	
						190•0		4•49-008	190•0	4•49-008	
						200•0		4•40-008	200•0	4•40-008	
						220•0		4•24-008	220•0	4•24-008	
						240•0		4•11-008	240•0	4•11-008	
						260•0		4•00-008	260•0	4•00-008	
						280•0		3•89-008	280•0	3•89-008	

HUST AND SPARKS(1971A), ALLOY AND STAINLESS STEELS - HS(1)

FLYNN(1971), ALLOY AND STAINLESS STEELS - F (FLINAR)

	LORENZ RATIO	TEMP 3•70-008	LORENZ RATIO	TEMP 4•29-008		LORENZ RATIO	TEMP 5•74-008		LORENZ RATIO	TEMP 6•0	
12•0		125•0		200•0	3•90-008		120•0		5•69-008	120•0	5•69-008
200•0	3•80-008		200•0	3•90-008		130•0		5•48-008	130•0	5•48-008	
300•0	3•75-008		300•0	3•75-008		140•0		5•12-008	140•0	5•12-008	
						150•0		4•96-008	150•0	4•96-008	
						160•0		4•83-008	160•0	4•83-008	
						170•0		4•70-008	170•0	4•70-008	
						180•0		4•59-008	180•0	4•59-008	
						190•0		4•49-008	190•0	4•49-008	
						200•0		4•40-008	200•0	4•40-008	
						220•0		4•24-008	220•0	4•24-008	
						240•0		4•11-008	240•0	4•11-008	
						260•0		4•00-008	260•0	4•00-008	
						280•0		3•89-008	280•0	3•89-008	

FLYNN(1971), ALLOY AND STAINLESS STEELS - F (FCI)

	LORENZ RATIO	TEMP 3•49-008	LORENZ RATIO	TEMP 4•29-008		LORENZ RATIO	TEMP 5•74-008		LORENZ RATIO	TEMP 6•0	
12•0		125•0		200•0	3•90-008		110•0		5•91-008	100•0	5•91-008
200•0	3•46-008		200•0	3•46-008		120•0		5•69-008	110•0	5•69-008	
300•0	3•49-008		300•0	3•49-008		130•0		5•48-008	120•0	5•48-008	
						140•0		5•12-008	130•0	5•12-008	
						150•0		4•96-008	140•0	4•96-008	
						160•0		4•83-008	150•0	4•83-008	
						170•0		4•70-008	160•0	4•70-008	
						180•0		4•59-008	170•0	4•59-008	
						190•0		4•49-008	180•0	4•49-008	
						200•0		4•40-008	190•0	4•40-008	
						220•0		4•24-008	200•0	4•24-008	
						240•0		4•11-008	220•0	4•11-008	
						260•0		4•00-008	240•0	4•00-008	
						280•0		3•89-008	260•0	3•89-008	

Stainless and Alloy Steels (cont.)

HUST AND SPARKS(1971A), ALLOY AND STAINLESS STEELS - HS(WH)

HUST AND SPARKS(1971E), ALLOY AND STAINLESS STEELS - HS(12)

LORENZ		LORENZ		LORENZ	
TEMP	RATIO	TEMP	RATIO	TEMP	RATIO
6.0	6.79-008	6.0	6.79-008	5.0	5.34-008
6.0	6.99-008	7.0	6.99-008	6.0	5.43-008
8.0	7.20-008	8.0	7.40-008	7.0	5.60-008
8.0	7.41-008	9.0	7.41-008	8.0	5.80-008
10.0	7.61-008	10.0	7.61-008	9.0	5.99-008
12.0	7.92-008	12.0	7.92-008	10.0	6.16-008
14.0	8.14-008	14.0	8.14-008	12.0	6.45-008
16.0	8.29-008	16.0	8.29-008	14.0	6.68-008
18.0	8.38-008	18.0	8.38-008	16.0	6.85-008
20.0	8.44-008	20.0	8.44-008	18.0	6.99-008
25.0	8.46-008	25.0	8.46-008	20.0	7.09-008
30.0	8.39-008	30.0	8.39-008	25.0	7.26-008
35.0	8.27-008	35.0	8.27-008	30.0	7.33-008
40.0	8.12-008	40.0	8.12-008	35.0	7.34-008
45.0	7.95-008	45.0	7.95-008	40.0	7.31-008
50.0	7.77-008	50.0	7.77-008	45.0	7.24-008
55.0	7.59-008	55.0	7.59-008	50.0	7.15-008
60.0	7.41-008	60.0	7.41-008	55.0	7.04-008
65.0	7.24-008	65.0	7.24-008	60.0	6.92-008
70.0	7.06-008	70.0	7.06-008	65.0	6.80-008
75.0	6.89-008	75.0	6.89-008	70.0	6.67-008
80.0	6.73-008	80.0	6.73-008	75.0	6.53-008
85.0	6.57-008	85.0	6.57-008	80.0	6.44-008
90.0	6.41-008	90.0	6.41-008	85.0	6.27-008
95.0	6.27-008	95.0	6.27-008	90.0	6.14-008
100.0	6.13-008	100.0	6.13-008	95.0	6.01-008
110.0	5.87-008	110.0	5.87-008	100.0	5.89-008
120.0	5.64-008	120.0	5.64-008	110.0	5.66-008
130.0	5.43-008	130.0	5.43-008	120.0	5.45-008
140.0	5.24-008	140.0	5.24-008	130.0	5.26-008
150.0	5.07-008	150.0	5.07-008	140.0	5.10-008
160.0	4.92-008	160.0	4.92-008	150.0	4.94-008
170.0	4.79-008	170.0	4.79-008	160.0	4.81-008
180.0	4.67-008	180.0	4.67-008	170.0	4.69-008
190.0	4.56-008	190.0	4.56-008	180.0	4.58-008
200.0	4.47-008	200.0	4.47-008	190.0	4.49-008
220.0	4.30-008	220.0	4.30-008	200.0	4.40-008
240.0	4.16-008	240.0	4.16-008	220.0	4.26-008
260.0	4.04-008	260.0	4.04-008	240.0	4.13-008
280.0	3.93-008	280.0	3.93-008	260.0	4.02-008
				280.0	3.91-008
					280.0

Stainless and Alloy Steels (cont.)

HUST AND SPARKS(1971B), ALLOY AND STAINLESS STEELS - HS(2886)

HUST AND SPARKS(1971B), ALLOY AND STAINLESS STEELS - HS(2886A)

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
5.71-008	7.0	5.71-008	7.0	6.68-008	5.0	6.68-008	5.0
8.0	8.0	5.87-008	8.0	6.83-008	6.0	6.83-008	6.0
9.0	9.0	6.12-008	9.0	7.08-008	7.0	7.08-008	7.0
10.0	10.0	6.16-008	10.0	7.34-008	8.0	7.34-008	8.0
12.0	12.0	6.37-008	12.0	7.58-008	9.0	7.58-008	9.0
14.0	14.0	6.52-008	14.0	7.80-008	10.0	7.80-008	10.0
16.0	16.0	6.62-008	16.0	8.15-008	12.0	8.15-008	12.0
18.0	18.0	6.69-008	18.0	8.38-008	14.0	8.38-008	14.0
20.0	20.0	6.73-008	20.0	8.54-008	16.0	8.54-008	16.0
25.0	25.0	6.76-008	25.0	8.64-008	18.0	8.64-008	18.0
30.0	30.0	6.73-008	30.0	8.70-008	20.0	8.70-008	20.0
35.0	35.0	6.67-008	35.0	8.72-008	25.0	8.72-008	25.0
40.0	40.0	6.59-008	40.0	8.64-008	30.0	8.64-008	30.0
45.0	45.0	6.49-008	45.0	8.52-008	35.0	8.52-008	35.0
50.0	50.0	6.38-008	50.0	8.26-008	40.0	8.26-008	40.0
55.0	55.0	6.27-008	55.0	8.19-008	45.0	8.19-008	45.0
60.0	60.0	6.15-008	60.0	8.01-008	50.0	8.01-008	50.0
65.0	65.0	6.03-008	65.0	7.82-008	55.0	7.82-008	55.0
70.0	70.0	5.92-008	70.0	7.63-008	60.0	7.63-008	60.0
75.0	75.0	5.80-008	75.0	7.44-008	65.0	7.44-008	65.0
80.0	80.0	5.68-008	80.0	7.26-008	70.0	7.26-008	70.0
85.0	85.0	5.57-008	85.0	7.08-008	75.0	7.08-008	75.0
90.0	90.0	5.46-008	90.0	6.90-008	80.0	6.90-008	80.0
95.0	95.0	5.36-008	95.0	6.74-008	85.0	6.74-008	85.0
100.0	100.0	5.26-008	100.0	6.57-008	90.0	6.57-008	90.0
110.0	110.0	5.07-008	110.0	6.42-008	95.0	6.42-008	95.0
120.0	120.0	4.90-008	120.0	6.27-008	100.0	6.27-008	100.0
130.0	130.0	4.75-008	130.0	6.00-008	110.0	6.00-008	110.0
140.0	140.0	4.61-008	140.0	5.75-008	120.0	5.75-008	120.0
150.0	150.0	4.48-008	150.0	5.53-008	130.0	5.53-008	130.0
160.0	160.0	4.37-008	160.0	5.34-008	140.0	5.34-008	140.0
170.0	170.0	4.27-008	170.0	5.16-008	150.0	5.16-008	150.0
180.0	180.0	4.19-008	180.0	5.01-008	160.0	5.01-008	160.0
190.0	190.0	4.11-008	190.0	4.87-008	170.0	4.87-008	170.0
200.0	200.0	4.04-008	200.0	4.75-008	180.0	4.75-008	180.0
220.0	220.0	3.92-008	220.0	4.64-008	190.0	4.64-008	190.0
240.0	240.0	3.82-008	240.0	4.54-008	200.0	4.54-008	200.0
260.0	260.0	3.73-008	260.0	4.37-008	220.0	4.37-008	220.0
280.0	280.0	3.64-008	280.0	4.22-008	240.0	4.22-008	240.0
				4.10-008	260.0	4.10-008	260.0
				3.99-008	280.0	3.99-008	280.0

Stainless and Alloy Steels (cont.)

HUST AND SPARKS(1971D), ALLOY AND STAINLESS STEELS - HS122)

HUST(1970A), ALLOY AND STAINLESS STEELS - HS347)

LORENZ RATIO		LORENZ RATIO		LORENZ RATIO	
TEMP	TEMP	TEMP	TEMP	TEMP	TEMP
6.0	5.63-008	6.0	5.63-008	6.0	4.66-008
7.0	5.74-008	7.0	5.74-008	7.0	4.58-008
8.0	5.89-008	8.0	5.89-008	8.0	4.74-008
9.0	6.05-008	9.0	6.05-008	9.0	4.89-008
10.0	6.20-008	10.0	6.20-008	10.0	5.04-008
12.0	6.47-008	12.0	6.47-008	12.0	5.30-008
14.0	6.69-008	14.0	6.69-008	14.0	5.52-008
16.0	6.86-008	16.0	6.86-008	16.0	5.69-008
18.0	7.00-008	18.0	7.00-008	18.0	5.82-008
20.0	7.11-008	20.0	7.11-008	20.0	5.93-008
25.0	7.29-008	25.0	7.29-008	25.0	6.12-008
30.0	7.38-008	30.0	7.38-008	30.0	6.22-008
35.0	7.39-008	35.0	7.39-008	35.0	6.26-008
40.0	7.35-008	40.0	7.35-008	40.0	6.26-008
45.0	7.29-008	45.0	7.29-008	45.0	6.14-008
50.0	7.19-008	50.0	7.19-008	50.0	6.19-008
55.0	7.08-008	55.0	7.08-008	55.0	6.12-008
60.0	6.96-008	60.0	6.96-008	60.0	6.04-008
65.0	6.83-008	65.0	6.83-008	65.0	5.96-008
70.0	6.69-008	70.0	6.69-008	70.0	5.87-008
75.0	6.56-008	75.0	6.56-008	75.0	5.77-008
80.0	6.42-008	80.0	6.42-008	80.0	5.68-008
85.0	6.29-008	85.0	6.29-008	85.0	5.58-008
90.0	6.16-008	90.0	6.16-008	90.0	5.48-008
95.0	6.03-008	95.0	6.03-008	95.0	5.39-008
100.0	5.91-008	100.0	5.91-008	100.0	5.30-008
110.0	5.68-008	110.0	5.68-008	110.0	5.13-008
120.0	5.47-008	120.0	5.47-008	120.0	4.97-008
130.0	5.28-008	130.0	5.28-008	130.0	4.82-008
140.0	5.11-008	140.0	5.11-008	140.0	4.69-008
150.0	4.96-008	150.0	4.96-008	150.0	4.56-008
160.0	4.82-008	160.0	4.82-008	160.0	4.46-008
170.0	4.70-008	170.0	4.70-008	170.0	4.36-008
180.0	4.59-008	180.0	4.59-008	180.0	4.27-008
190.0	4.50-008	190.0	4.50-008	190.0	4.19-008
200.0	4.41-008	200.0	4.41-008	200.0	4.12-008
220.0	4.26-008	220.0	4.26-008	220.0	3.99-008
240.0	4.13-008	240.0	4.13-008	240.0	3.88-008
260.0	4.01-008	260.0	4.01-008	260.0	3.78-008
280.0	3.91-008	280.0	3.91-008	280.0	3.68-008

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)					
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal CONDUCTIVITY	
TEMP	6.73-008	90.0	5.41-005	1.12-001	
90.0	5.28-008	14.0	5.43-005	1.31-001	
147.0	4.79-008	191.0	6.48-005	1.48-001	
197.0	5.17-008	237.0	6.33-005	1.82-001	
237.0	5.60-008	294.0	7.32-005	2.25-001	

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)					
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal CONDUCTIVITY	
TEMP	89.0	7.21-008	91.0	5.57-005	1.17-001
91.0	6.05-008	129.0	5.61-005	1.17-001	
129.0	5.67-008	150.0	5.97-005	1.31-001	
150.0	5.47-008	183.0	6.16-005	1.38-001	
183.0	5.42-008	198.0	6.46-005	1.55-001	
198.0	5.39-008	219.0	6.55-005	1.74-001	
219.0	5.46-008	235.0	6.58-005	1.85-001	
235.0	6.03-008	299.0	6.93-005	2.38-001	

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)					
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal CONDUCTIVITY	
TEMP	93.0	92.0	5.69-005	1.14-001	
155.0	5.39-008	155.0	6.23-005	1.34-001	
199.0	5.39-008	237.0	6.62-005	1.52-001	
237.0	5.39-008	293.0	6.64-005	1.84-001	
293.0	5.73-008	293.0	7.53-005	2.23-001	

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)					
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal CONDUCTIVITY	
TEMP	88.0	7.73-008	89.0	6.36-005	1.07-001
147.0	5.87-008	147.0	6.84-005	1.26-001	
198.0	5.47-008	198.0	7.27-005	1.49-001	
233.0	5.62-008	233.0	7.57-005	1.73-001	
295.0	6.17-008	295.0	8.28-005	2.20-001	

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)					
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal CONDUCTIVITY	
TEMP	84.1	4.46-008	189.0	8.44-006	1.51-001
122.1	3.71-008	120.0	9.0-006	1.09-005	
153.1	3.42-008	-120.0	1.25-005	4.80-001	
195.1	3.21-008	-79.0	1.25-005	5.40-001	
235.1	3.35-008	-38.0	1.46-005	5.81-001	
296.1	3.51-008	23.0	1.81-005	5.57-001	

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

Stainless and Alloy Steels (cont.)

TYLER AND WILSON(1952), ALLOY AND STAINLESS STEELS - TW

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
60.0	5.22-008	60.0	5.22-008	60.0	5.22-008
100.0	5.24-008	100.0	5.24-008	100.0	5.24-008
140.0	4.54-008	140.0	4.54-008	140.0	4.54-008
180.0	4.12-008	180.0	4.12-008	180.0	4.12-008
220.0	3.91-008	220.0	3.91-008	220.0	3.91-008
260.0	3.84-008	260.0	3.84-008	260.0	3.84-008

TYLER, NESBITT, AND WILSON(1953), ALLOY AND STAINLESS STEELS - TNW

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
70.0	5.31-008	70.0	5.31-008	70.0	5.31-008
100.0	4.96-008	100.0	4.96-008	100.0	4.96-008
200.0	3.80-008	200.0	3.80-008	200.0	3.80-008
300.0	3.83-008	300.0	3.83-008	300.0	3.83-008

KOLHAAS AND KIERSPE(1965), ALLOY AND STAINLESS STEELS - KK(1)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
93.0	8.15-008	93.0	8.15-008	93.0	8.15-008
149.0	6.25-008	149.0	6.25-008	149.0	6.25-008
199.0	5.49-008	199.0	5.49-008	199.0	5.49-008
298.0	5.69-008	298.0	5.69-008	298.0	5.69-008

Carbon Steels

Cobalt

LEES(1908), CARBON STEEL - L

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
101.0	3.34-008	103.0	3.34-008	113.0	3.26-008	123.0	3.19-008
111.0	3.26-008	113.0	3.26-008	123.0	3.19-008	128.0	3.11-008
123.0	3.19-008	123.0	3.19-008	148.0	3.11-008	148.0	3.09-008
148.0	3.11-008	148.0	3.11-008	173.0	3.09-008	173.0	3.09-008
173.0	3.09-008	173.0	3.09-008	198.0	3.09-008	198.0	3.09-008
198.0	3.09-008	223.0	3.10-008	223.0	3.10-008	248.0	3.09-008
223.0	3.10-008	248.0	3.09-008	273.0	3.09-008	273.0	3.09-008
248.0	3.09-008	273.0	3.09-008	291.0	3.05-008	291.0	3.05-008
291.0	3.05-008						

RAOAKRISHNA AND NIELSEN(1965)*, CO - RN

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
103.0	3.34-008	113.0	3.26-008	123.0	3.19-008	148.0	3.11-008
113.0	3.26-008	123.0	3.19-008	148.0	3.11-008	173.0	3.09-008
123.0	3.19-008	148.0	3.11-008	173.0	3.09-008	198.0	3.09-008
148.0	3.11-008	173.0	3.09-008	198.0	3.09-008	223.0	3.10-008
173.0	3.09-008	198.0	3.09-008	223.0	3.10-008	248.0	3.09-008
198.0	3.09-008	223.0	3.10-008	248.0	3.09-008	273.0	3.09-008
223.0	3.10-008	248.0	3.09-008	273.0	3.09-008	291.0	3.05-008
248.0	3.09-008	273.0	3.09-008	291.0	3.05-008		

WHITE AND WOODS (1957B), CO - WW

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
101.0	2.36-008	113.0	2.44-008	123.0	2.22-008	148.0	1.88-008
113.0	2.36-008	123.0	2.44-008	148.0	2.00-008	173.0	1.64-008
123.0	2.36-008	148.0	2.00-008	173.0	1.44-008	198.0	1.20-008
148.0	2.00-008	173.0	1.64-008	198.0	1.44-008	223.0	1.16-008
173.0	1.64-008	198.0	1.44-008	223.0	1.16-008	248.0	1.12-008
198.0	1.44-008	223.0	1.16-008	248.0	1.12-008	273.0	1.08-008
223.0	1.16-008	248.0	1.12-008	273.0	1.08-008	291.0	3.05-008
248.0	1.12-008	273.0	1.08-008	291.0	3.05-008		

WILKES,POWELL, AND OWENITT (1969)*, CO - wPO

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
100.0	1.60-008	110.0	1.88-008	120.0	1.94-008	130.0	1.97-008
110.0	1.60-008	120.0	1.88-008	130.0	1.94-008	140.0	1.97-008
120.0	1.88-008	130.0	1.94-008	140.0	1.97-008	150.0	2.00-008
130.0	1.94-008	140.0	1.97-008	150.0	2.00-008	160.0	2.03-008
140.0	1.97-008	150.0	2.00-008	160.0	2.03-008	170.0	2.06-008
150.0	2.00-008	160.0	2.03-008	170.0	2.06-008	180.0	2.09-008
160.0	2.03-008	170.0	2.06-008	180.0	2.09-008	190.0	2.12-008
170.0	2.06-008	180.0	2.09-008	190.0	2.12-008	200.0	2.15-008
180.0	2.09-008	190.0	2.12-008	200.0	2.15-008	210.0	2.18-008
190.0	2.12-008	200.0	2.15-008	210.0	2.18-008	220.0	2.21-008
200.0	2.15-008	210.0	2.18-008	220.0	2.21-008	230.0	2.24-008
210.0	2.18-008	220.0	2.21-008	230.0	2.24-008	240.0	2.27-008
220.0	2.21-008	230.0	2.24-008	240.0	2.27-008	250.0	2.30-008
230.0	2.24-008	240.0	2.27-008	250.0	2.30-008	260.0	2.33-008
240.0	2.27-008	250.0	2.30-008	260.0	2.33-008	270.0	2.36-008
250.0	2.30-008	260.0	2.33-008	270.0	2.36-008	280.0	2.39-008
260.0	2.33-008	270.0	2.36-008	280.0	2.39-008	290.0	2.42-008
270.0	2.36-008	280.0	2.39-008	290.0	2.42-008		

Nickel

WHITE AND TAINSH(1967), NI - WT

	LORENZ RATIO	ELECTRICAL RESISTIVITY	Thermal Conductivity
TEMP	TEMP	1.85-008	1.59-001
20.0	1.47-008	20.0	1.60-007
50.0	1.29-008	50.0	4.05-000
80.0	1.33-008	80.0	5.59-007
90.0	1.52-008	90.0	1.99-000
			1.73+000
GRIEG AND HARRISON(1965), NI - GH			
TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
5.0	2.41-008	5.0	2.41-008
10.0	2.25-008	10.0	2.26-008
20.0	1.90-008	20.0	1.90-008
30.0	1.53-008	30.0	1.53-008
40.0	1.32-008	40.0	1.32-008
50.0	1.26-008	50.0	1.26-008
60.0	1.28-008	60.0	1.28-008
70.0	1.33-008	70.0	1.33-008
80.0	1.39-008	80.0	1.39-008
90.0	1.45-008	90.0	1.45-008

KEMP, KLEMENS, AND WHITE (1956), NI - KK_W

	LORENZ RATIO	TEMP	LORENZ RATIO
TEMP	TEMP	2.35-008	2.35-008
5.0	2.35-008	15.0	2.03-008
15.0	2.03-008	25.0	1.57-008
25.0	1.57-008	35.0	1.36-008
35.0	1.36-008	45.0	1.25-008
45.0	1.25-008	55.0	1.21-008
55.0	1.21-008	65.0	1.20-008
65.0	1.20-008	75.0	1.25-008
75.0	1.25-008	90.0	1.50-008
90.0	1.50-008	115.0	1.71-008
115.0	1.71-008	135.0	1.85-008
			1.85-008

LEES(1908), NI - L

	LORENZ RATIO	TEMP	LORENZ RATIO
TEMP	TEMP	2.92-008	2.92-008
103.0	2.92-008	103.0	2.92-008
113.0	2.82-008	113.0	2.92-008
123.0	2.73-008	123.0	2.73-008
148.0	2.63-008	148.0	2.63-008
173.0	2.59-008	173.0	2.59-008
198.0	2.56-008	198.0	2.56-008
223.0	2.55-008	223.0	2.55-008
248.0	2.57-008	248.0	2.57-008
273.0	2.59-008	273.0	2.59-008
291.0	2.59-008	291.0	2.59-008

AOYAMA(1940), NI ALLOYS - A

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
73.1	6.66-008	-200.0	1.59-008
173.1	4.60-008	-100.0	1.10-008
273.1	3.31-008	0.0	7.90-009

ESTERMANN AND ZIMMERMANN(1952), NI ALLOY - EZ(MOT)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
4.2	3.70-008	4.2	3.70-008
63.3	6.20-008	63.3	6.20-008
77.3	5.50-008	77.3	5.50-008
77.3	5.90-008	77.3	5.90-008
4.2	3.70-008	4.2	3.70-008
14.0	5.60-008	14.0	5.60-008
20.4	6.00-008	20.4	6.00-008
73.7	5.70-008	73.7	5.70-008
77.0	4.50-008	77.0	4.50-008
77.0	5.80-008	77.0	5.80-008

ESTERMANN AND ZIMMERMANN(1952), NI ALLOY - EZ(MAT)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
2.6	4.30-008	2.6	4.30-008
4.3	6.40-008	4.3	6.40-008
14.0	9.80-008	14.0	9.80-008
20.5	1.09-007	20.5	1.09-007
63.7	7.20-008	63.7	7.20-008
63.7	7.20-008	63.7	7.20-008
68.4	7.50-008	68.4	7.50-008
77.0	6.90-008	77.0	6.90-008

ESTERMANN AND ZIMMERMANN(1952), NI ALLOY - EZ(MOR)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
2.6	2.35-008	2.6	2.35-008
2.7	2.27-008	2.7	2.27-008
4.2	3.00-008	4.2	3.00-008
10.1	3.70-008	10.1	3.70-008
20.4	5.60-008	20.4	5.60-008
20.6	5.90-008	20.6	5.90-008
63.3	5.30-008	63.3	5.30-008
75.5	5.30-008	75.5	5.30-008
77.0	6.00-008	77.0	6.00-008

ESTERMANN AND ZIMMERMANN(1952), NI ALLOY - EZ(IOT)

TEMP	LORENZ RATIO	TEMP	LORENZ RATIO
63.3	1.01-007	63.3	1.01-007
73.4	1.03-007	73.4	1.03-007
77.0	1.05-007	77.0	1.05-007
2.6	3.70-008	2.6	3.70-008
4.2	6.40-008	4.2	6.40-008
9.6	8.00-008	9.6	8.00-008

Nickel Alloys (cont.)

ESTERMANN AND ZIMMERMANN (1952) • NI ALLOY - EZ110T)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
14.0	1.01-007	14.0	1.01-007	14.0	1.01-007	14.0	1.01-007	14.0
20.4	1.10-007	20.4	1.10-007	20.4	1.10-007	20.4	1.10-007	20.4
63.3	1.03-007	63.3	1.03-007	63.3	1.03-007	63.3	1.03-007	63.3
72.4	1.03-007	72.4	1.03-007	72.4	1.03-007	72.4	1.03-007	72.4
77.0	1.05-007	77.0	1.05-007	77.0	1.05-007	77.0	1.05-007	77.0
2.6	3.07-008	2.6	3.07-008	2.6	3.07-008	2.6	3.07-008	2.6
4.2	6.40-008	4.2	6.40-008	4.2	6.40-008	4.2	6.40-008	4.2
4.2	5.70-008	4.2	5.70-008	4.2	5.70-008	4.2	5.70-008	4.2
9.6	8.00-008	9.6	8.00-008	9.6	8.00-008	9.6	8.00-008	9.6

HUST AND SPARKS (1970) • NI ALLOYS - HPW(718A)

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
6.0	1.73-007	6.0	1.73-007	6.0	1.73-007	6.0	1.73-007	6.0
7.0	1.75-007	7.0	1.75-007	7.0	1.75-007	7.0	1.75-007	7.0
8.0	1.78-007	8.0	1.78-007	8.0	1.78-007	8.0	1.78-007	8.0
9.0	1.80-007	9.0	1.80-007	9.0	1.80-007	9.0	1.80-007	9.0
10.0	1.82-007	10.0	1.82-007	10.0	1.82-007	10.0	1.82-007	10.0
12.0	1.84-007	12.0	1.84-007	12.0	1.84-007	12.0	1.84-007	12.0
14.0	1.84-007	14.0	1.84-007	14.0	1.84-007	14.0	1.84-007	14.0
16.0	1.82-007	16.0	1.82-007	16.0	1.82-007	16.0	1.82-007	16.0
18.0	1.80-007	18.0	1.80-007	18.0	1.80-007	18.0	1.80-007	18.0
20.0	1.77-007	20.0	1.77-007	20.0	1.77-007	20.0	1.77-007	20.0
25.0	1.69-007	25.0	1.69-007	25.0	1.69-007	25.0	1.69-007	25.0
30.0	1.61-007	30.0	1.61-007	30.0	1.61-007	30.0	1.61-007	30.0
35.0	1.52-007	35.0	1.52-007	35.0	1.52-007	35.0	1.52-007	35.0
40.0	1.44-007	40.0	1.44-007	40.0	1.44-007	40.0	1.44-007	40.0
45.0	1.37-007	45.0	1.37-007	45.0	1.37-007	45.0	1.37-007	45.0
50.0	1.30-007	50.0	1.30-007	50.0	1.30-007	50.0	1.30-007	50.0
55.0	1.24-007	55.0	1.24-007	55.0	1.24-007	55.0	1.24-007	55.0
60.0	1.18-007	60.0	1.18-007	60.0	1.18-007	60.0	1.18-007	60.0
65.0	1.13-007	65.0	1.13-007	65.0	1.13-007	65.0	1.13-007	65.0
70.0	1.08-007	70.0	1.08-007	70.0	1.08-007	70.0	1.08-007	70.0
75.0	1.03-007	75.0	1.03-007	75.0	1.03-007	75.0	1.03-007	75.0
80.0	9.87-008	80.0	9.87-008	80.0	9.87-008	80.0	9.87-008	80.0
85.0	9.47-008	85.0	9.47-008	85.0	9.47-008	85.0	9.47-008	85.0
90.0	9.11-008	90.0	9.11-008	90.0	9.11-008	90.0	9.11-008	90.0
95.0	8.77-008	95.0	8.77-008	95.0	8.77-008	95.0	8.77-008	95.0
100.0	8.46-008	100.0	8.46-008	100.0	8.46-008	100.0	8.46-008	100.0
110.0	7.90-008	110.0	7.90-008	110.0	7.90-008	110.0	7.90-008	110.0
120.0	7.43-008	120.0	7.43-008	120.0	7.43-008	120.0	7.43-008	120.0
130.0	7.01-008	130.0	7.01-008	130.0	7.01-008	130.0	7.01-008	130.0
140.0	6.66-008	140.0	6.66-008	140.0	6.66-008	140.0	6.66-008	140.0
150.0	6.34-008	150.0	6.34-008	150.0	6.34-008	150.0	6.34-008	150.0
160.0	6.07-008	160.0	6.07-008	160.0	6.07-008	160.0	6.07-008	160.0
170.0	5.83-008	170.0	5.83-008	170.0	5.83-008	170.0	5.83-008	170.0
180.0	5.62-008	180.0	5.62-008	180.0	5.62-008	180.0	5.62-008	180.0
190.0	5.44-008	190.0	5.44-008	190.0	5.44-008	190.0	5.44-008	190.0
200.0	5.27-008	200.0	5.27-008	200.0	5.27-008	200.0	5.27-008	200.0
220.0	4.99-008	220.0	4.99-008	220.0	4.99-008	220.0	4.99-008	220.0
240.0	4.75-008	240.0	4.75-008	240.0	4.75-008	240.0	4.75-008	240.0
260.0	4.55-008	260.0	4.55-008	260.0	4.55-008	260.0	4.55-008	260.0
280.0	4.37-008	280.0	4.37-008	280.0	4.37-008	280.0	4.37-008	280.0

HUST, POWELL, AND WEITZEL(1971), NI ALLOYS - HPW(X)

HUST, POWELL, AND WEITZEL(1971), NI ALLOYS - HPW(718)

	LORENZ		LORENZ		LORENZ	
	RATIO	TEMP	RATIO	TEMP	RATIO	TEMP
1.47-007	7.0	1.47-007	7.0	1.47-007	7.0	1.03-007
1.51-007	8.0	1.51-007	8.0	1.51-007	8.0	1.08-007
1.53-007	9.0	1.53-007	9.0	1.53-007	9.0	1.12-007
1.53-007	10.0	1.55-007	10.0	1.55-007	10.0	1.16-007
1.57-007	12.0	1.57-007	12.0	1.57-007	12.0	1.22-007
1.56-007	14.0	1.56-007	14.0	1.56-007	14.0	1.26-007
1.54-007	16.0	1.54-007	16.0	1.54-007	16.0	1.28-007
1.51-007	18.0	1.51-007	18.0	1.51-007	18.0	1.29-007
1.49-007	20.0	1.49-007	20.0	1.49-007	20.0	1.29-007
1.41-007	25.0	1.41-007	25.0	1.41-007	25.0	1.28-007
1.33-007	30.0	1.33-007	30.0	1.33-007	30.0	1.25-007
1.26-007	35.0	1.26-007	35.0	1.26-007	35.0	1.21-007
1.19-007	40.0	1.19-007	40.0	1.19-007	40.0	1.17-007
1.13-007	45.0	1.13-007	45.0	1.13-007	45.0	1.13-007
1.08-007	50.0	1.08-007	50.0	1.08-007	50.0	1.09-007
1.03-007	55.0	1.03-007	55.0	1.03-007	55.0	1.05-007
9.83-008	60.0	9.83-008	60.0	9.83-008	60.0	1.01-007
9.41-008	65.0	9.41-008	65.0	9.41-008	65.0	9.75-008
9.02-008	70.0	9.02-008	70.0	9.02-008	70.0	9.41-008
8.66-008	75.0	8.66-008	75.0	8.66-008	75.0	9.08-008
8.32-008	80.0	8.32-008	80.0	8.32-008	80.0	8.77-008
8.02-008	85.0	8.02-008	85.0	8.02-008	85.0	8.49-008
7.73-008	90.0	7.73-008	90.0	7.73-008	90.0	8.22-008
7.44-008	95.0	7.44-008	95.0	7.44-008	95.0	7.96-008
7.23-008	100.0	7.23-008	100.0	7.23-008	100.0	7.73-008
6.79-008	110.0	6.79-008	110.0	6.79-008	110.0	7.30-008
6.42-008	120.0	6.42-008	120.0	6.42-008	120.0	6.93-008
6.10-008	130.0	6.10-008	130.0	6.10-008	130.0	6.60-008
5.83-008	140.0	5.83-008	140.0	5.83-008	140.0	6.32-008
5.59-008	150.0	5.59-008	150.0	5.59-008	150.0	6.07-008
5.39-008	160.0	5.39-008	160.0	5.39-008	160.0	5.86-008
5.21-008	170.0	5.21-008	170.0	5.21-008	170.0	5.67-008
5.05-008	180.0	5.05-008	180.0	5.05-008	180.0	5.50-008
4.92-008	190.0	4.92-008	190.0	4.92-008	190.0	5.35-008
4.80-008	200.0	4.80-008	200.0	4.80-008	200.0	5.22-008
4.59-008	220.0	4.59-008	220.0	4.59-008	220.0	4.98-008
4.42-008	240.0	4.42-008	240.0	4.42-008	240.0	4.79-008
4.22-008	260.0	4.22-008	260.0	4.22-008	260.0	4.61-008
4.14-008	280.0	4.14-008	280.0	4.14-008	280.0	4.44-008
4.01-008	300.0	4.01-008	300.0	4.01-008	300.0	4.44-008

Platinum

POWELL, TYE, AND WOODMAN(1967), PT ~ PTW

	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP	LORENZ RATIO	TEMP
100.0	2.18-008	100.0	2.18-008	100.0	2.18-008	100.0	2.18-008	100.0
200.0	2.42-008	200.0	2.42-008	200.0	2.42-008	200.0	2.42-008	200.0
300.0	2.69-008	300.0	2.69-008	300.0	2.69-008	300.0	2.69-008	300.0
WHITE AND WOODS (1957), PT ~ WW								
	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity				
6.0	2.62-008	6.0	1.31-008	1.20+001				
10.0	2.00-008	10.0	1.55-008	1.28+001				
15.0	1.57-008	15.0	2.80-008	8.40+000				
20.0	1.40-008	20.0	5.65-008	4.60+000				
30.0	1.28-008	30.0	1.93-007	2.00+000				
40.0	1.13-008	40.0	4.63-007	1.50+000				
50.0	1.54-008	50.0	7.72-007	1.00+000				
75.0	2.31-008	75.0	1.75-006	1.00+000				
100.0	2.81-008	100.0	2.81-006	1.00+000				

MOORE, MCELROY, AND BARRONI(1966), PT ~ MMB (1)

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity				
90.0	2.24-008	90.0	2.74-006	7.36-001				
100.0	2.25-008	100.0	3.08-006	7.30-001				
120.0	2.59-008	120.0	3.80-006	7.22-001				
140.0	2.33-008	140.0	4.55-006	7.16-001				
160.0	2.37-008	160.0	5.34-006	7.11-001				
180.0	2.42-008	180.0	6.18-006	7.06-001				
200.0	2.46-008	200.0	6.99-006	7.04-001				
220.0	2.49-008	220.0	7.79-006	7.03-001				
240.0	2.52-008	240.0	8.59-006	7.03-001				
260.0	2.54-008	260.0	9.39-006	7.04-001				
280.0	2.56-008	280.0	1.02-005	7.04-001				
300.0	2.59-008	300.0	1.10-005	7.05-001				

MOORE, MCELROY, AND BARRONI(1966), PT ~ MMB (2)

	LORENZ RATIO	TEMP	ELECTRICAL RESISTIVITY	Thermal Conductivity				
90.0	1.87-008	90.0	2.39-006	7.35-001				
100.0	2.06-008	100.0	2.76-006	7.30-001				
120.0	2.21-008	120.0	3.62-006	7.33-001				
140.0	2.30-008	140.0	4.44-006	7.25-001				
160.0	2.38-008	160.0	5.20-006	7.19-001				
180.0	2.43-008	180.0	6.12-006	7.16-001				
200.0	2.47-008	200.0	6.93-006	7.14-001				
220.0	2.51-008	220.0	7.73-006	7.14-001				
240.0	2.54-008	240.0	8.53-006	7.15-001				
260.0	2.56-008	260.0	9.30-006	7.17-001				
280.0	2.58-008	280.0	1.01-005	7.19-001				
300.0	2.61-008	300.0	1.09-005	7.21-001				

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A comprehensive review and compilation of the world literature on Lorenz ratio of technically important metals and alloys is presented. Lorenz ratio, electrical resistivity, thermal conductivity and characterization data are compiled in tabular form and the Lorenz ratio data are presented in graphical form as well. Data are included here only if the research reported both thermal conductivity and electrical resistivity of the specimens. No attempt has been made to smooth data or present recommended values.

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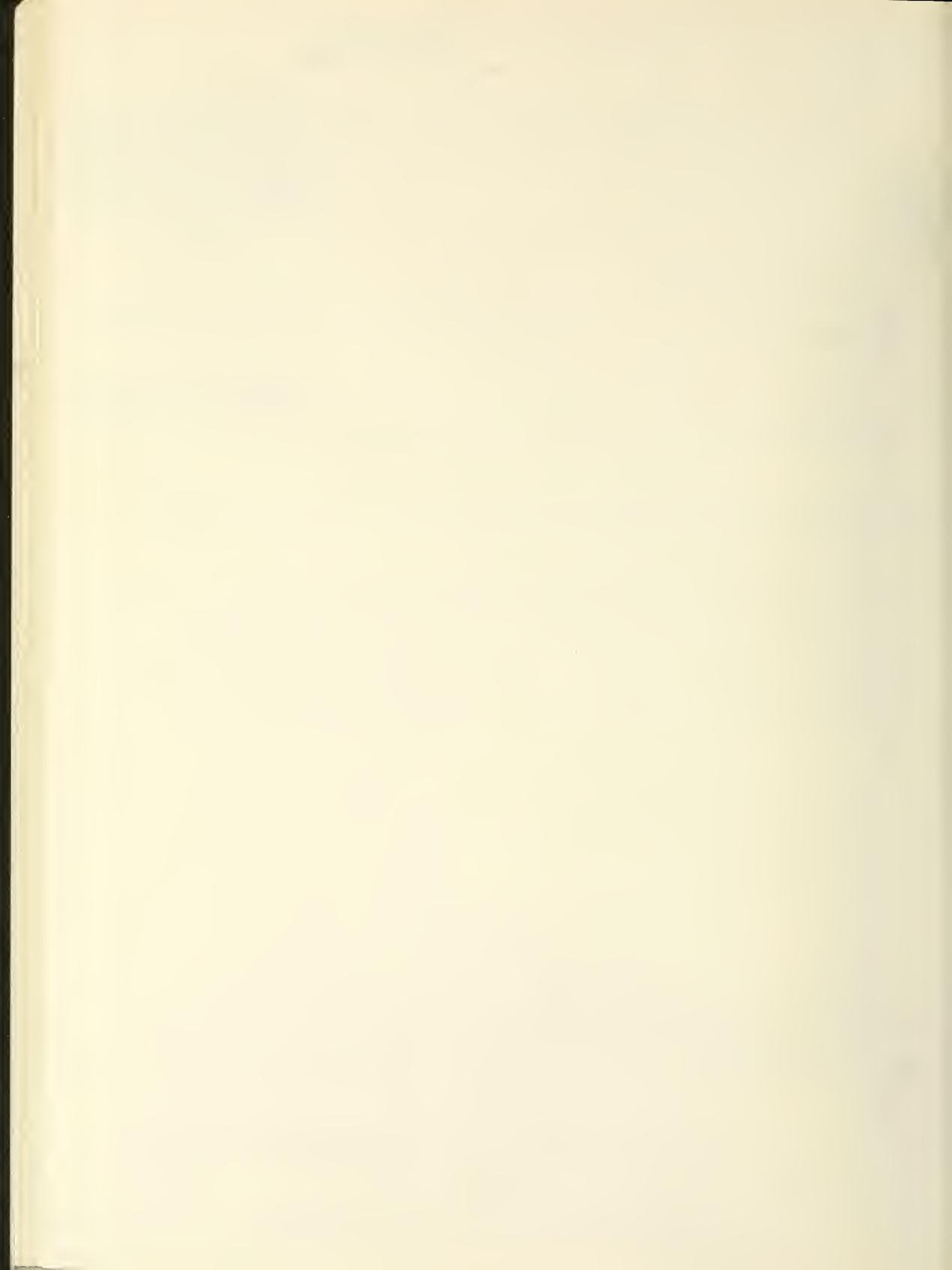
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